



Isotopic anomalies in meteorites: an introduction

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Geologists use fossils to reconstruct paleo-environments

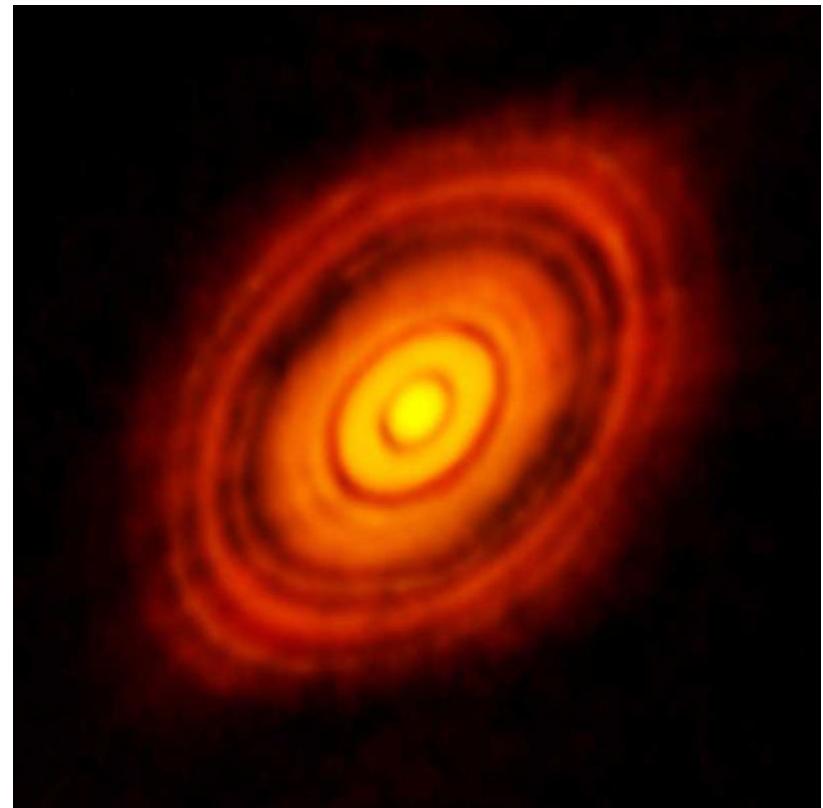
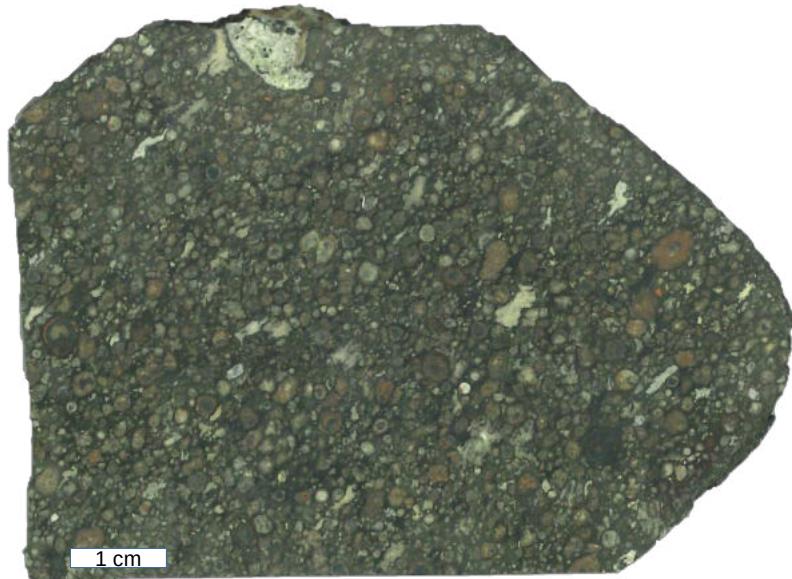


Rußbach fossil wall

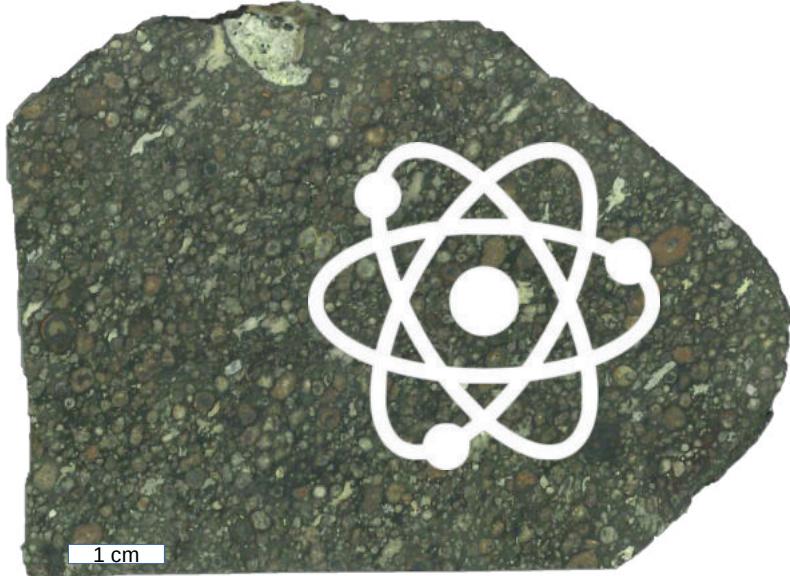


Cretaceous reef fauna

Meteorites as fossils of solar accretion disk

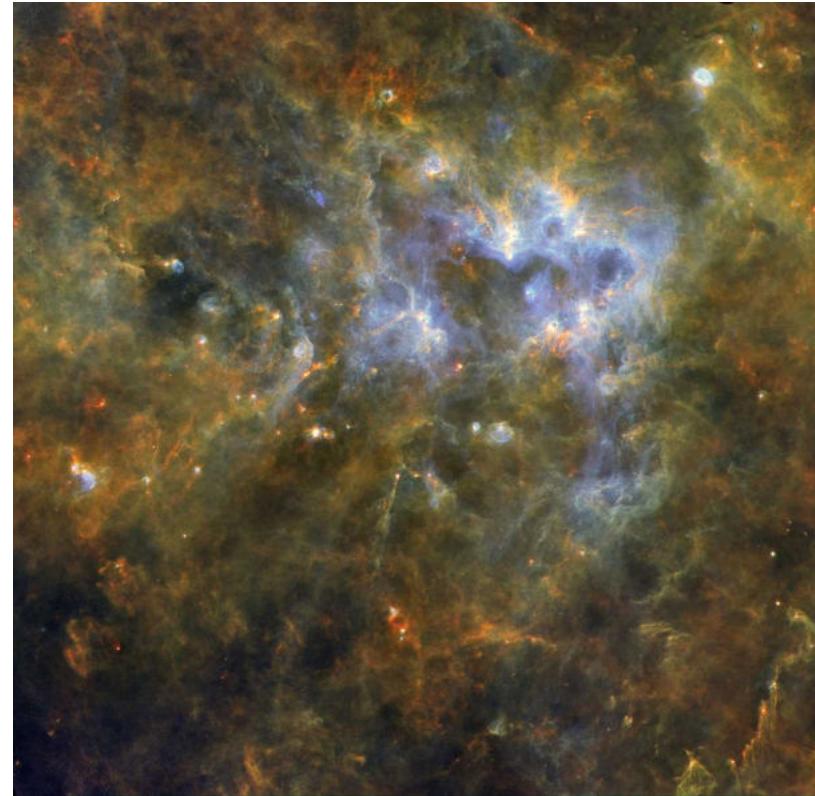


Isotopic anomalies in meteorites



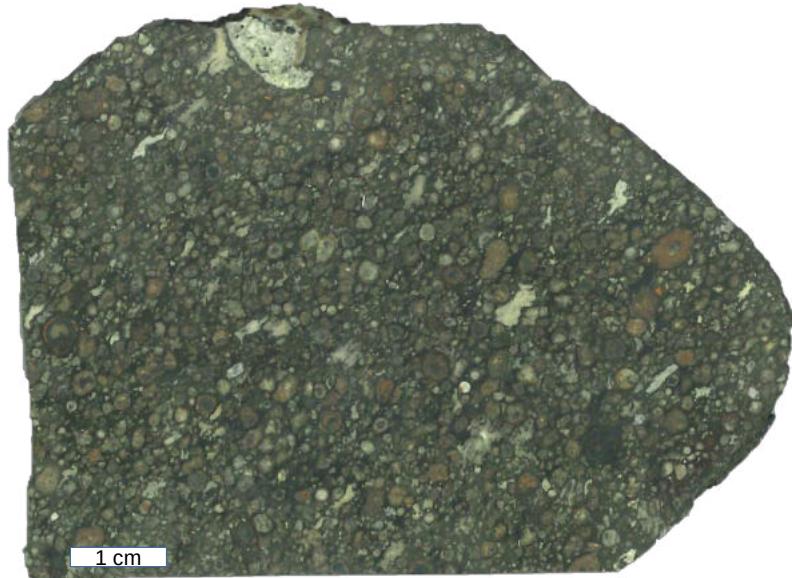
Tracer of solar system evolution

Isotopic anomalies in meteorites



What is the solar system made of?

Isotopic anomalies in meteorites



How did the disk form/disk dynamics?

Isotopic anomalies in meteorites



What is the Earth made of/How did it form?

Isotopic anomalies in meteorites



Heritage of Earth's volatiles?

Meteorites 101



Meteor, meteoroid



Meteorite

Meteorites 101

73665 official meteorites

473 falls

379 Mars

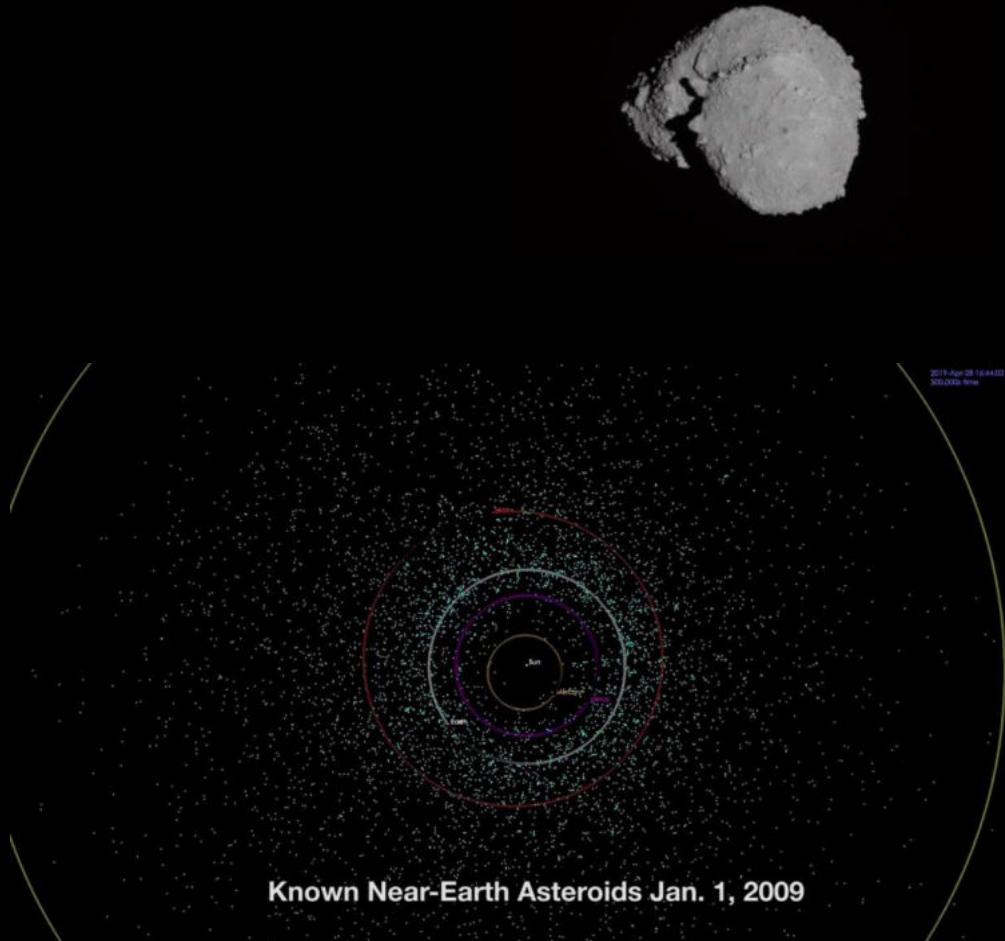
666 Moon

Search the [Meteoritical Bulletin Database](#)

Last update: 26 Feb 2024

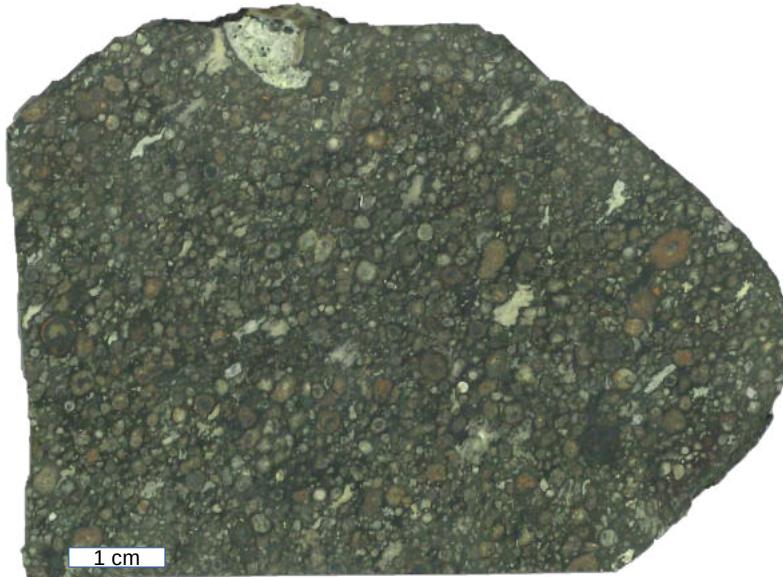
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<input type="radio"/> Text	<input type="radio"/> Starts with	All classifications	Sort by name	
<input type="radio"/> Places	<input type="radio"/> Exact	NonAntarctic	50 lines/page	
<input type="radio"/> Classes	<input type="radio"/> Sounds like	Falls	Normal table	(no time limit)
<input type="radio"/> Years		Non-NWAS		
		Has strewnfield		
		Has photo		
			<input type="checkbox"/> Limit to approved meteorite names	

Search text:



Meteorite 1o1: phenomenological classification

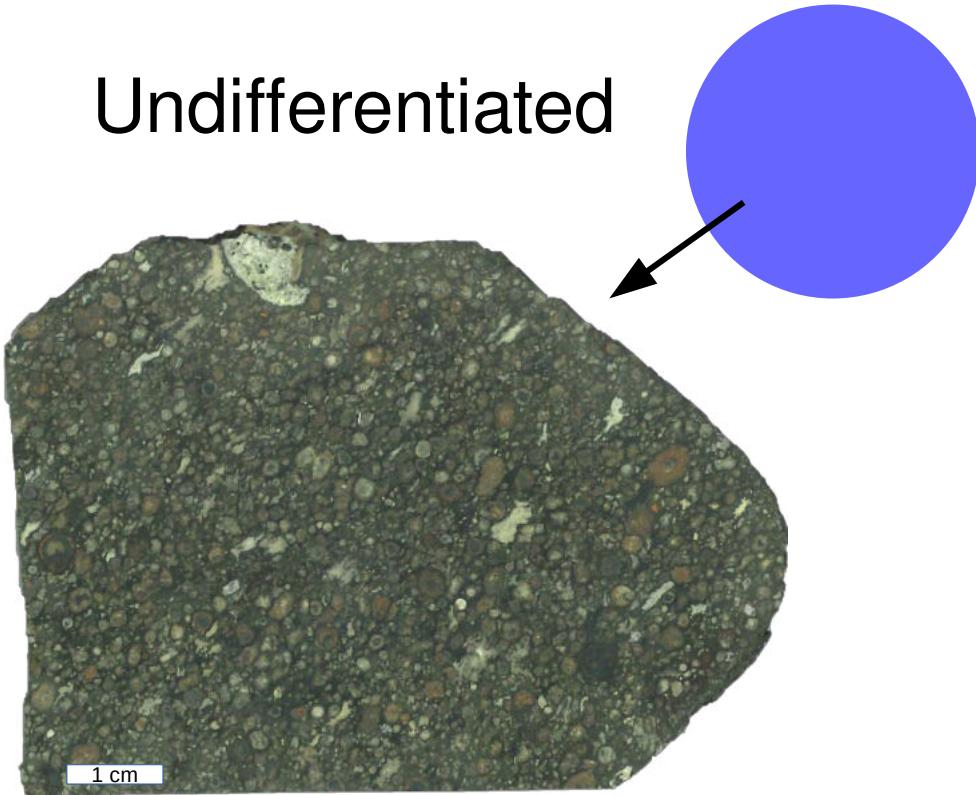
Undifferentiated



Chondrites: “nebular sediment”

Meteorite 1o1: phenomenological classification

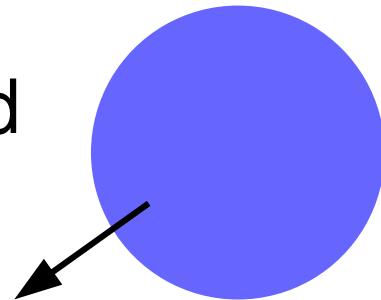
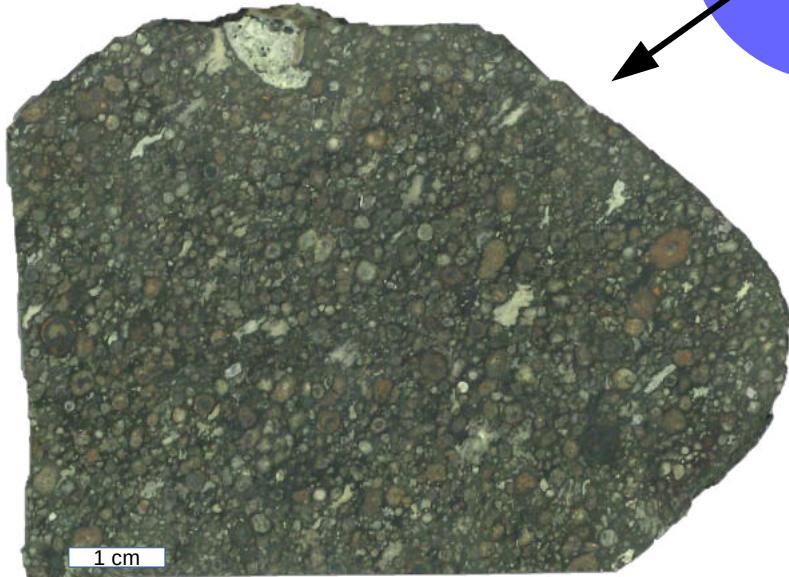
Undifferentiated



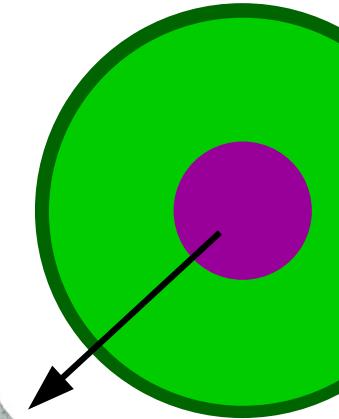
Chondrites: “nebular sediment”

Meteorite 1o1: phenomenological classification

Undifferentiated



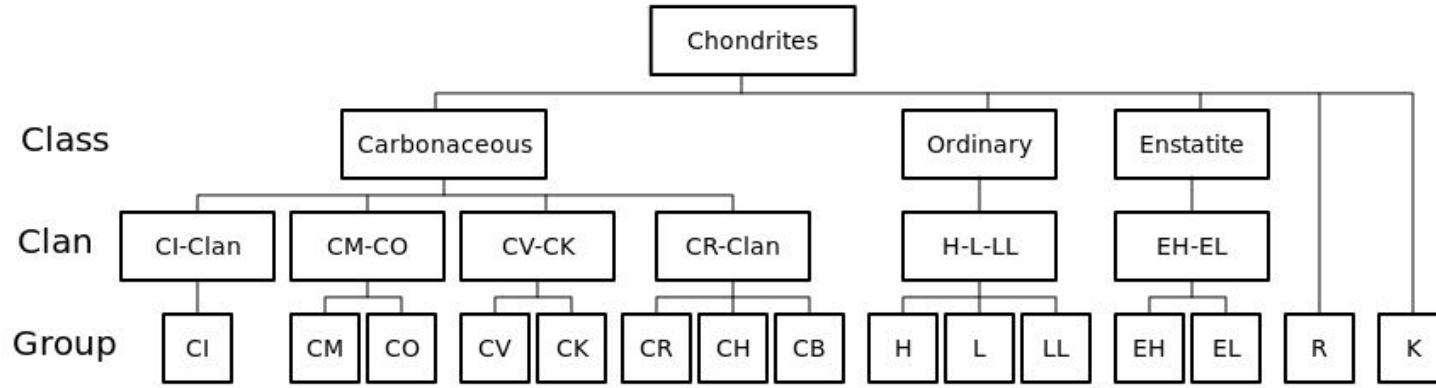
Differentiated



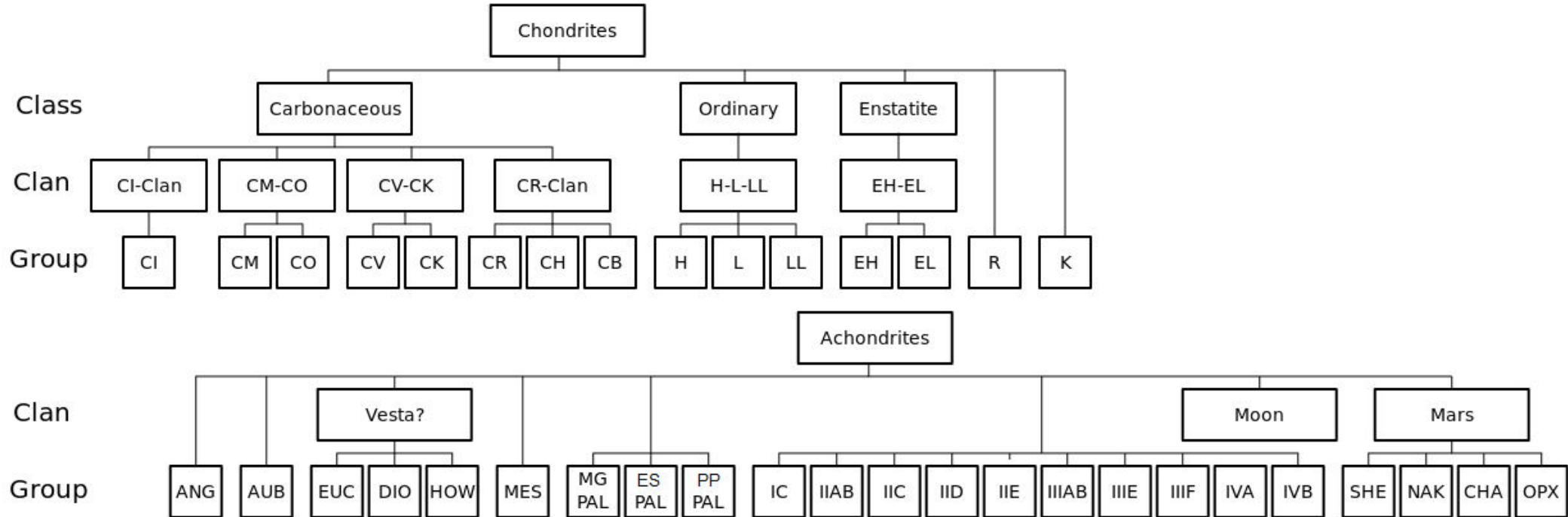
Chondrites: “nebular sediment”

Achondrites: crystallized from a melt

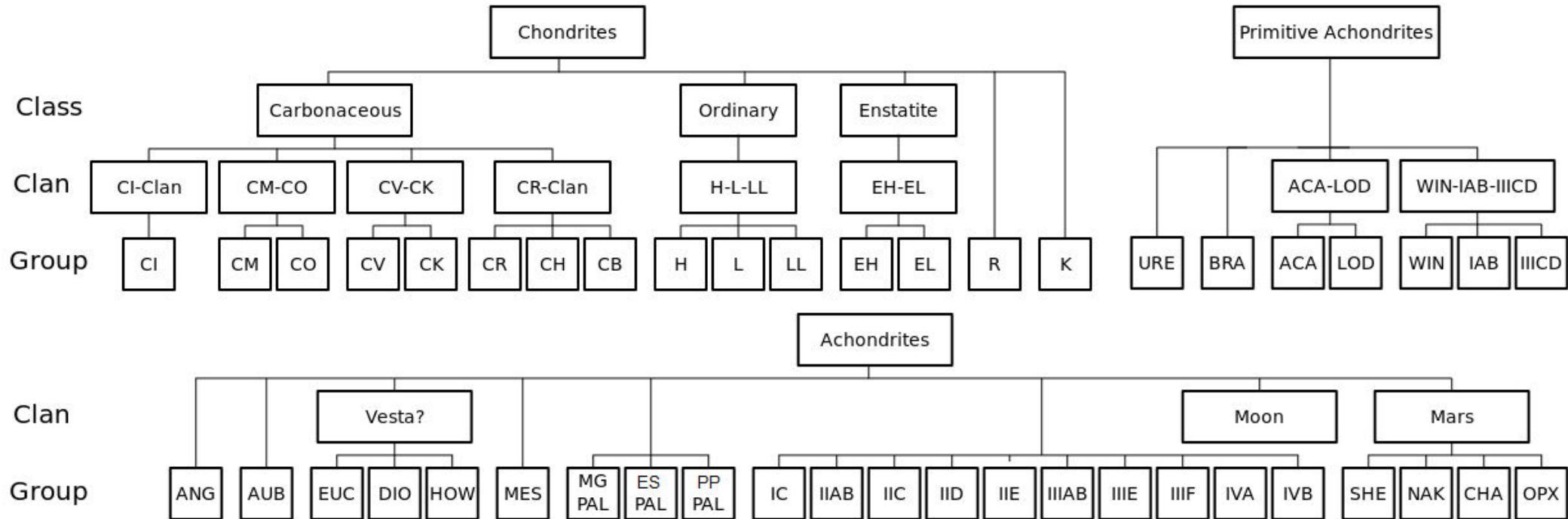
Meteorite 1o1: phenomenological classification



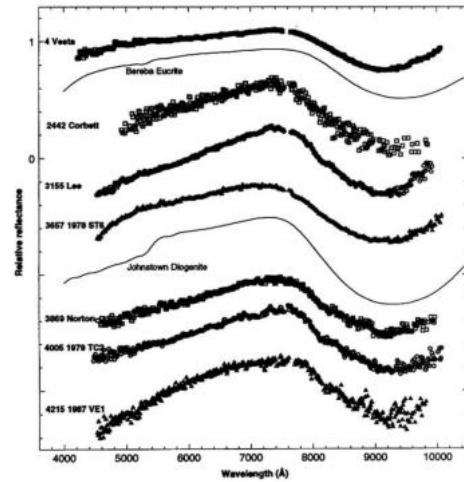
Meteorite 101: phenomenological classification



Meteorite 101: phenomenological classification

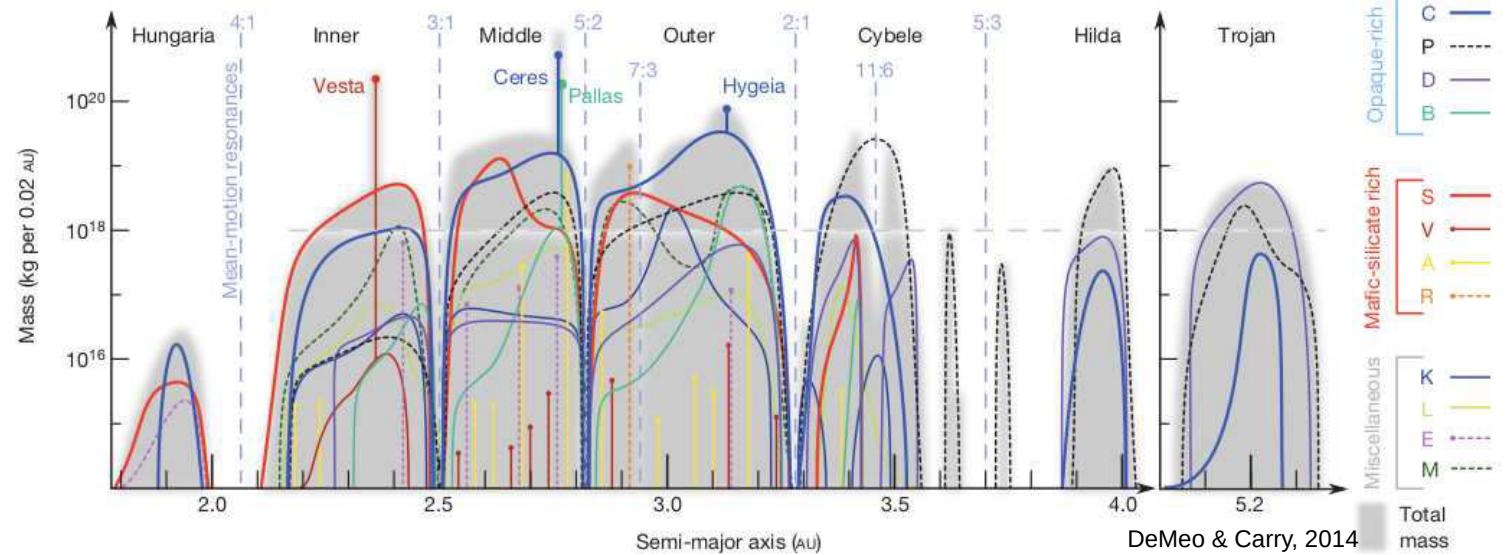


Meteorite - asteroid links

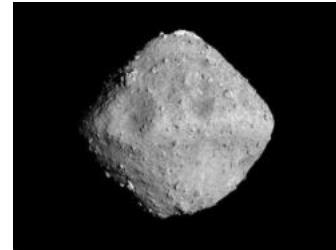


HED meteorites – 4Vesta

Binzel & Xu, 1993



Ordinary chondrites –
S-type Itokawa



CI chondrites –
Cb-type Ryugu / (B-type Bennu)

Isotope variations

mass-dependent

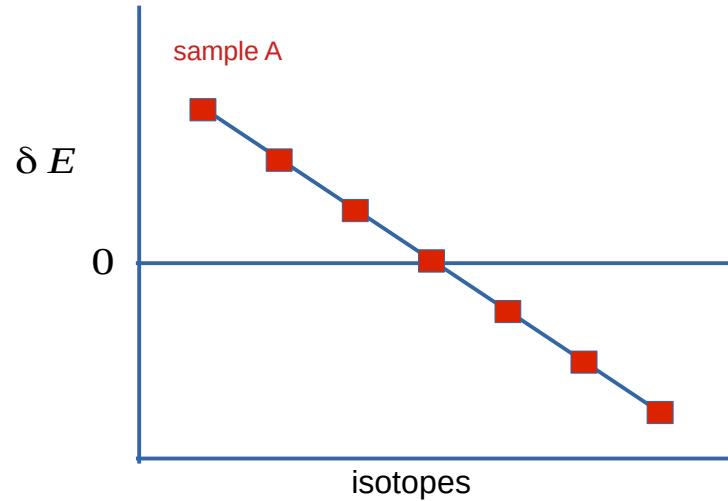
mass-independent

PROCESSES

AGES, TRACER

Isotope variations

mass-dependent

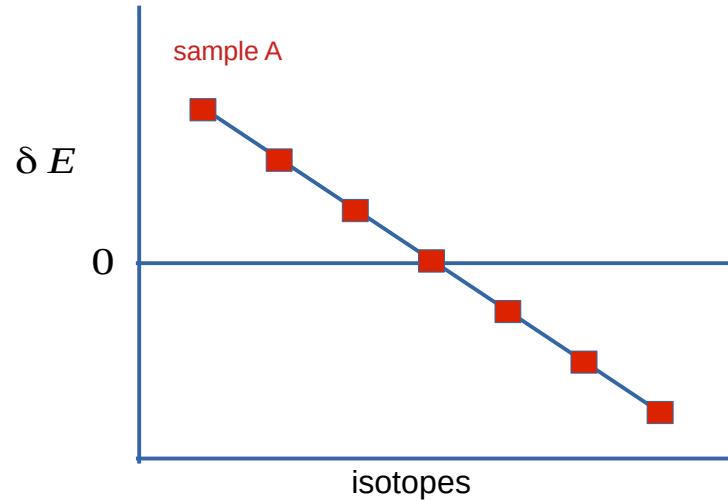


mass-independent

$$\delta^i E = \left[\frac{R_{sample}^{i,j}}{R_{standard}^{i,j}} - 1 \right] \times 10^3$$

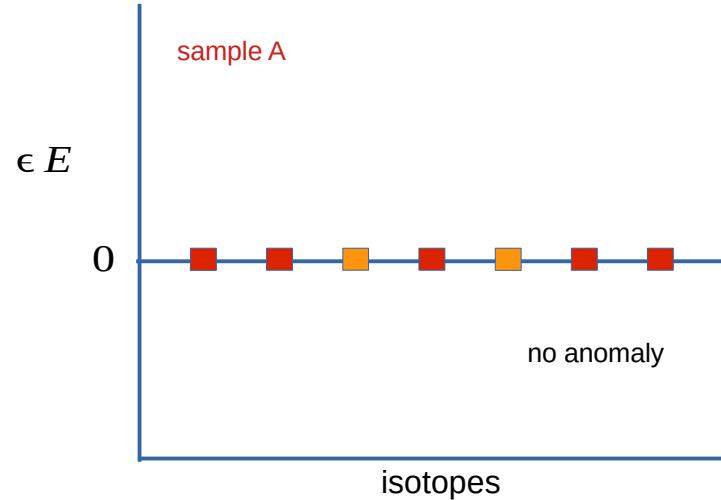
Isotope variations

mass-dependent



$$\delta^i E = \left[\frac{R_{sample}^{i,j}}{R_{standard}^{i,j}} - 1 \right] \times 10^3$$

mass-independent

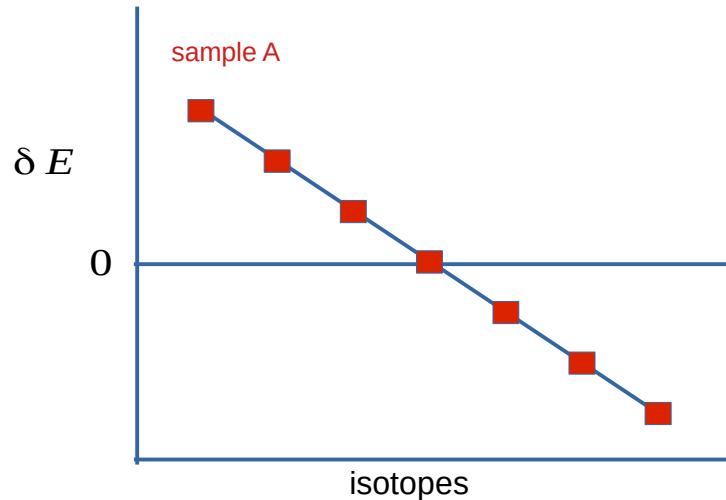


no anomaly

$$\epsilon^i E = \left[\frac{R_{sample}^{i,j}}{R_{standard}^{i,j}} - 1 \right] \times 10^4$$

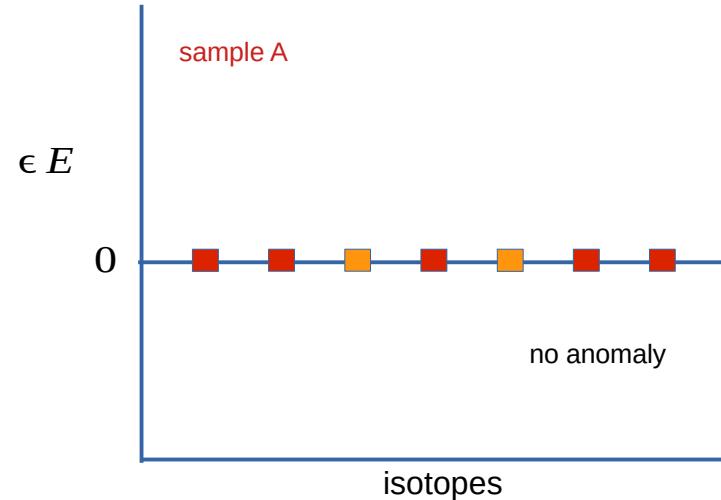
Isotope variations

mass-dependent



$$\delta^i E = \left[\frac{R_{sample}^{i,j}}{R_{standard}^{i,j}} - 1 \right] \times 10^3$$

mass-independent

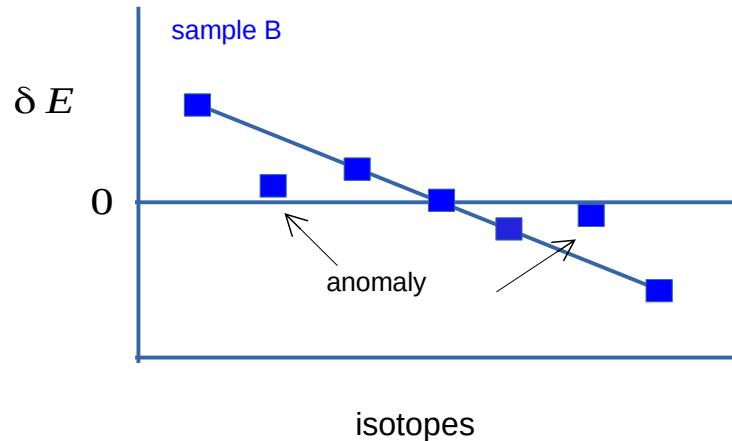


no anomaly

$$\epsilon^i E = \left[\frac{R_{sample}^{i,j} \times \left(\frac{m_i}{m_j} \right)^{\beta_{smp}}}{R_{standard}^{i,j} \times \left(\frac{m_i}{m_j} \right)^{\beta_{std}}} - 1 \right] \times 10^4$$

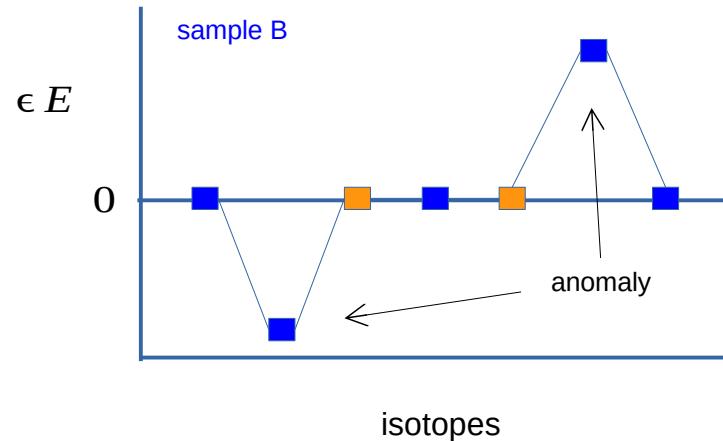
Isotope variations

mass-dependent



$$\delta^i E = \left[\frac{R_{sample}^{i,j}}{R_{standard}^{i,j}} - 1 \right] \times 10^3$$

mass-independent



$$\epsilon^i E = \left[\frac{R_{sample}^{i,j} \times \left(\frac{m_i}{m_j} \right)^{\beta_{smp}}}{R_{standard}^{i,j} \times \left(\frac{m_i}{m_j} \right)^{\beta_{std}}} - 1 \right] \times 10^4$$

Isotope variations

mass-dependent

sample B

mass-independent

sample B

Except presolar grain data all meteorite isotope anomaly data
is internally normalized
and hence may not show you the true absolute anomaly!

$$\delta^i E = \left[\frac{R_{\text{sample}}^{i,j}}{R_{\text{standard}}^{i,j}} - 1 \right] \times 10^3$$

$$\epsilon^i E = \left[\frac{R_{\text{sample}}^{i,j} \times \left(\frac{m_i}{m_j} \right)^{\beta_{\text{sample}}} - 1}{R_{\text{standard}}^{i,j} \times \left(\frac{m_i}{m_j} \right)^{\beta_{\text{std}}} - 1} \right] \times 10^4$$

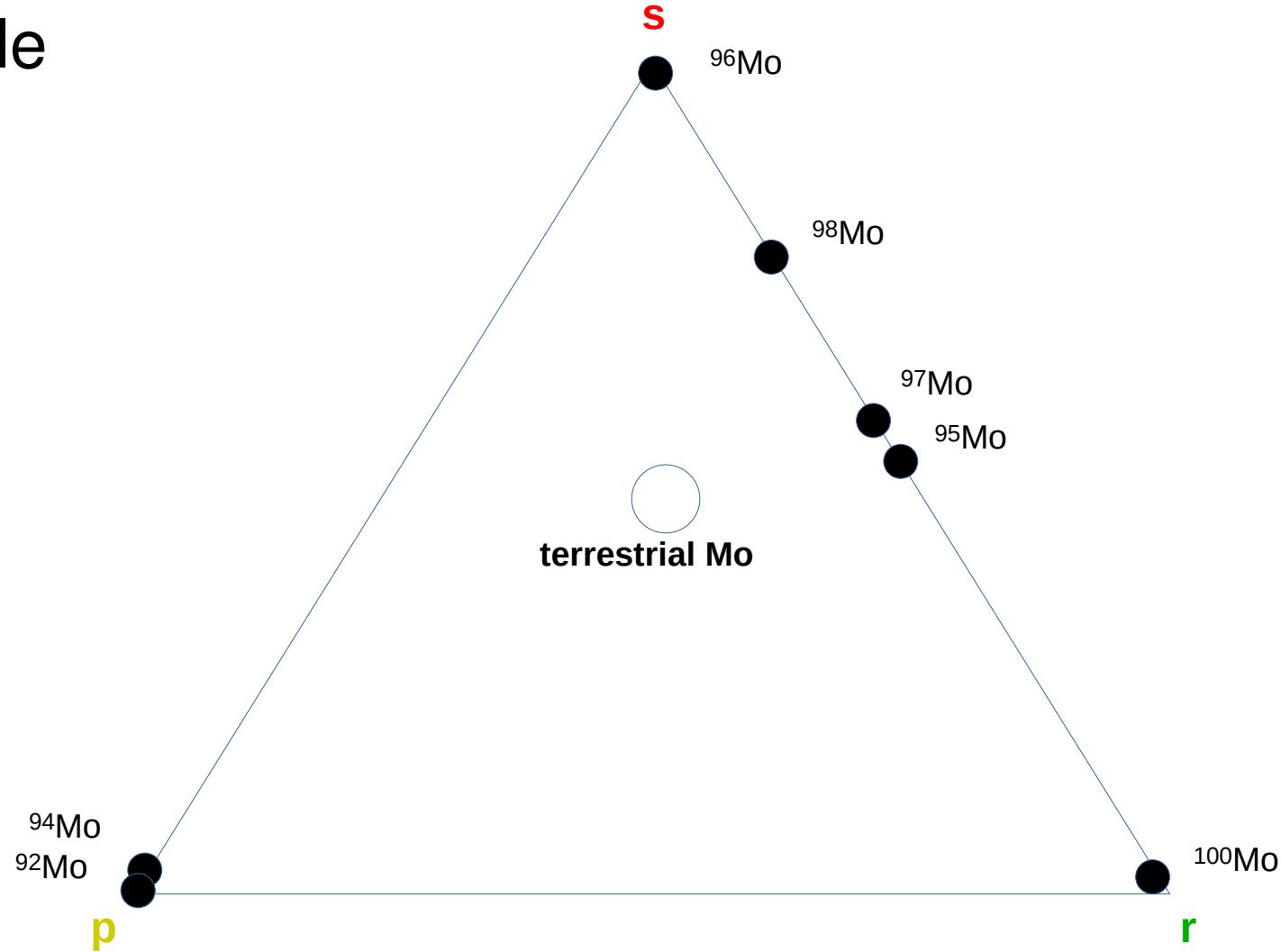
Example

Ru93	Ru94	Ru95	Ru96 5.52	Ru97 2.9d	Ru98 1.88	Ru99 12.7	Ru100 12.6	Ru101 17.0	Ru102 31.6	Ru103 40d	Ru104 18.7
Tc92	Tc93	Tc94	Tc95	Tc96	Tc97	Tc98	Tc99	Tc100	Tc101	Tc102	Tc103
Mo91	Mo92 14.84	Mo93	Mo94 9.25	Mo95 15.92	Mo96 16.68	Mo97 9.55	Mo98 24.13	Mo99 2.8d	Mo100 9.63 1.1E19a	Mo101	Mo102
Nb90	Nb91 680a	Nb92 3.6E7a	Nb93 100	Nb94 2E4a	Nb95 35d	Nb96 1d	Nb97	Nb98	Nb99	Nb100	Nb101
Zr89 3d	Zr90 51.45	Zr91 11.22	Zr92 17.15	Zr93 1.5E6a	Zr94 17.38	Zr95 64d	Zr96 2.80 3.9E19a	Zr97	Zr98	Zr99	Zr100

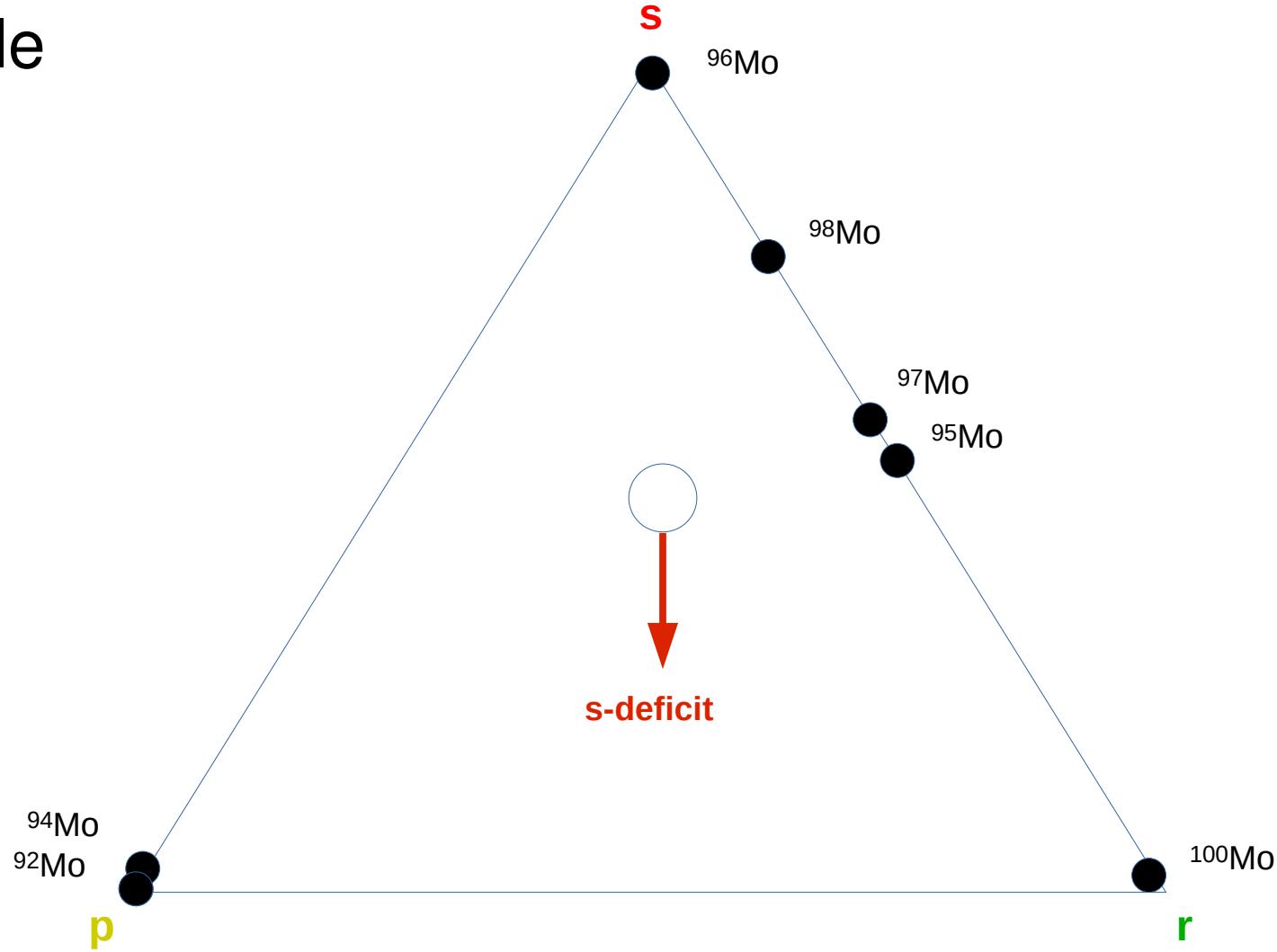
The diagram illustrates beta decay chains between various isotopes of Ruthenium (Ru), Technetium (Tc), Molybdenum (Mo), Niobium (Nb), and Zirconium (Zr). Red arrows indicate the direction of beta decay from one isotope to another. Isotopes are color-coded by element: Ru (orange), Tc (light blue), Mo (yellow), Nb (light orange), and Zr (cyan).

- Ru Chain:** Ru93 → Ru94 → Ru95 → Ru96 → Ru97 → Ru98 → Ru99 → Ru100 → Ru101 → Ru102 → Ru103 → Ru104.
- Tc Chain:** Tc92 → Tc93 → Tc94 → Tc95 → Tc96 → Tc97 → Tc98 → Tc99 → Tc100 → Tc101 → Tc102 → Tc103.
- Mo Chain:** Mo91 → Mo92 → Mo93 → Mo94 → Mo95 → Mo96 → Mo97 → Mo98 → Mo99 → Mo100 → Mo101 → Mo102.
- Nb Chain:** Nb90 → Nb91 → Nb92 → Nb93 → Nb94 → Nb95 → Nb96 → Nb97 → Nb98 → Nb99 → Nb100 → Nb101.
- Zr Chain:** Zr89 → Zr90 → Zr91 → Zr92 → Zr93 → Zr94 → Zr95 → Zr96 → Zr97 → Zr98 → Zr99 → Zr100.

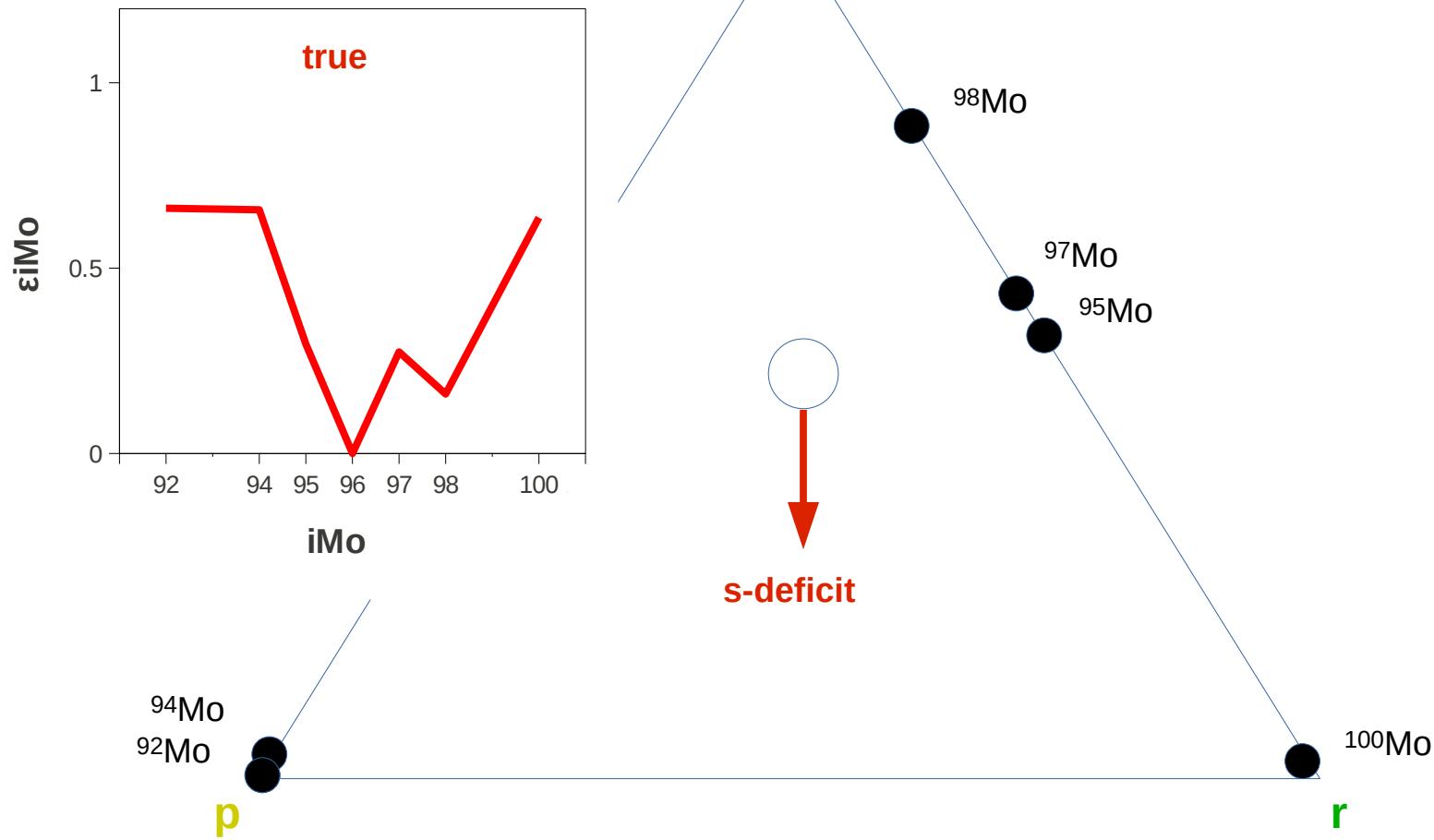
Example



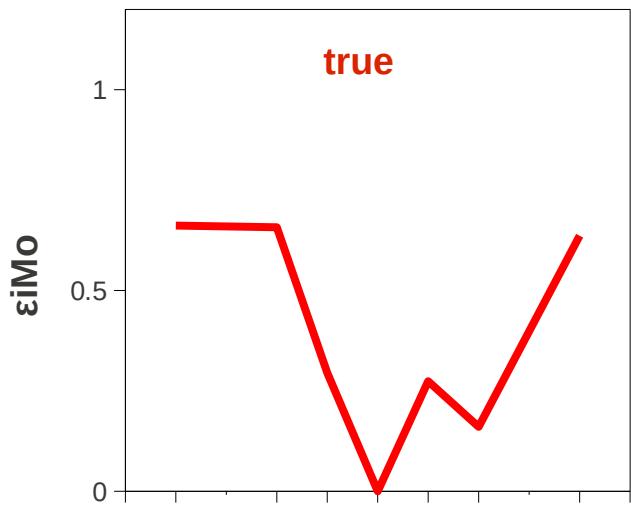
Example



Example



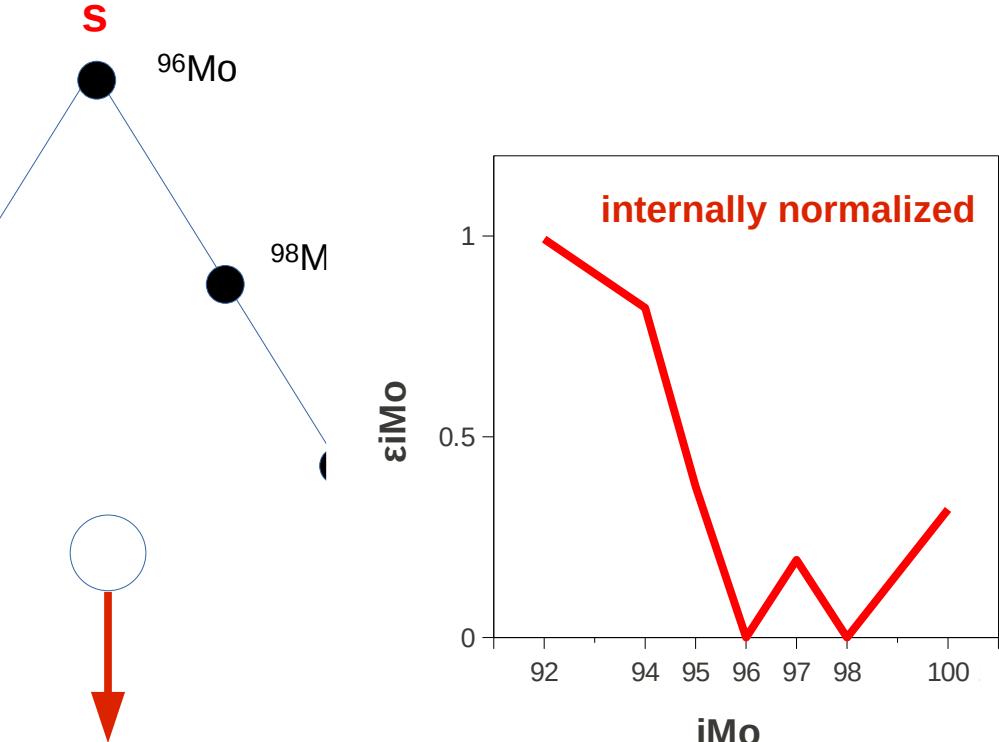
Example



$i\text{Mo}$

^{94}Mo
 ^{92}Mo

p

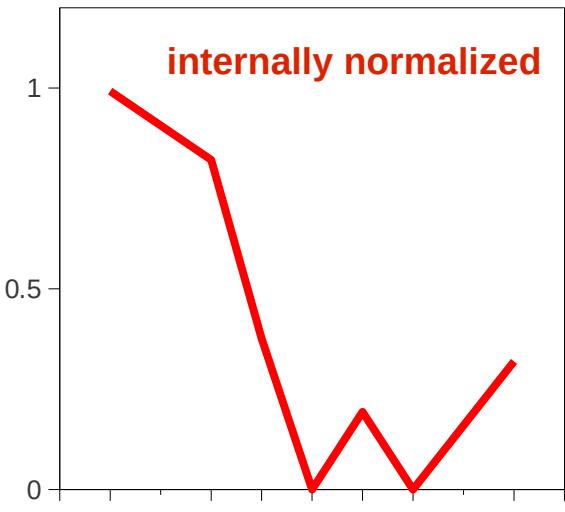


s-deficit

^{96}Mo

^{98}M

$\epsilon_{i\text{Mo}}$

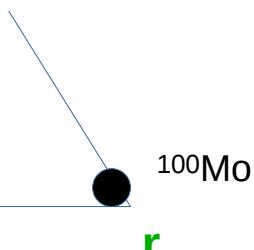
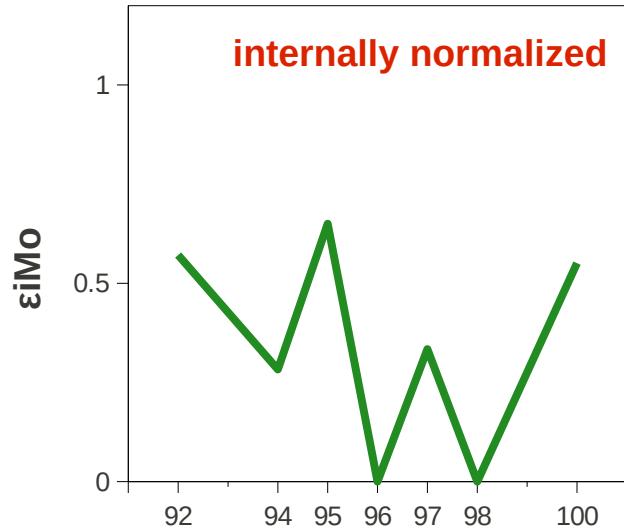
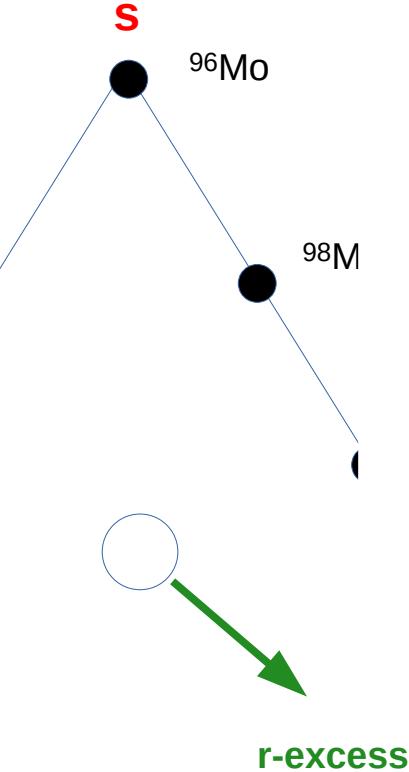
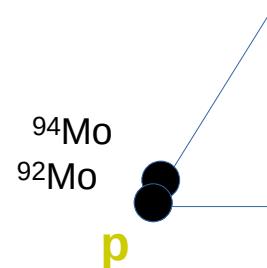
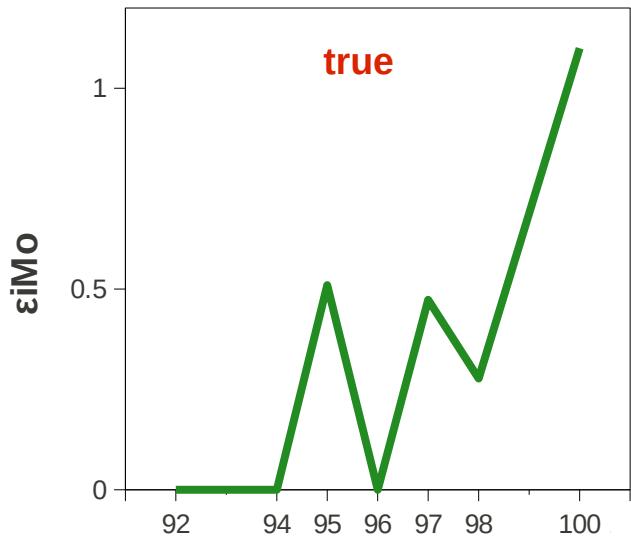


$i\text{Mo}$

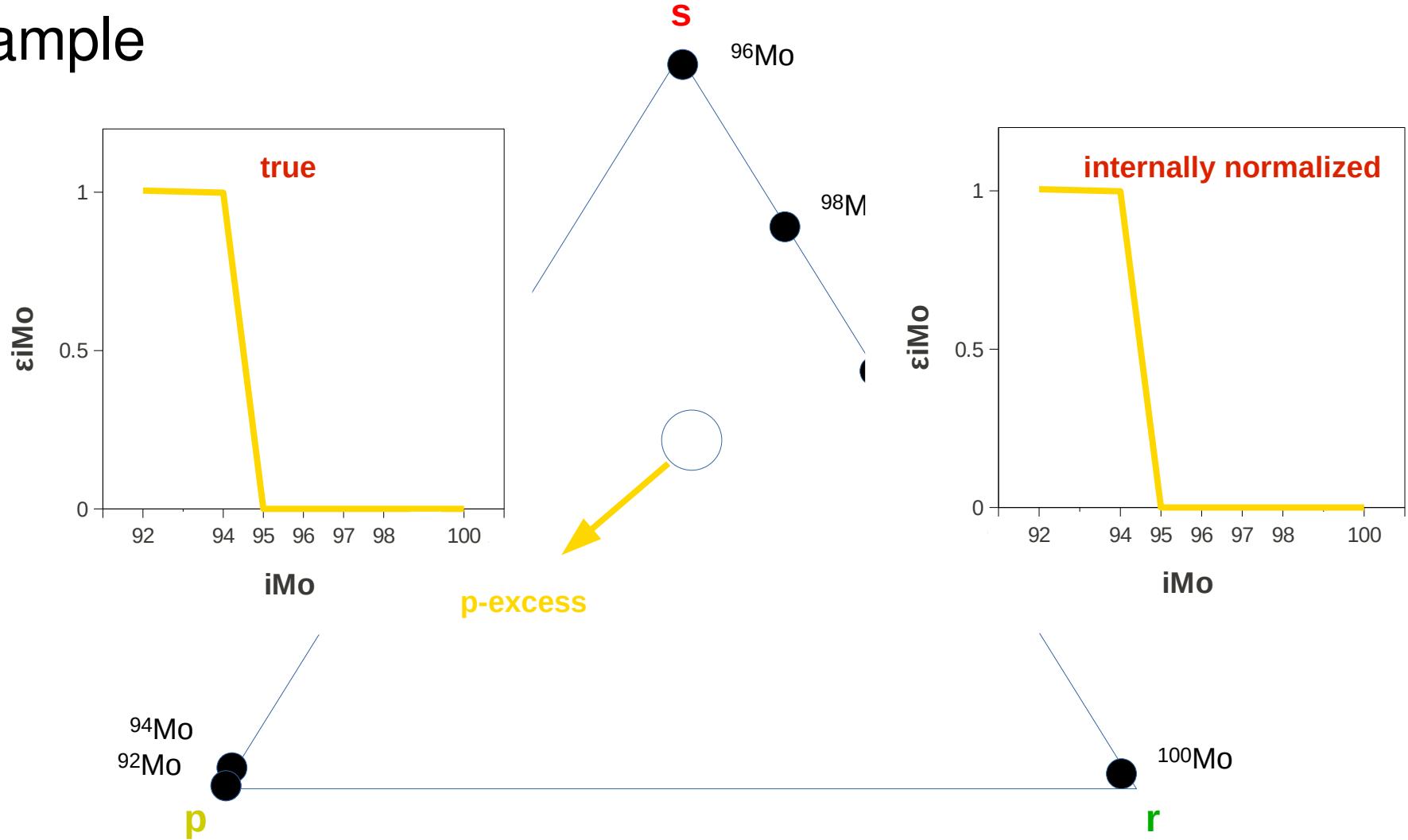
^{100}Mo

r

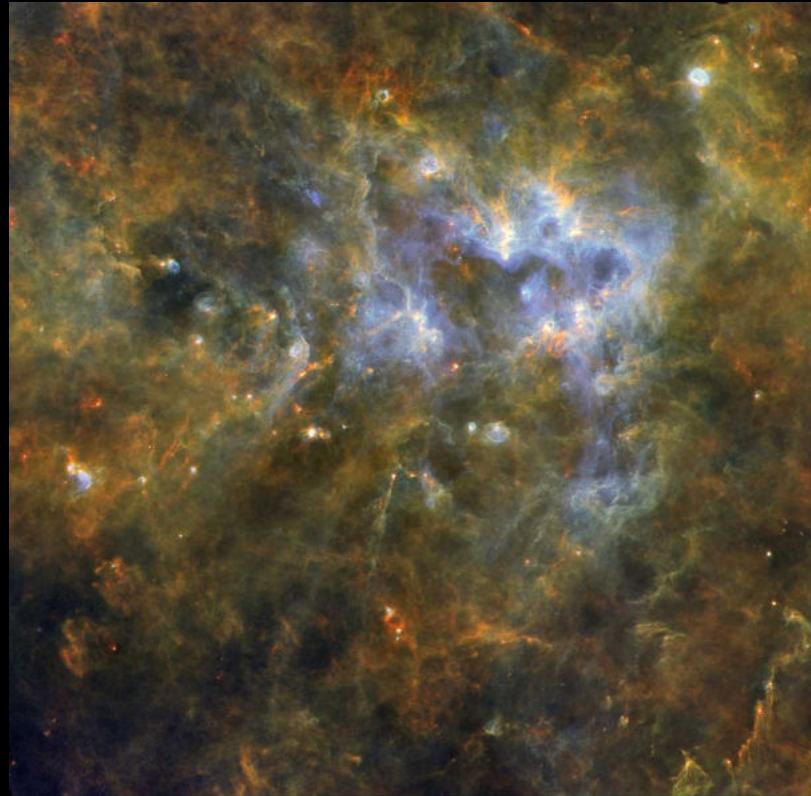
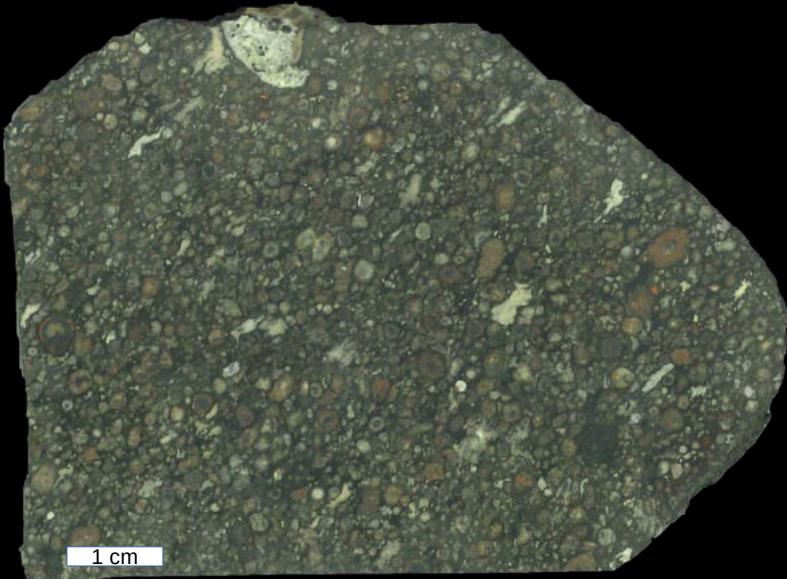
Example



Example

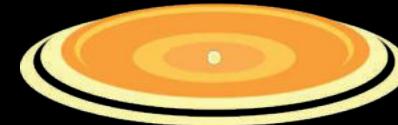
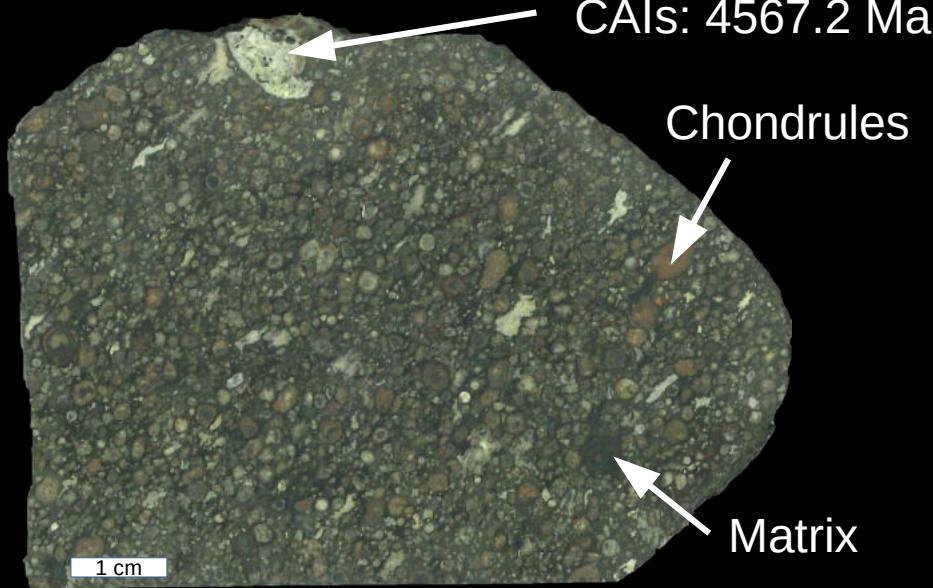


Isotopic anomalies in meteorites



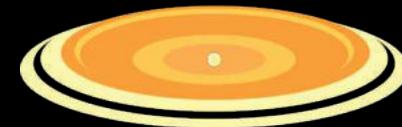
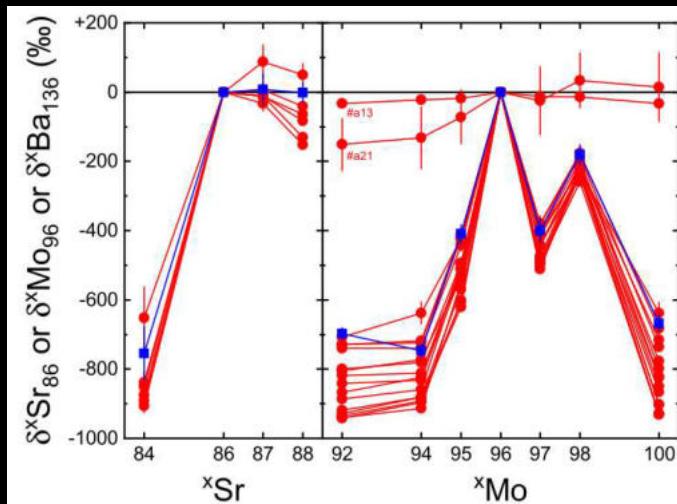
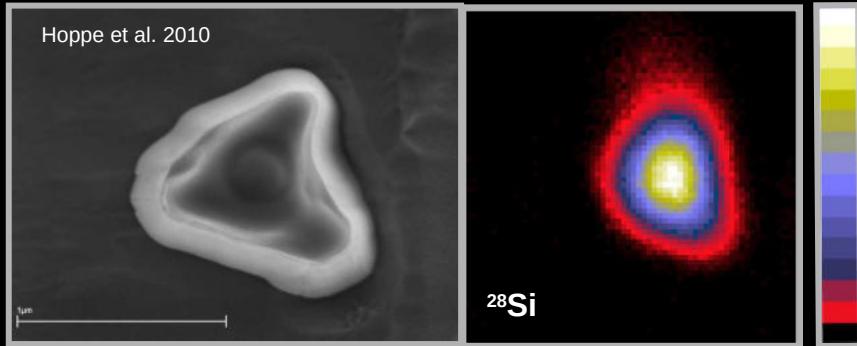
What is the solar system made of?

Chondrites



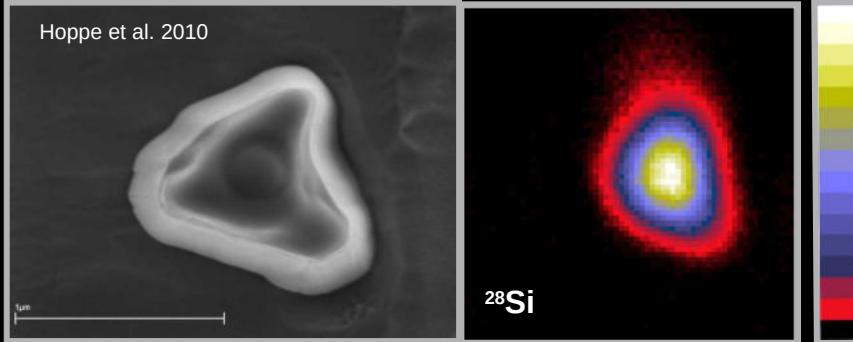
Presolar grains

(Lewis et al., 1987)



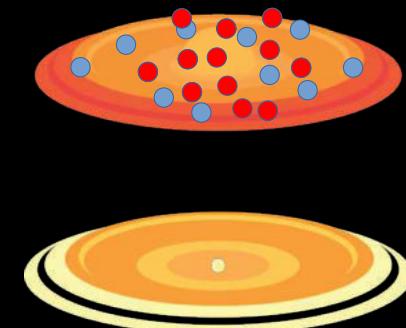
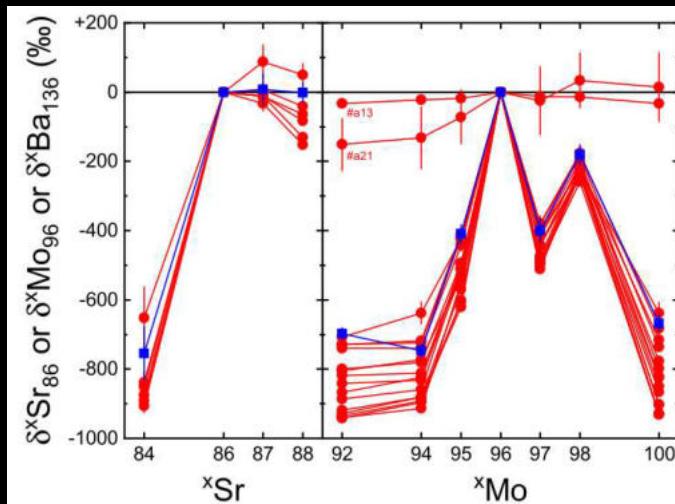
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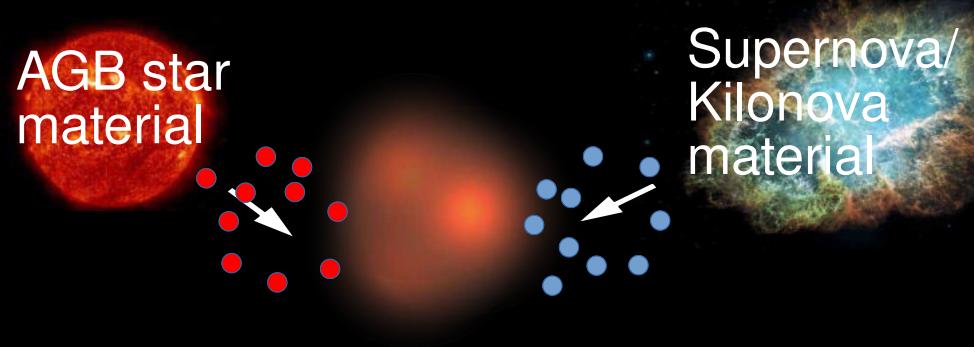
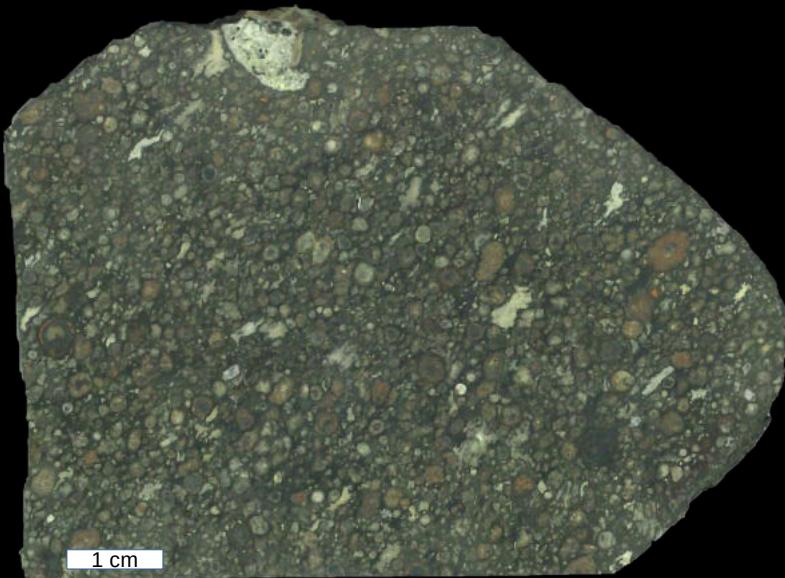


AGB star
material

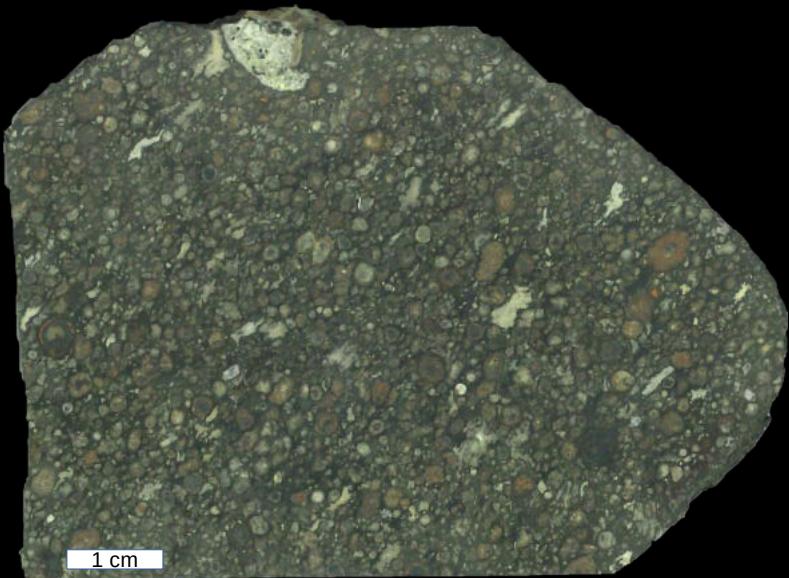
Supernova/
Kilonova
material



Short-lived nuclides

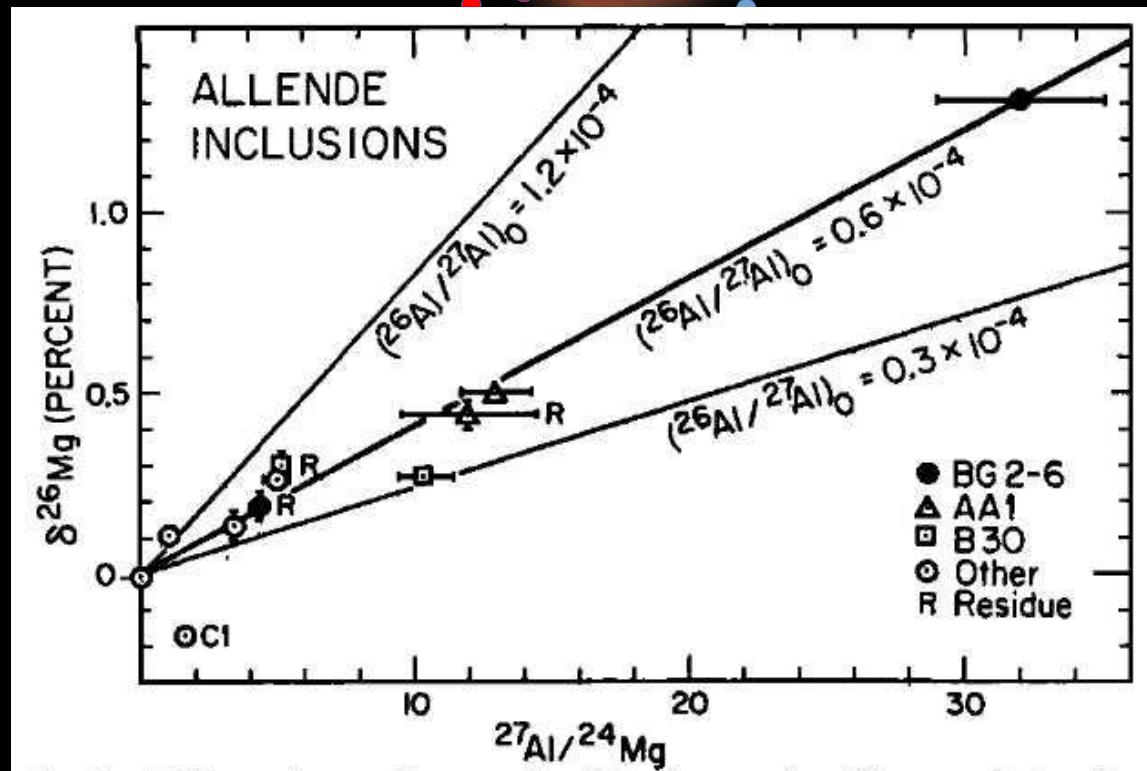


Short-lived nuclides

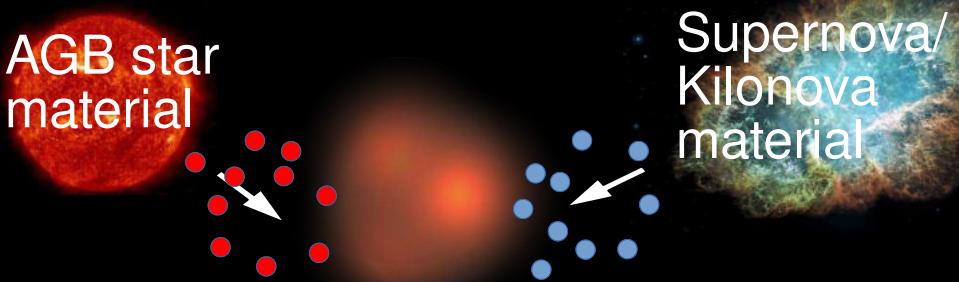


AGB star
material

Supernova/
Kilonova
material

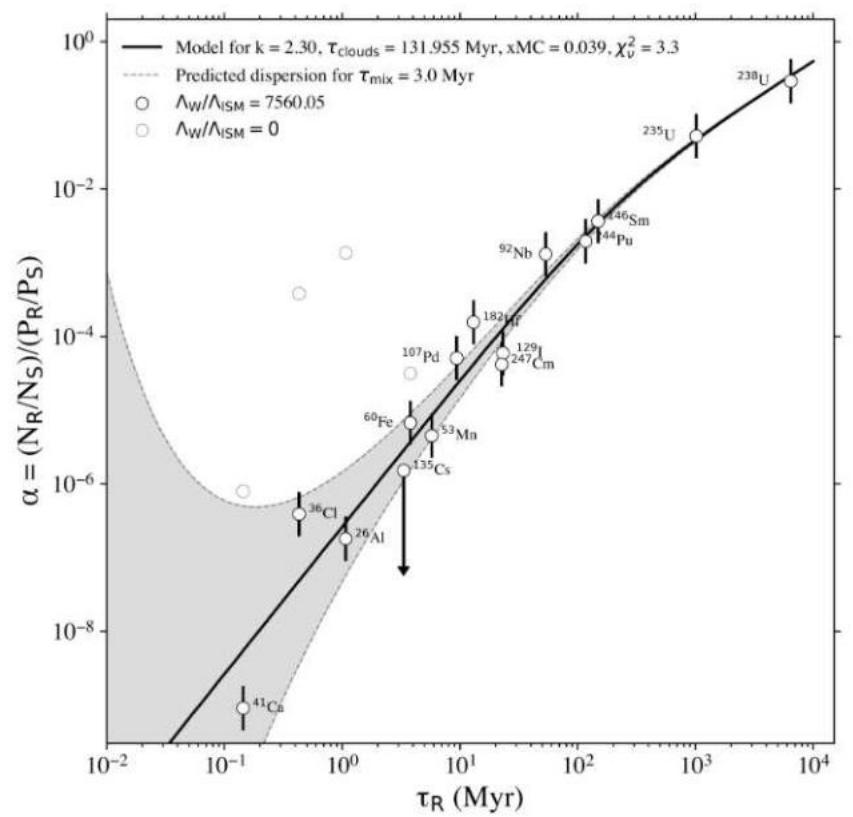
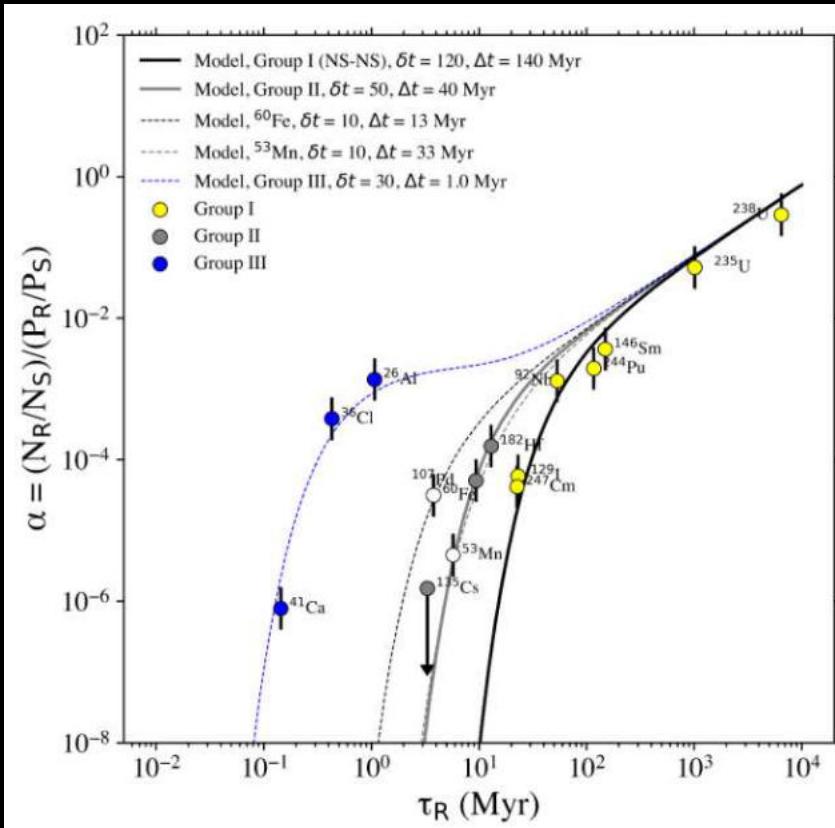
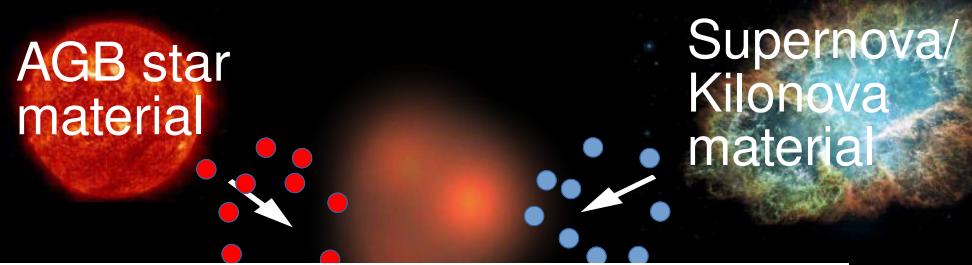


Short-lived nuclides

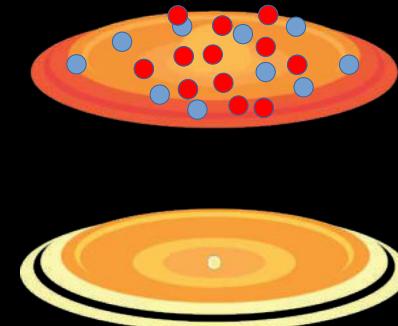
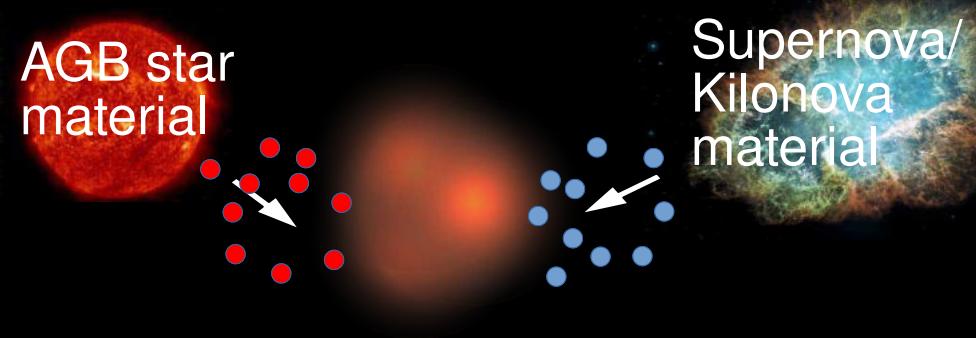
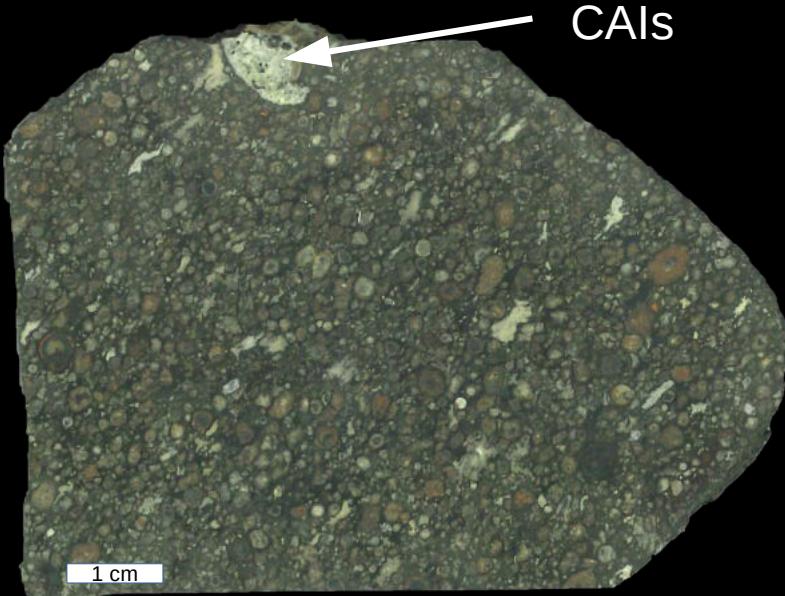


SLR	Daughter	Half-life (Myr)	Ref.	Solar System Abundance at $t=0$	Ref.	ISM Production Ratios	Ref.
^{41}Ca	^{41}K	0.0994 ± 0.0015	[1]	$^{41}\text{Ca}/^{40}\text{Ca} \approx 4.2 \times 10^{-9}$	[2]	$^{41}\text{Ca}/^{40}\text{Ca} = 2.3 \times 10^{-3}$	[3]
^{36}Cl	$^{36}\text{Ar}, ^{36}\text{S}$	$0.301 \pm 0.002^\dagger$		$^{36}\text{Cl}/^{35}\text{Cl} \approx (1.7 - 3) \times 10^{-5}$	[4]	$^{36}\text{Cl}/^{35}\text{Cl} = 2.63 \times 10^{-2}$	[5]
^{26}Al	^{26}Mg	0.717 ± 0.017	[6]	$^{26}\text{Al}/^{27}\text{Al} = 5.23 \times 10^{-5}$	[7]	$^{26}\text{Al}/^{27}\text{Al} = 1.667 \times 10^{-2}$	[8]
^{10}Be	^{10}B	1.387 ± 0.012	[9]	$^{10}\text{Be}/^{9}\text{Be} = (7.1 \pm 0.2) \times 10^{-4}$	[10]		
^{60}Fe	^{60}Fe	2.62 ± 0.04	[11]	$^{60}\text{Fe}/^{56}\text{Fe} = (0.9 \pm 0.1) \times 10^{-8}$	[12]	$^{60}\text{Fe}/^{56}\text{Fe} = 1.23 \times 10^{-4}$	[13]
^{53}Mn	^{53}Cr	3.98 ± 0.11	[14]	$^{53}\text{Mn}/^{55}\text{Mn} = (7.8 \pm 0.4) \times 10^{-6}$	[15]	$^{53}\text{Mn}/^{55}\text{Mn} = 7.52 \times 10^{-1}$	[16]
^{107}Pd	^{107}Ag	6.5 ± 0.3	[17]	$^{107}\text{Pd}/^{108}\text{Pd} \approx (7.7 \pm 0.5) \times 10^{-5}$	[18]	$^{107}\text{Pd}/^{108}\text{Pd} = 6.5 \times 10^{-1}$	[19]
^{182}Hf	^{182}W	8.896 ± 0.089	[20]	$^{182}\text{Hf}/^{180}\text{Hf} = (1.04 \pm 0.1) \times 10^{-4}$	[21]	$^{182}\text{Hf}/^{180}\text{Hf} = 2.9 \times 10^{-1}$	[22]
^{247}Cm	^{235}U	$15.6 \pm 0.5^\dagger$		$^{247}\text{Cm}/^{235}\text{U} = (5.6 \pm 0.3) \times 10^{-5}$	[23]	$^{247}\text{Cm}/^{232}\text{Th} = 1.01 \times 10^{-1}$	[24]
^{129}I	^{129}Xe	16.14 ± 0.12	[25]	$^{129}\text{I}/^{127}\text{I} = (1.74 \pm 0.02) \times 10^{-4}$	[26]	$^{129}\text{I}/^{127}\text{I} = 1.25 \times 10^0$	[27]
^{205}Pb	^{205}Tl	$17.3 \pm 0.7^\dagger$		$^{205}\text{Pb}/^{204}\text{Pb} = (1.4 \pm 0.3) \times 10^{-4}$	[28]		
^{92}Nb	^{92}Zr	$34.7 \pm 0.7^\dagger$		$^{92}\text{Nb}/^{93}\text{Nb} = (1.7 \pm 0.6) \times 10^{-5}$	[29]	$^{92}\text{Nb}/^{93}\text{Nb} = 5.65 \times 10^{-3}$	[30]
^{146}Sm	^{142}Nd	103 ± 5	[31]	$^{146}\text{Sm}/^{144}\text{Sm} = (8.28 \pm 0.44) \times 10^{-3}$	[32]	$^{146}\text{Sm}/^{144}\text{Sm} = 9.5 \times 10^{-1}$	[33]
^{244}Pu	$^{236}\text{U}, ^{232}\text{Th}$	$80.0 \pm 0.9^\dagger$		$^{244}\text{Pu}/^{238}\text{U} = (7 \pm 1) \times 10^{-3}$	[34]	$^{244}\text{Pu}/^{232}\text{Th} = 6.67 \times 10^{-1}$	[35]
^{135}Cs	^{135}Ba	2.3 ± 0.3^a		$^{135}\text{Cs}/^{133}\text{Cs} = (4.8 \pm 0.8) \times 10^{-4}$	[36]	$^{135}\text{Cs}/^{133}\text{Cs} = 8.0 \times 10^{-1}$	[37]
				$^{135}\text{Cs}/^{133}\text{Cs} < 2.8 \times 10^{-6}$			

Short-lived nuclides

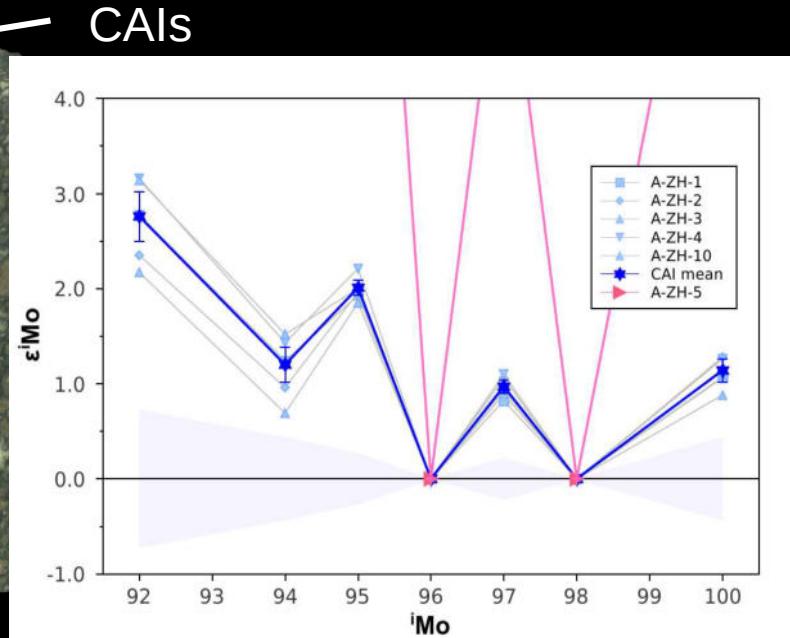


Isotopic anomalies

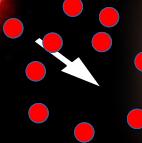


How did disk form? /disk dynamics?

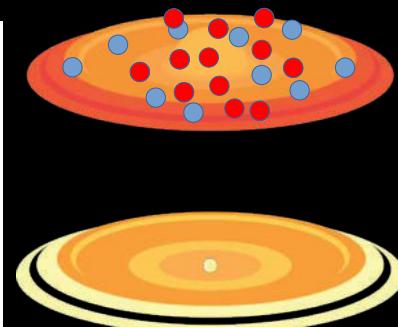
Isotopic anomalies



AGB star material



Supernova/
Kilonova
material



SN/r-process enrichment in CAIs

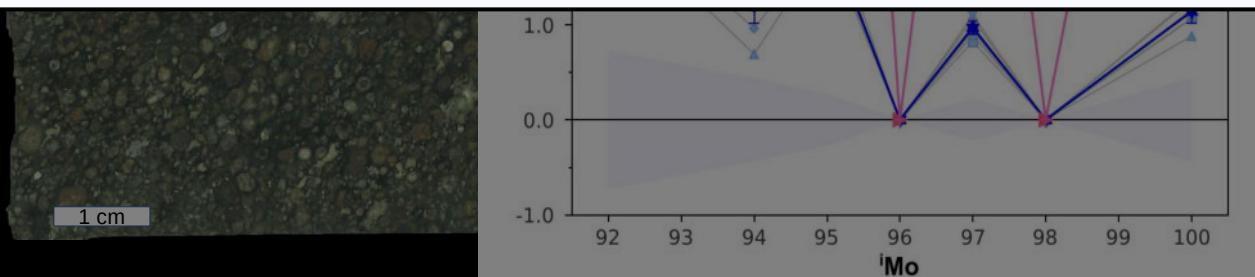
Isotopic anomalies

AGB star material

Supernova/
Kilonova
material

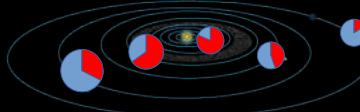
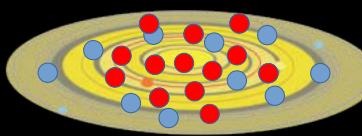
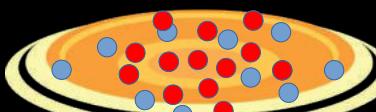
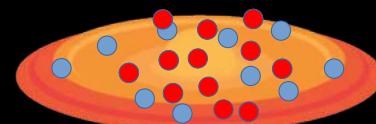


Materials forming in disk inherit cosmic memory!



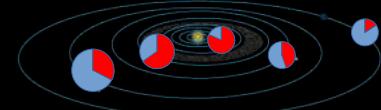
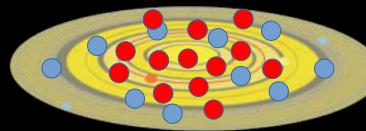
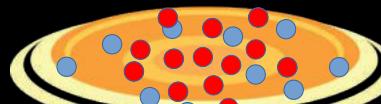
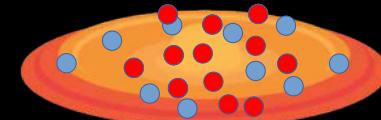
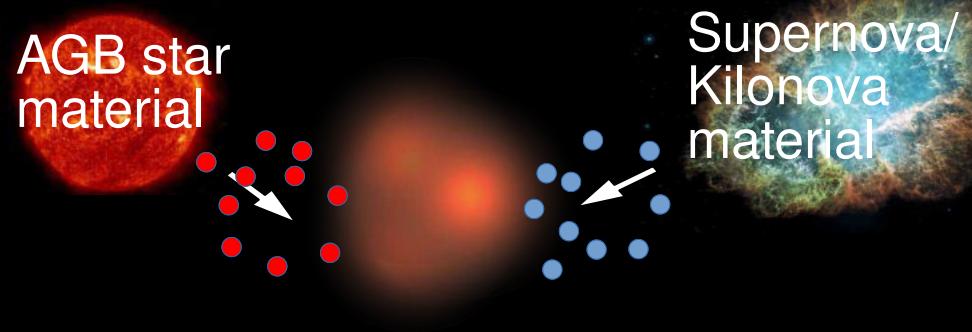
SN/r-process enrichment in CAIs

Distribution

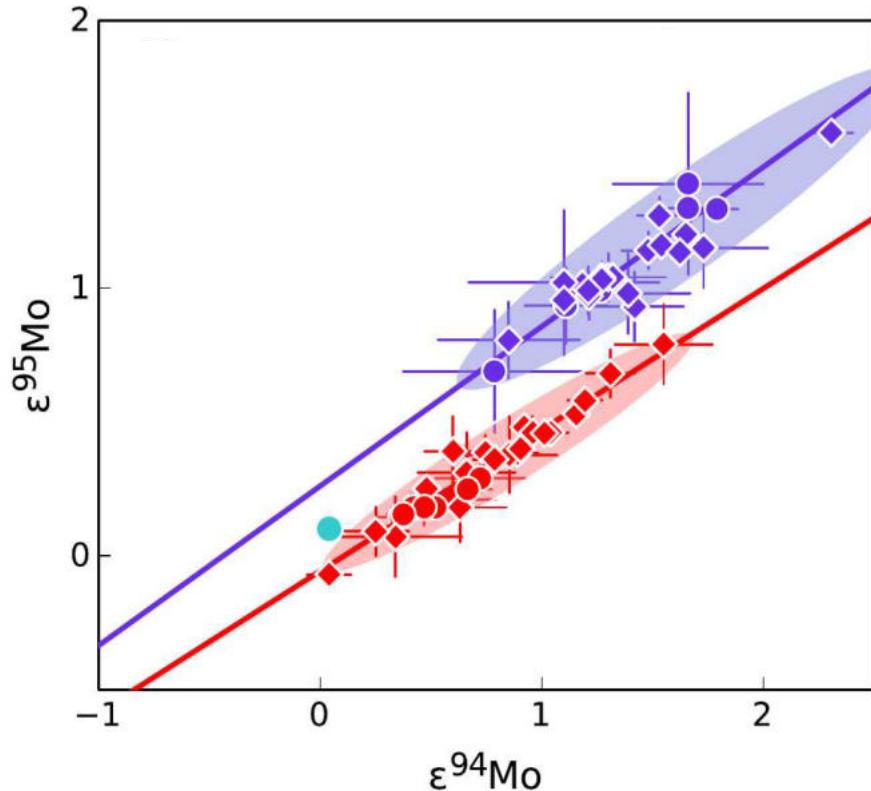
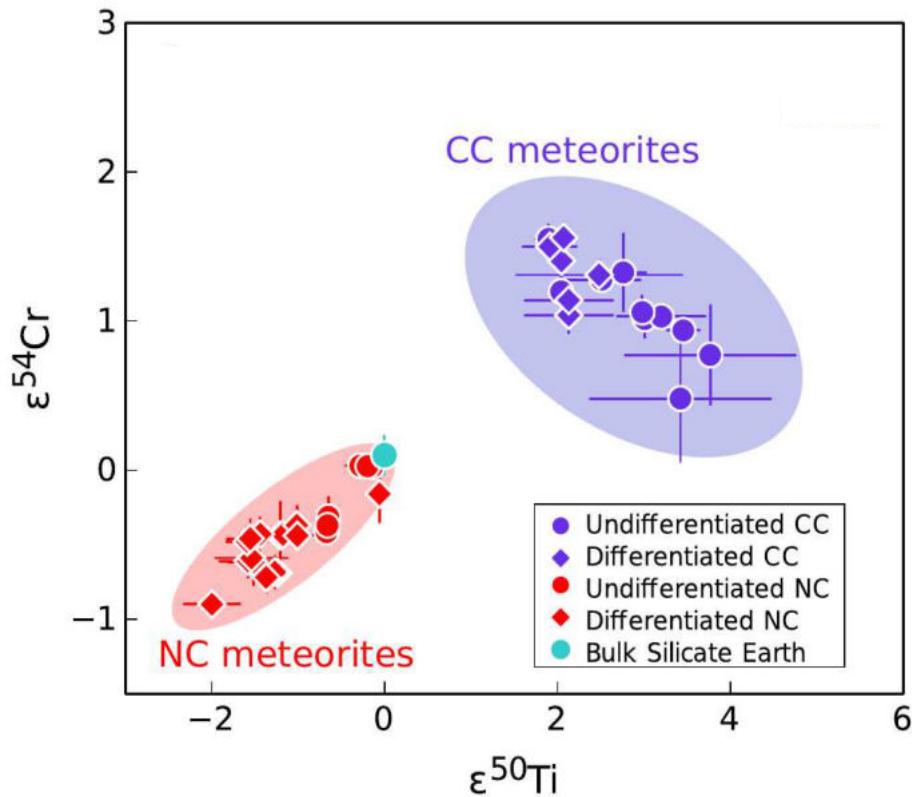


Distribution

Different bodies have distinct isotopic compositions



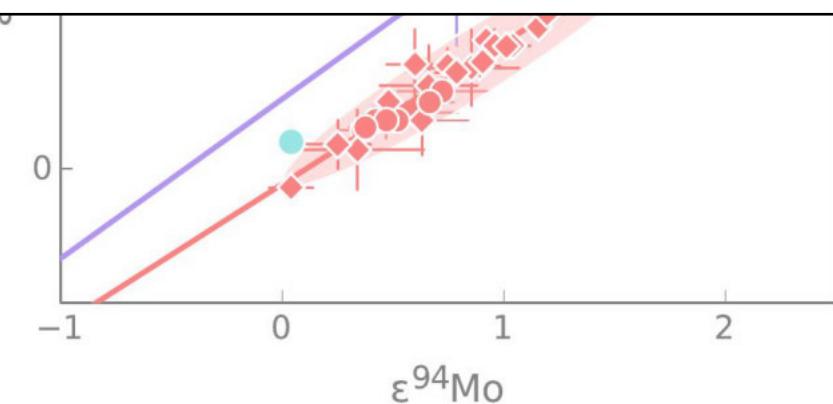
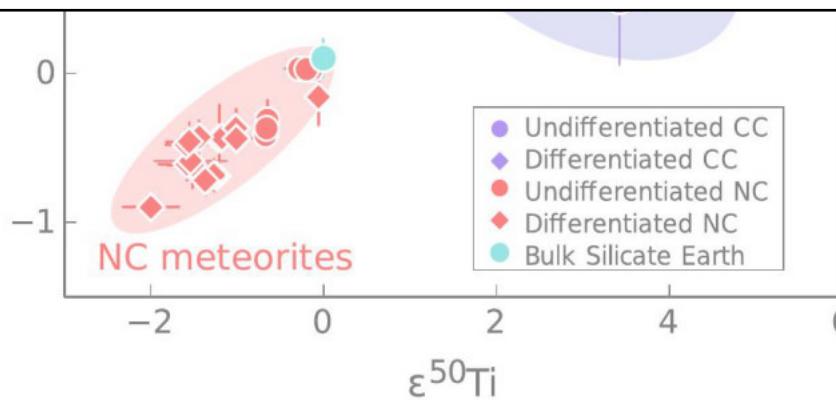
Planetary-scale nucleosynthetic isotope anomalies



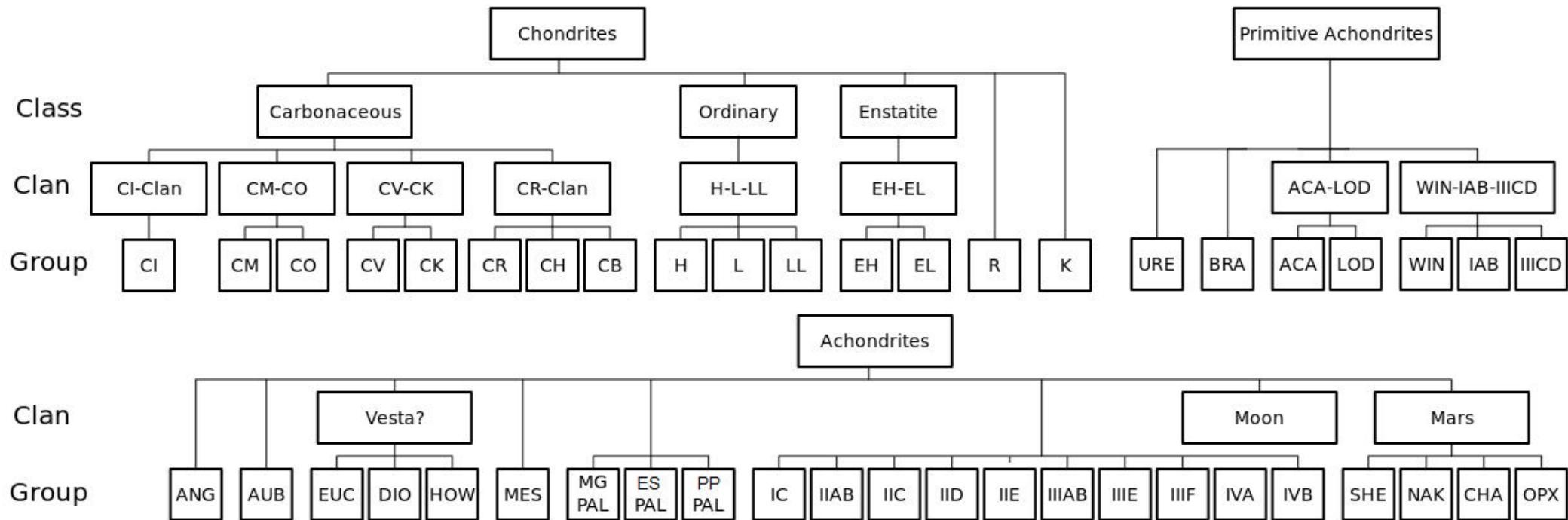
Planetary-scale nucleosynthetic isotope anomalies



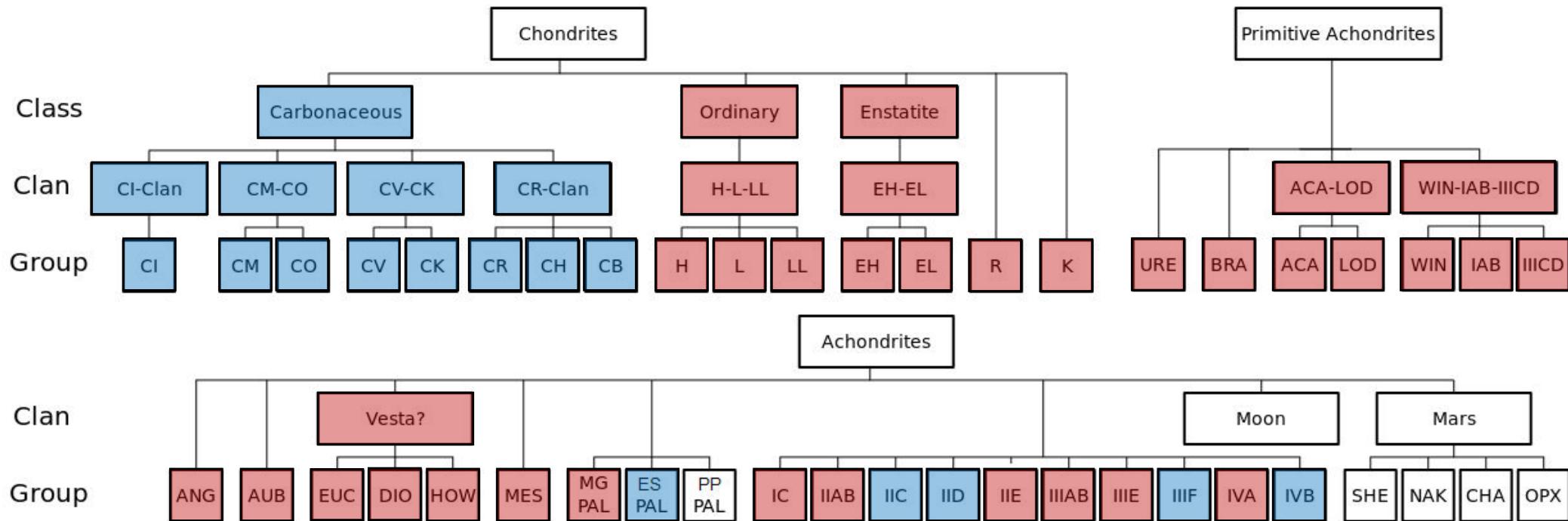
Fundamental isotopic dichotomy



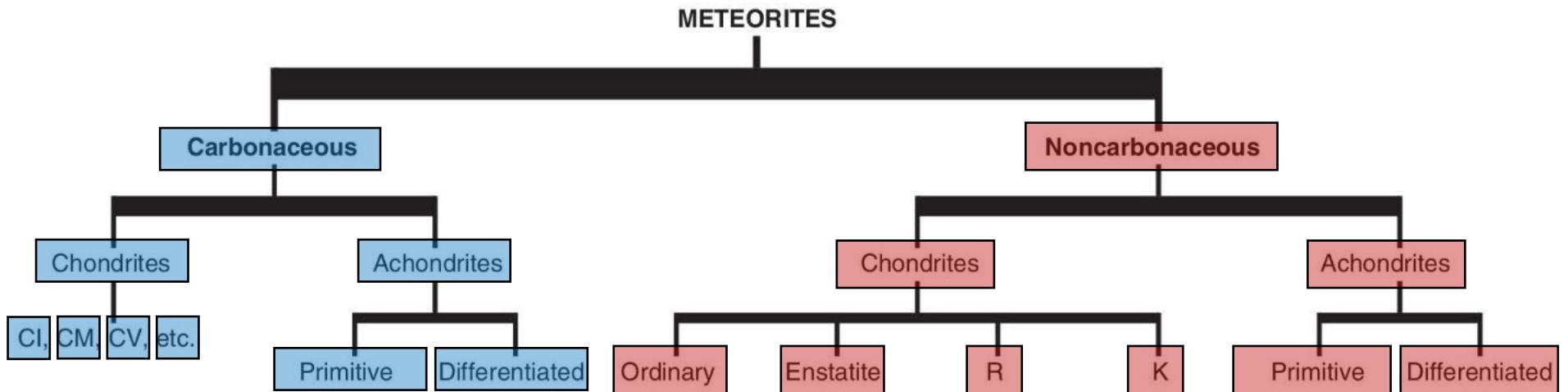
Phenomenological classification

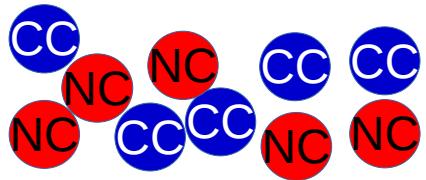
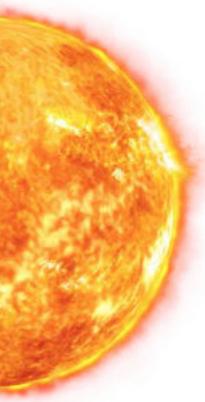


Genetic classification



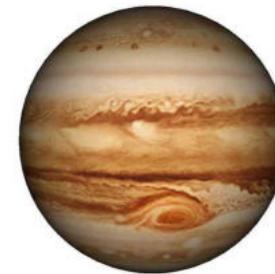
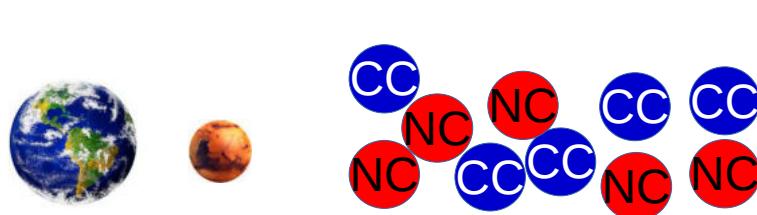
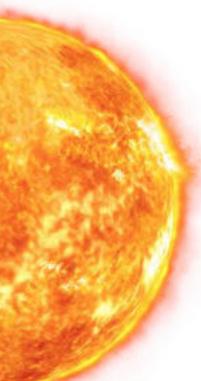
Genetic classification

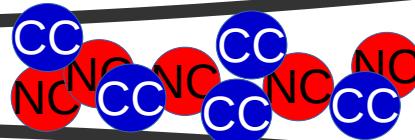




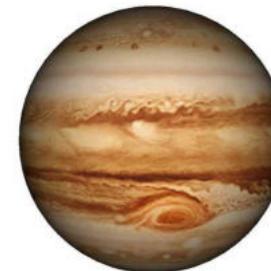


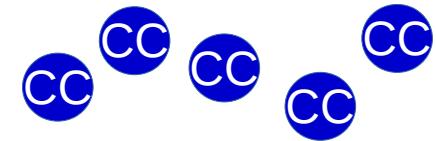
Temporal variation?



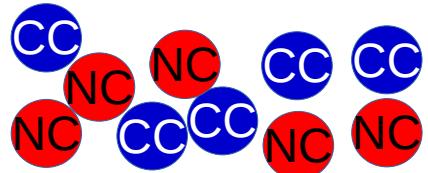


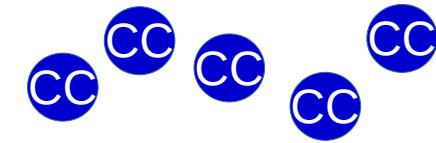
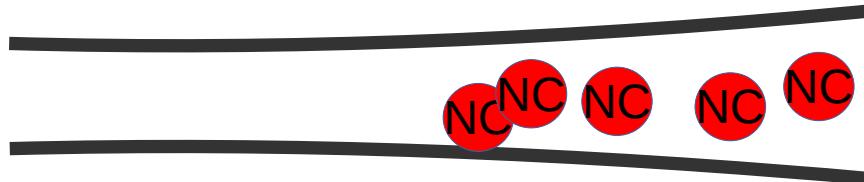
Temporal variation?



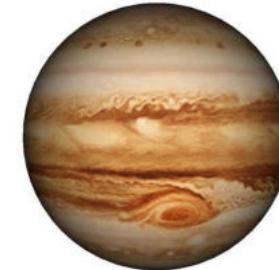
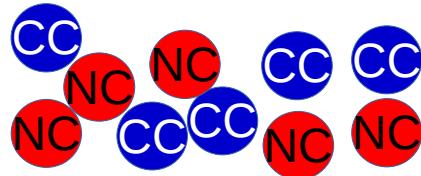


Spatial variation?

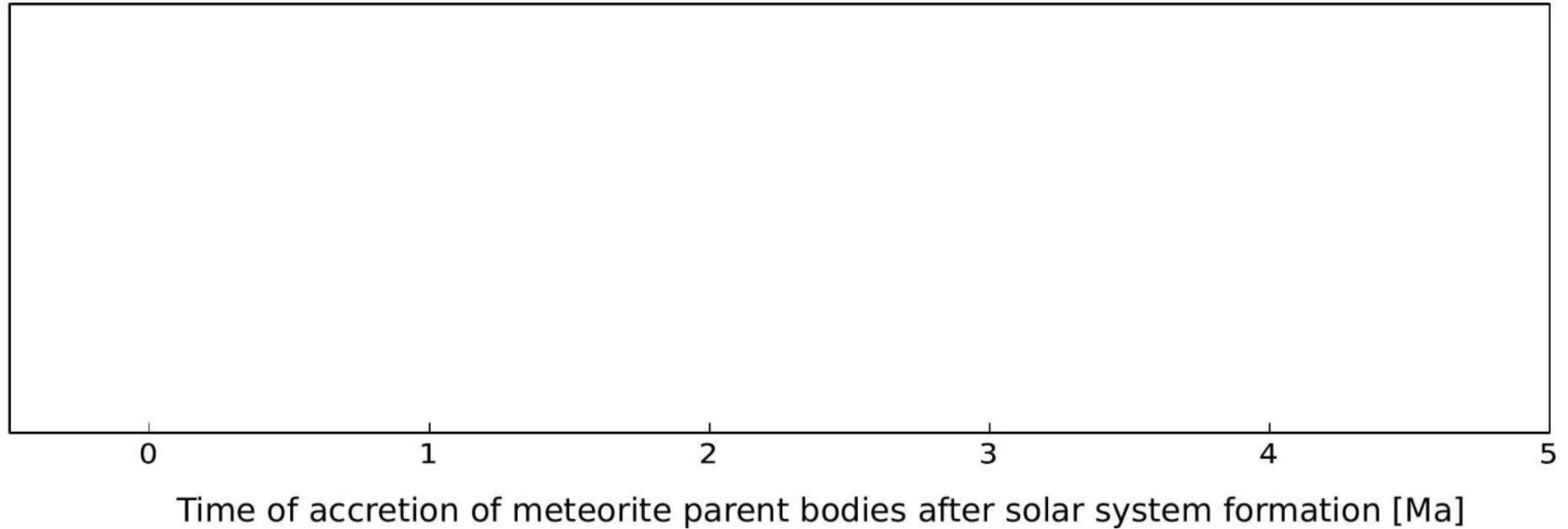




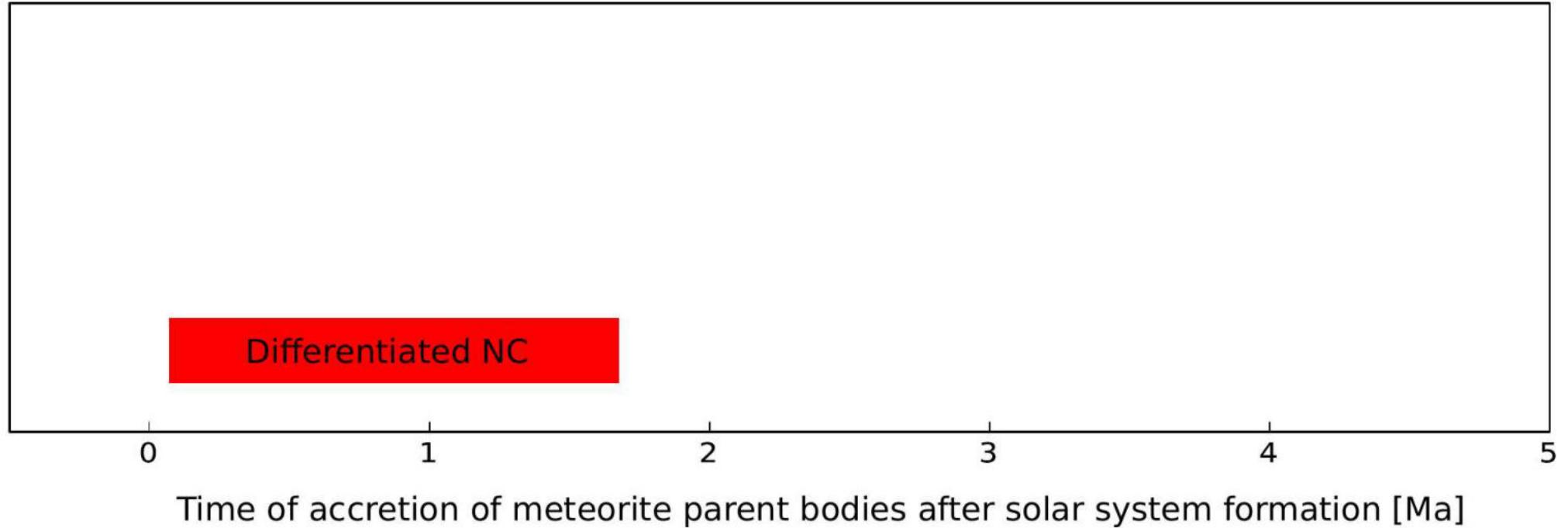
Spatial variation?



Approach: Combining age + anomaly information



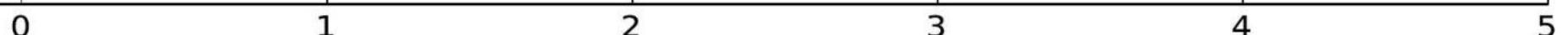
Combining age + anomaly information



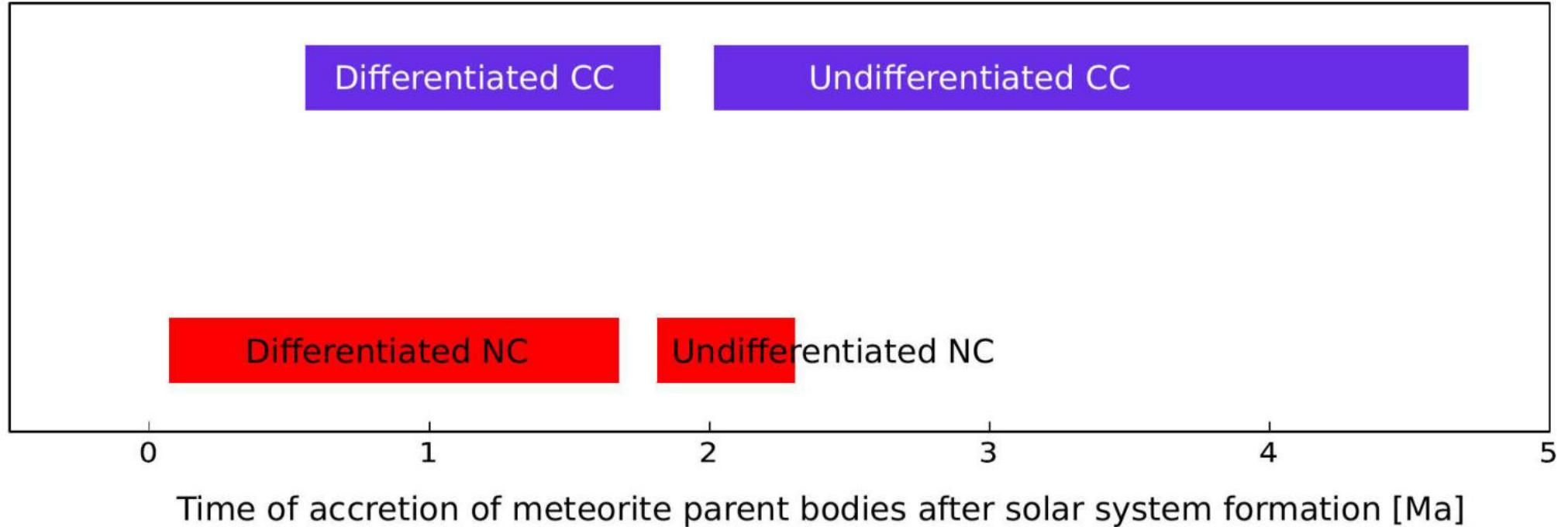
Combining age + anomaly information

Differentiated CC

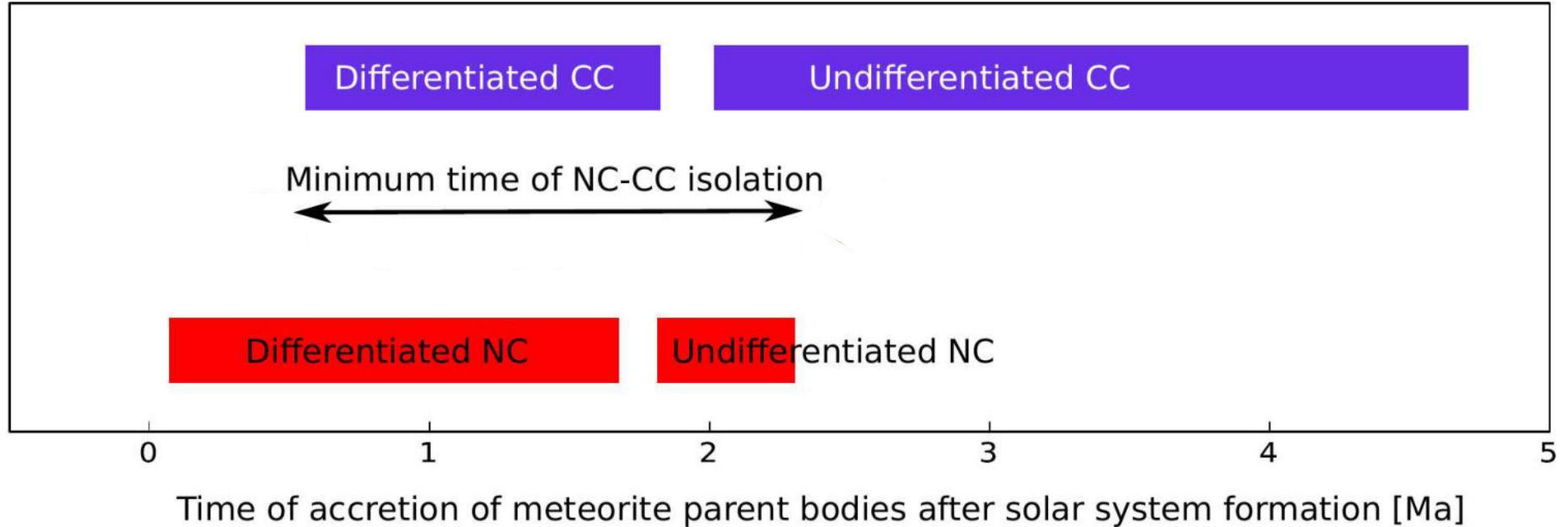
Differentiated NC



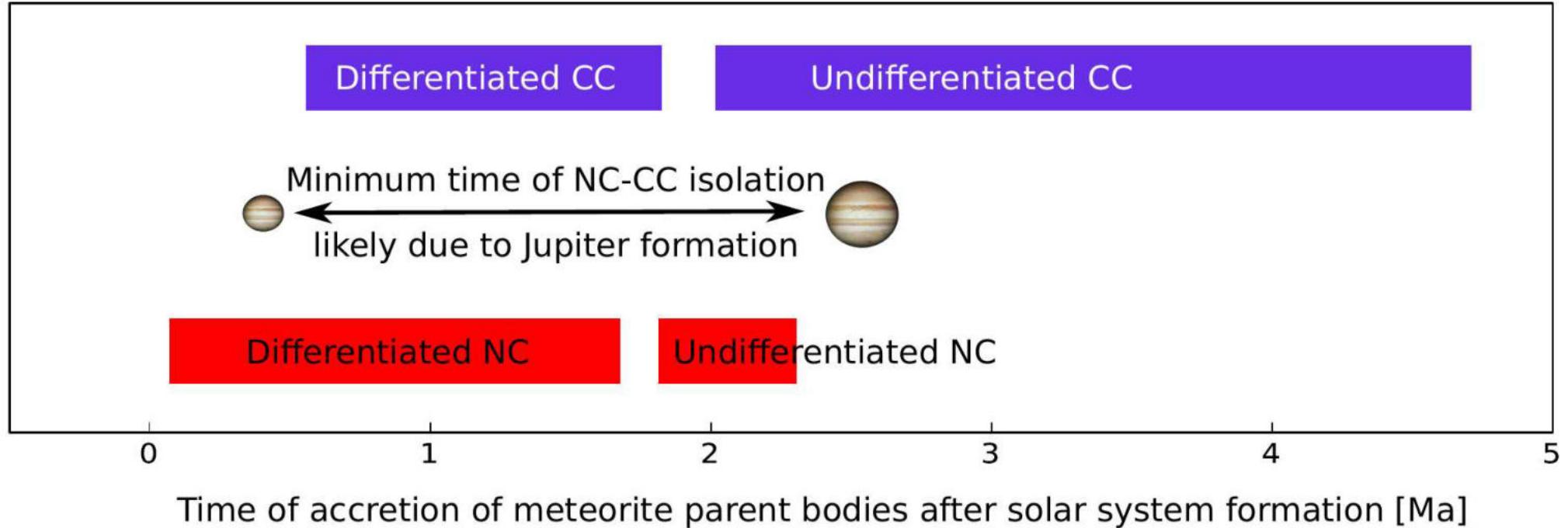
Combining age + anomaly information



Combining age + anomaly information

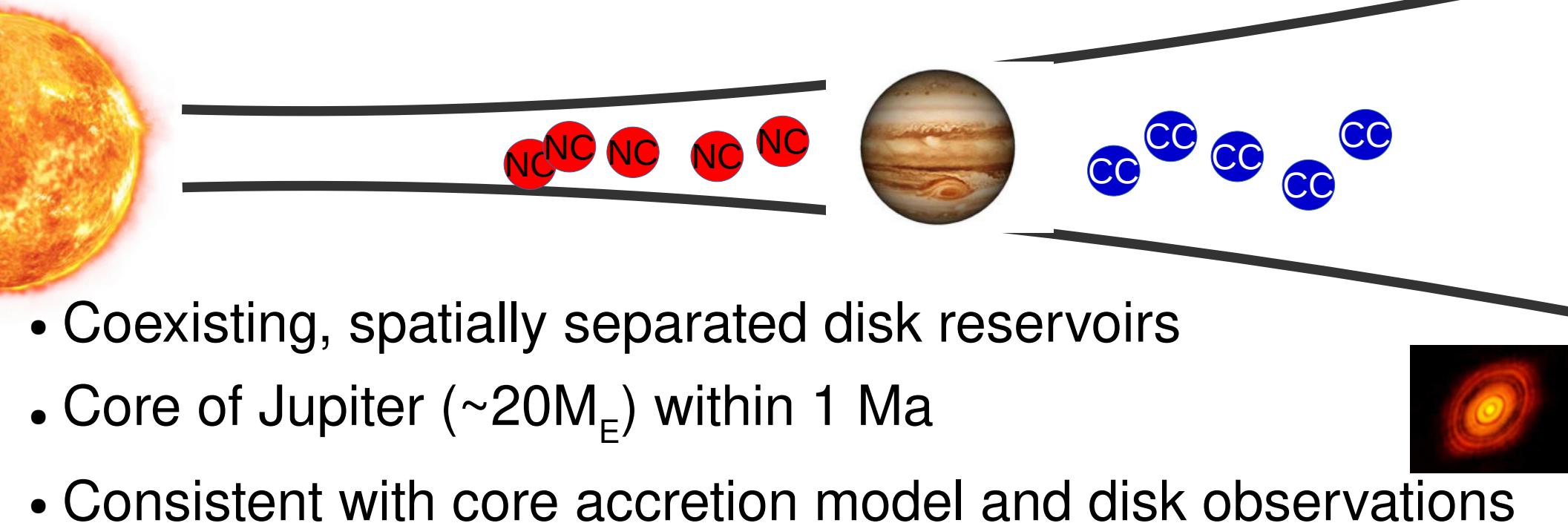


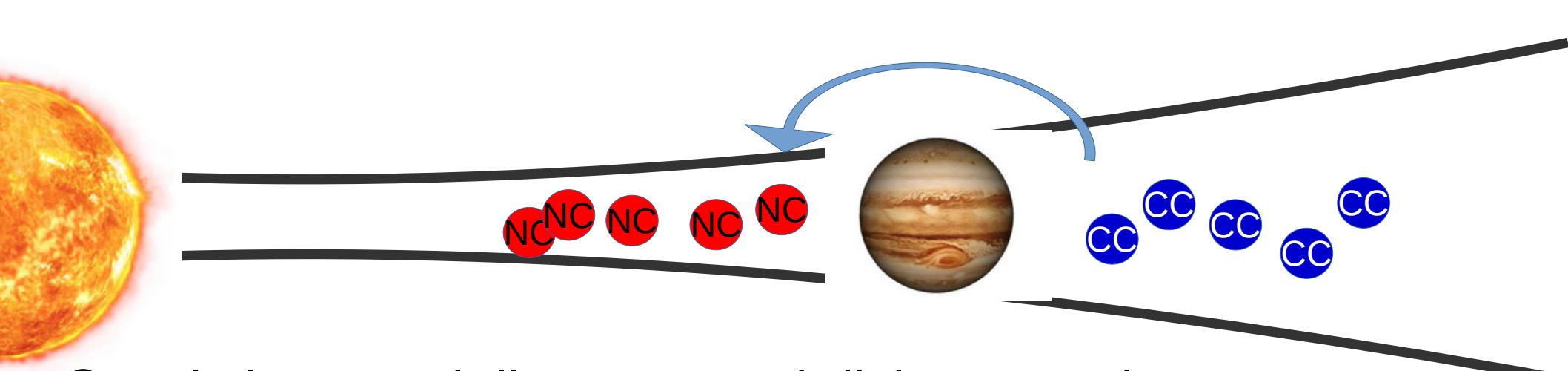
Combining age + anomaly information





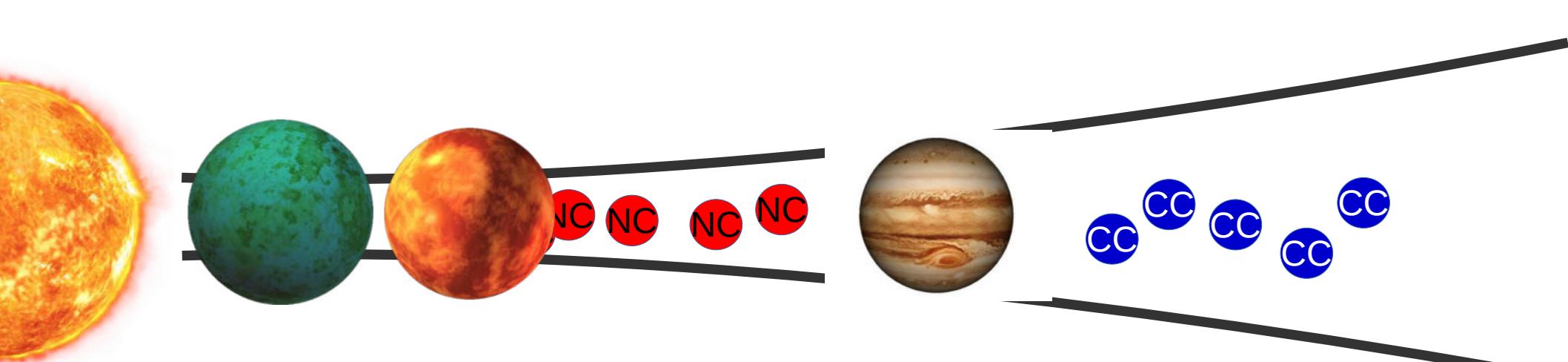
- Coexisting, spatially separated disk reservoirs



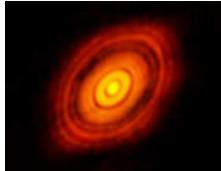


- Coexisting, spatially separated disk reservoirs
- Core of Jupiter ($\sim 20M_E$) within 1 Ma
- Consistent with core accretion model and disk observations
- Naturally explains scattering of CC bodies into inner solar system





- Coexisting, spatially separated disk reservoirs
- Core of Jupiter ($\sim 20M_E$) within 1 Ma
- Consistent with core accretion model and disk observations
- Naturally explains scattering of CC bodies into inner solar system
- Jupiter formation explains why solar system lacks super-Earth





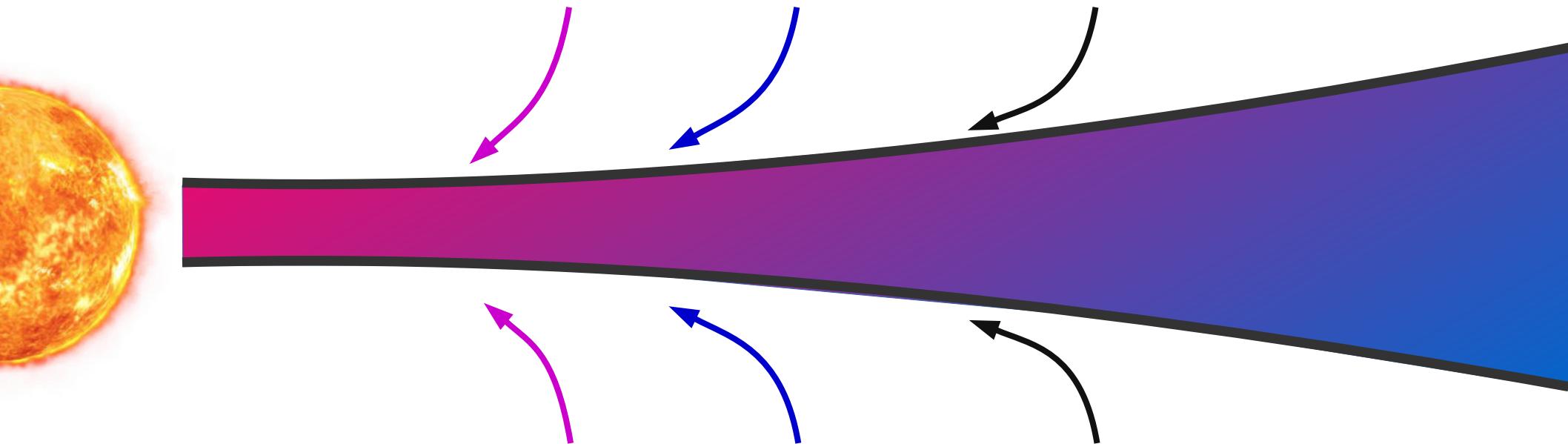
NC NC NC NC NC

CC CC CC CC CC CC

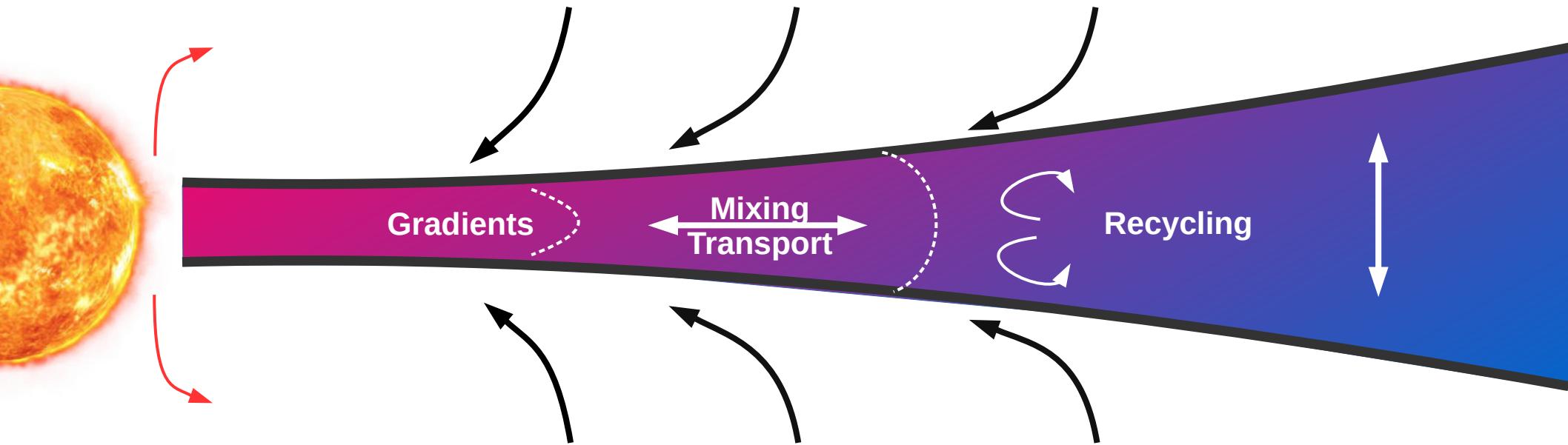


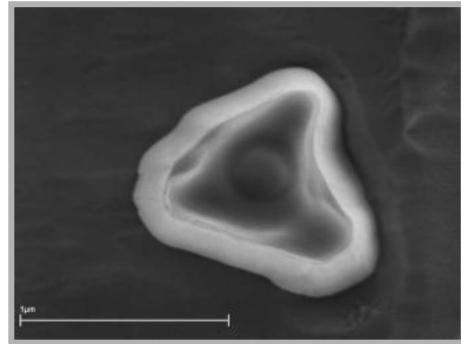
WHAT CONTROLS PLANETARY SCALE ANOMALIES?

Heterogeneous infall?



Disk processing?





Control by presolar grains or disk products?

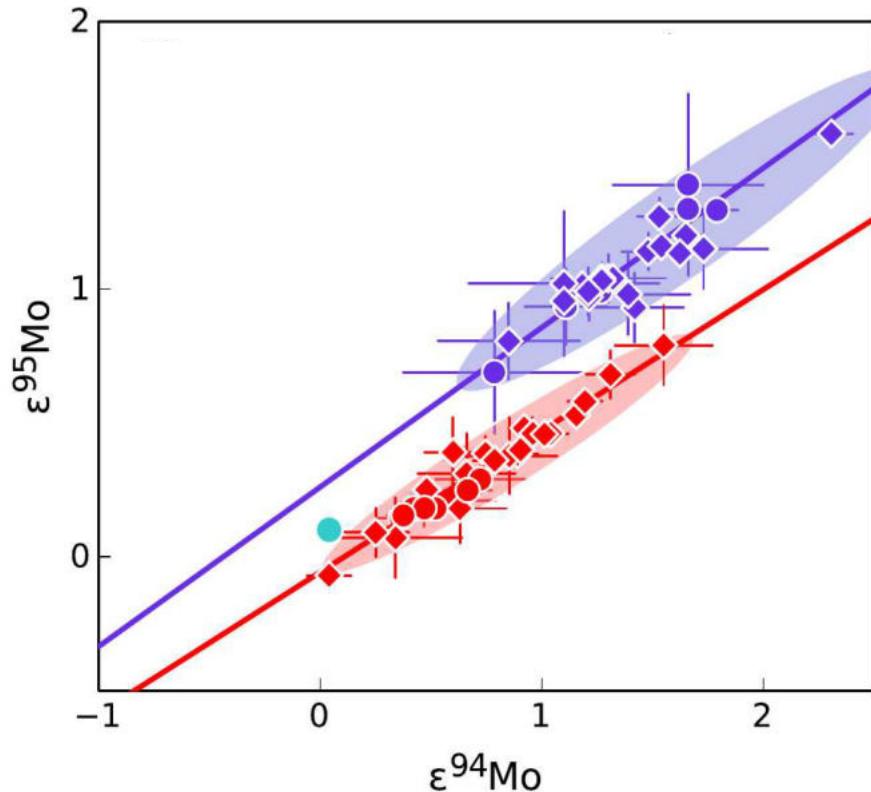
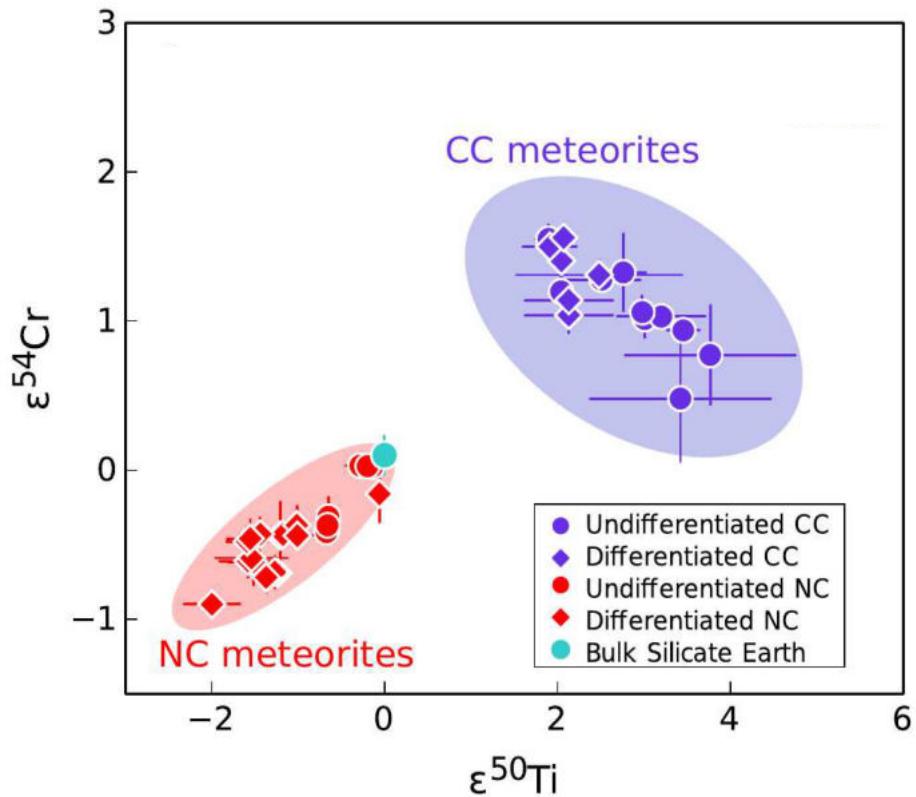


Approach

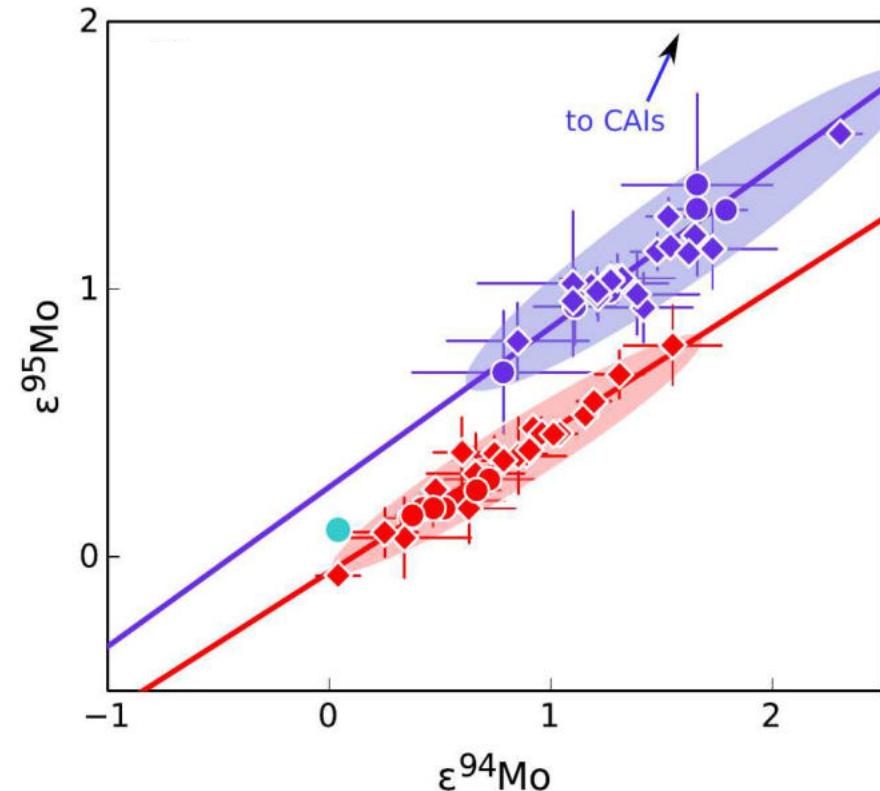
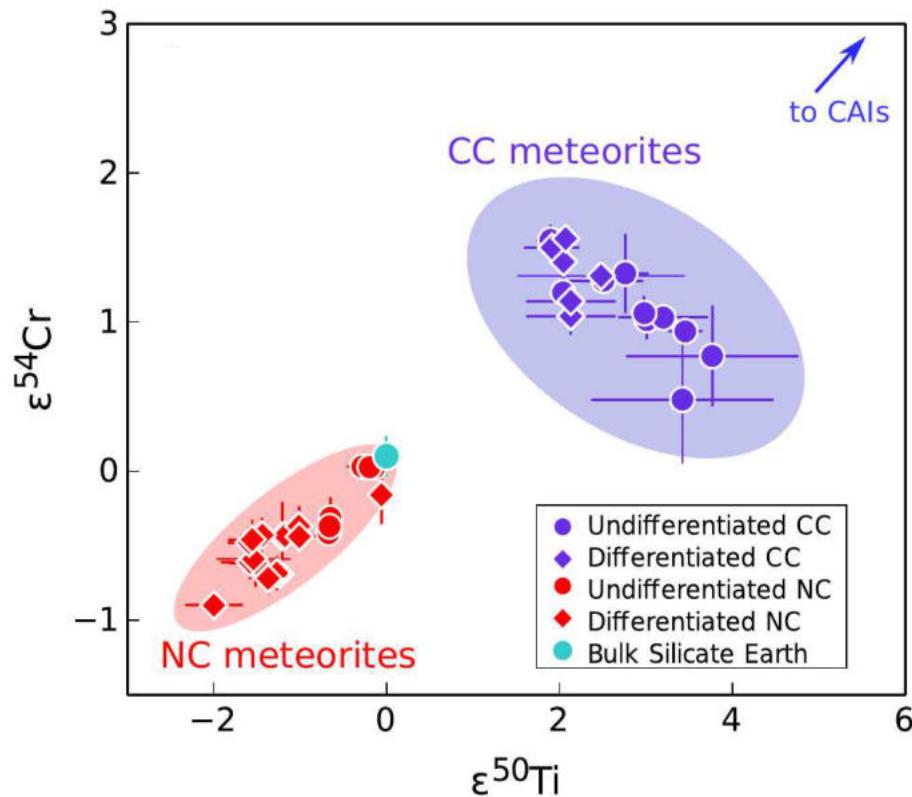
Investigate multi-element anomaly relations between:

- Presolar grains
- CAIs
- Chondrules
- Matrix
- Bulk bodies



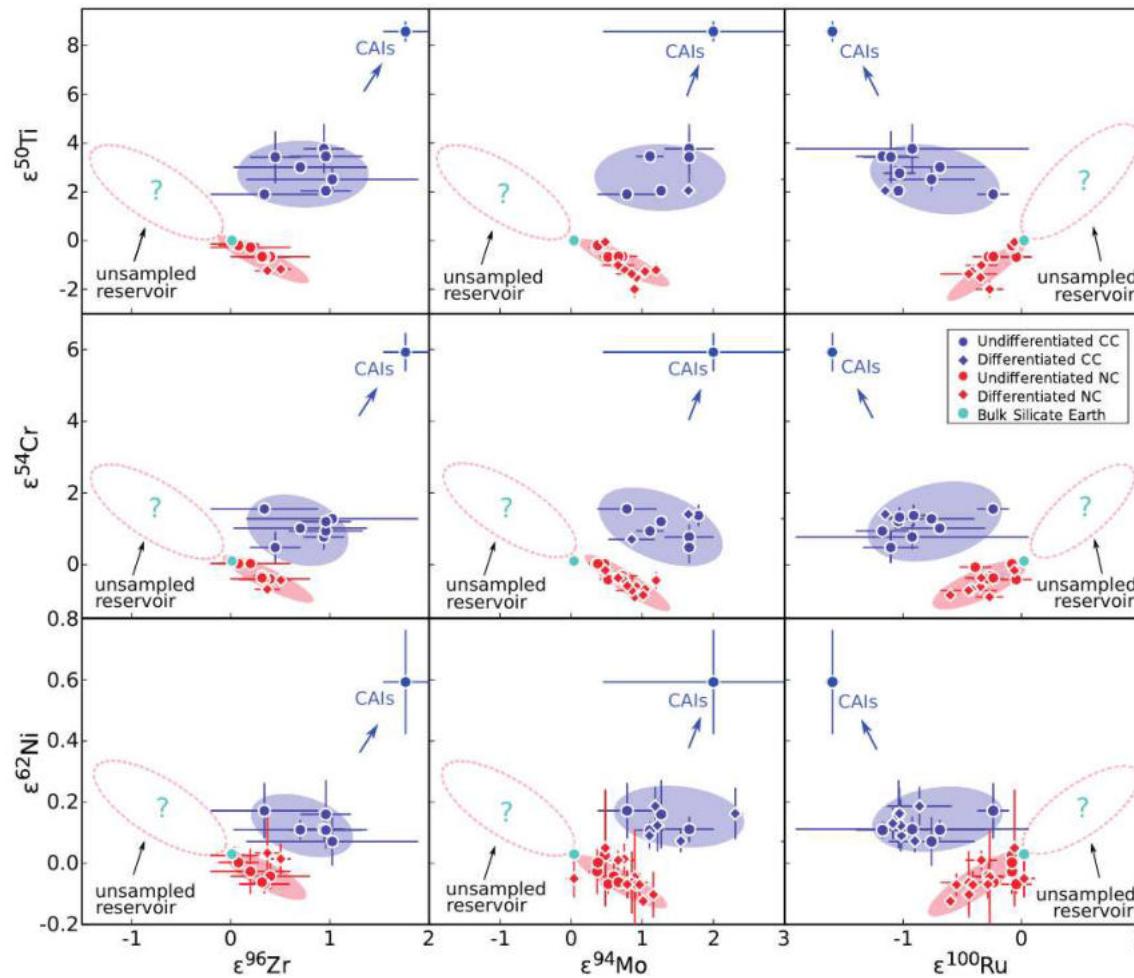


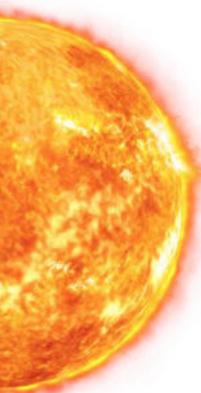
CC offset from NC towards CAI



Offset not limited to refractory elements!

CC offset from NC towards CAI



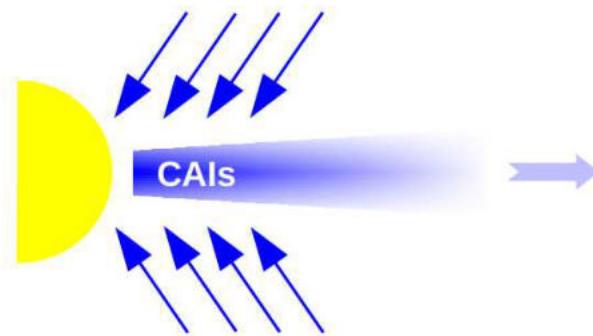
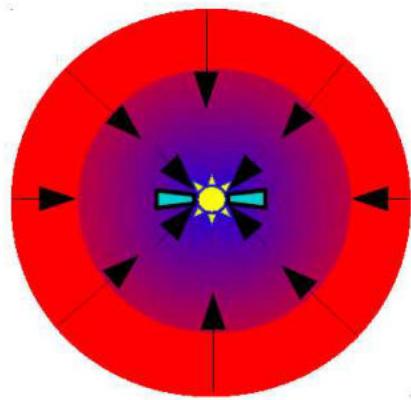


WHAT CONTROLS PLANETARY SCALE ANOMALIES?

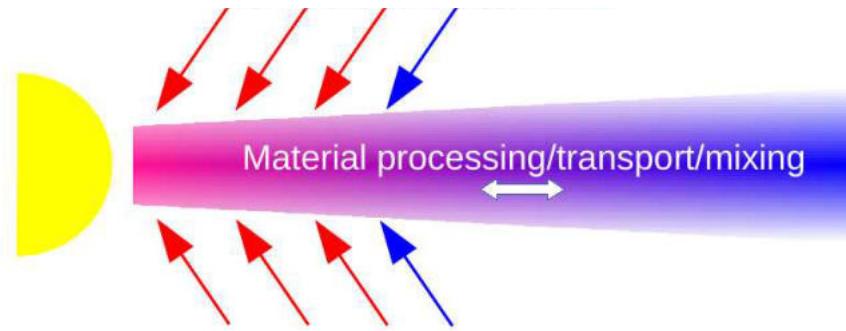
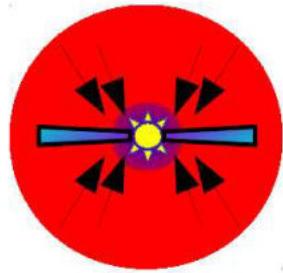
Variable mixing of disk materials!

CC = NC + inclusion-like chondritic (IC) reservoir

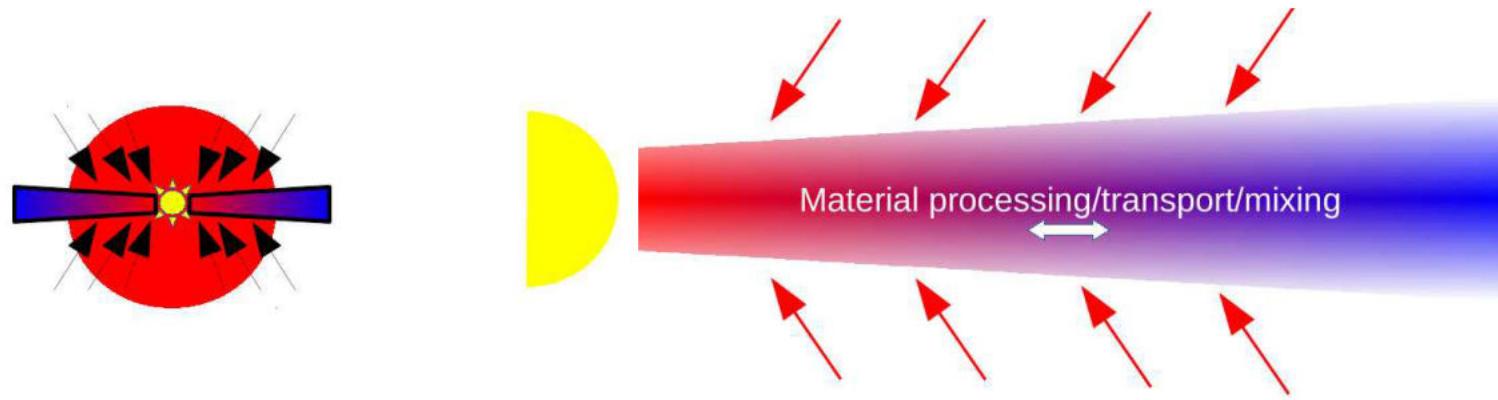
Model for the formation of the solar system



Model for the formation of the solar system



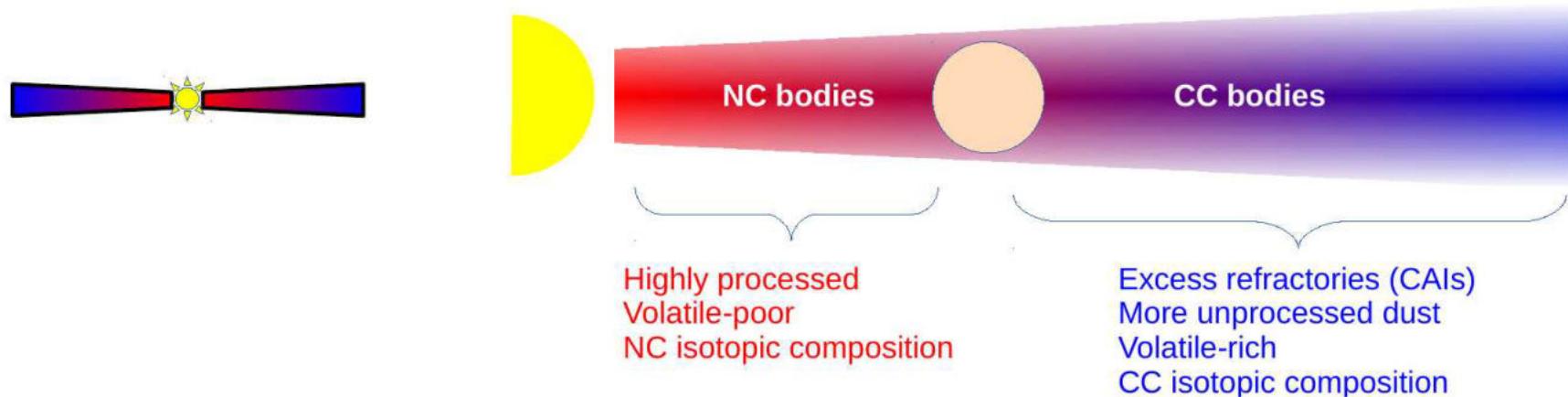
Model for the formation of the solar system



Model for the formation of the solar system



Model for the formation of the solar system



Model for the formation of the solar system

Isotopic variability set very early!

→ **Reinforces usability of anomalies for tracing genetics**

Volatile-poor
NC isotopic composition

More unprocessed dust
Volatile-rich
CC isotopic composition

How did the terrestrial planets form?

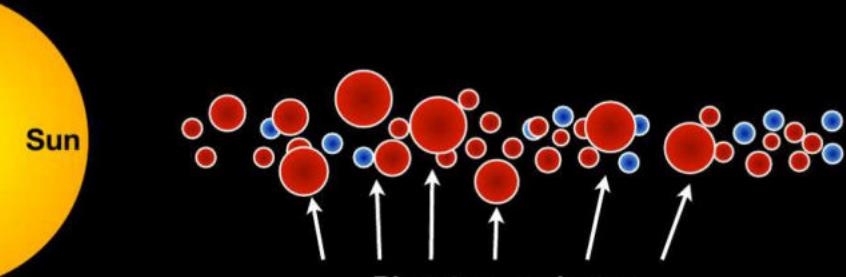


Oligarchic growth in 10s to 100s Ma
Limited outer solar system input

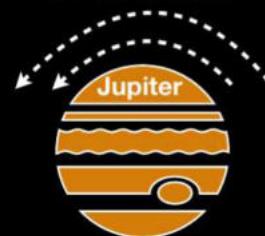


Pebble accretion in < 5 Ma
Significant outer solar system input

I. Classic accretion

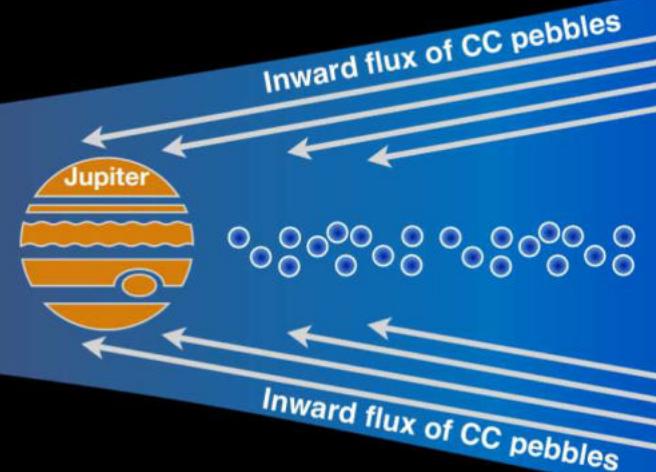
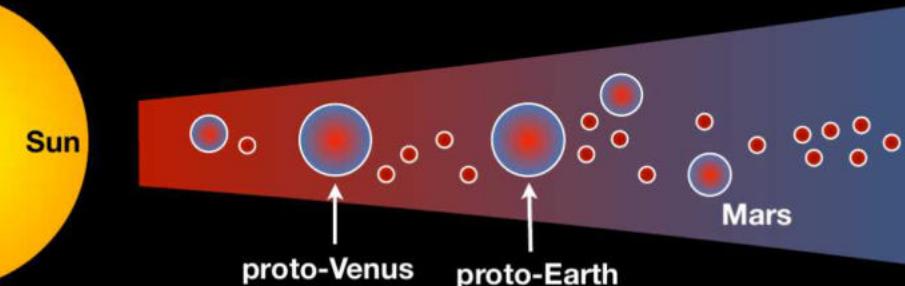


Inward scattering
of CC bodies



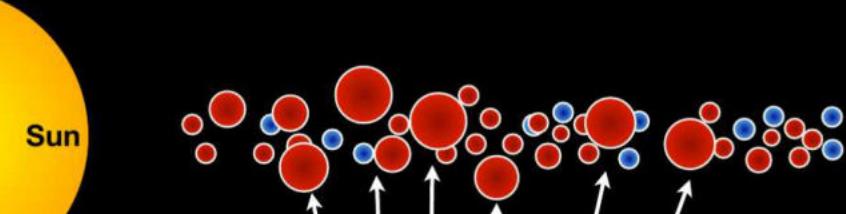
~5%

II. Pebble accretion

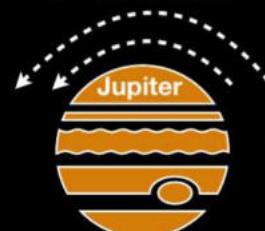


~50%

I. Classic accretion



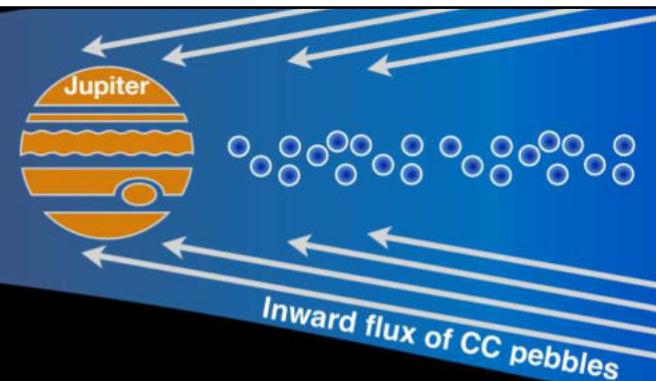
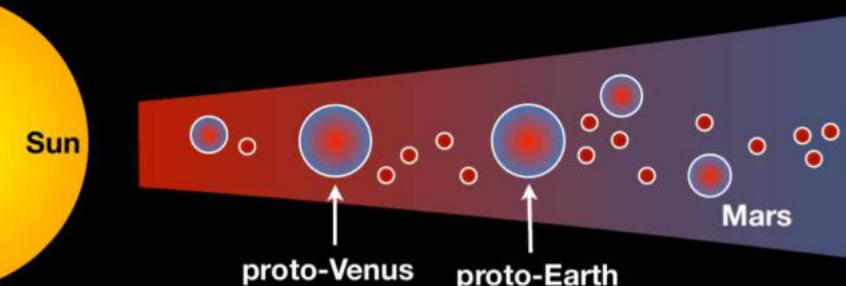
Inward scattering
of CC bodies



~5%

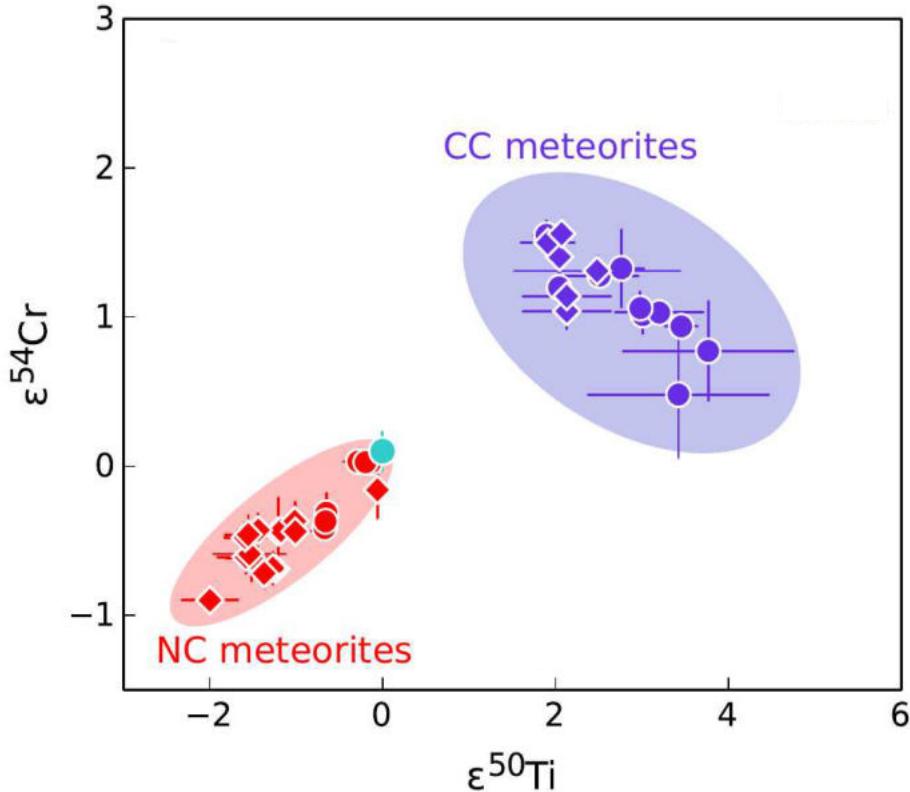
How much blue stuff is in the Earth?

II. P

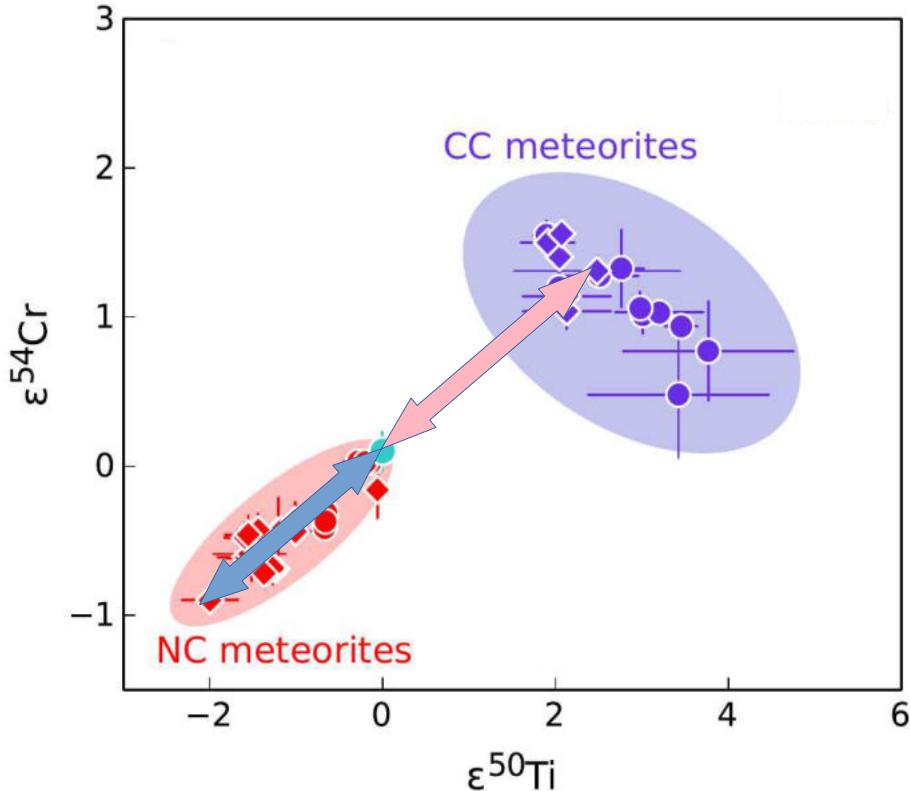


~50%

CC amount in Earth, Mars

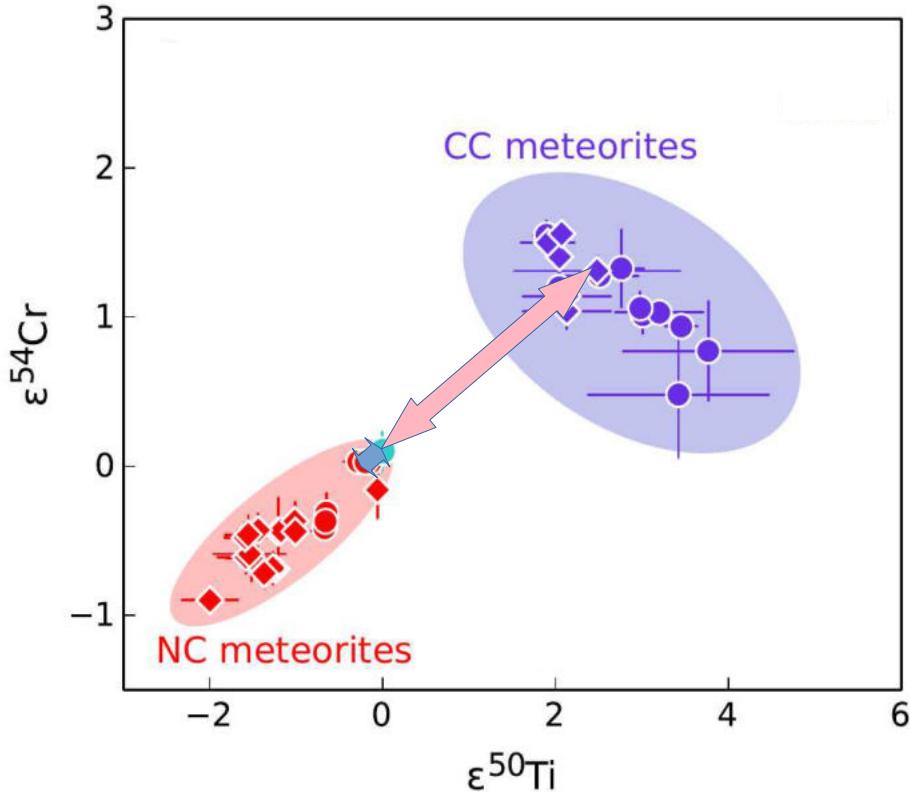


CC amount in Earth, Mars



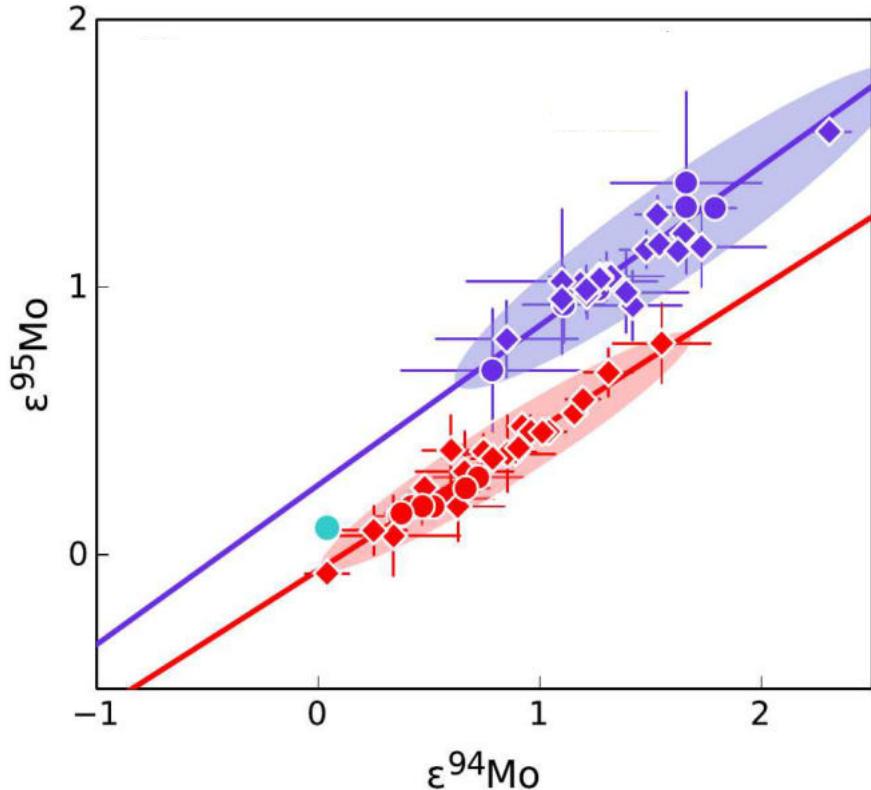
~40% CC

CC amount in Earth, Mars



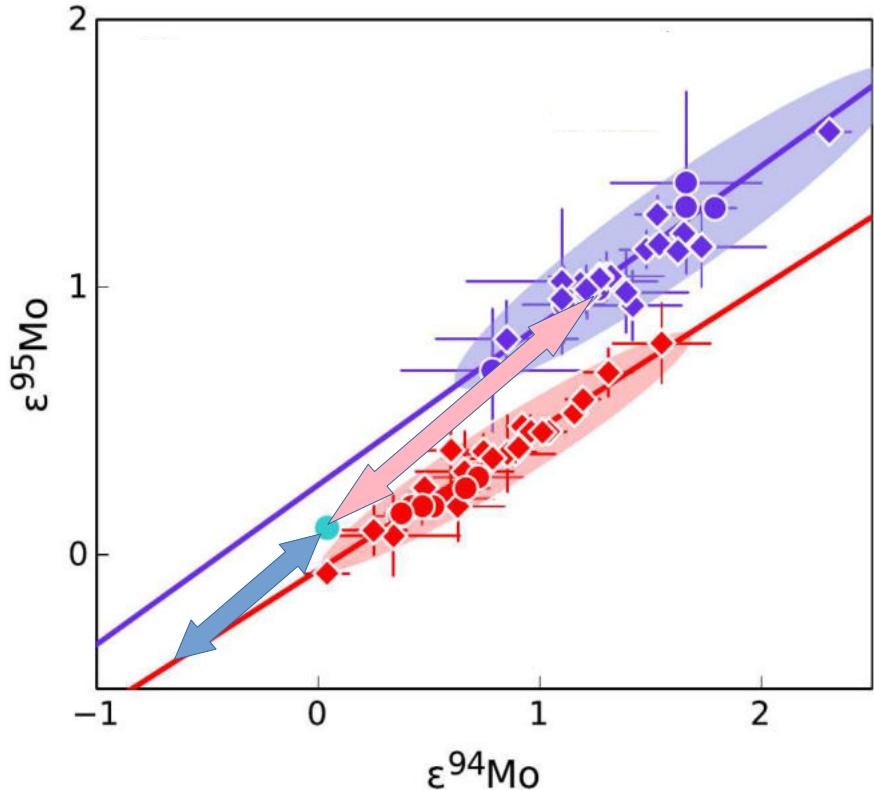
<5% CC

CC amount in Earth, Mars



Earth contains material unsampled by meteorites!

CC amount in Earth, Mars



Earth contains material unsampled by meteorites!

CC amount in Earth, Mars



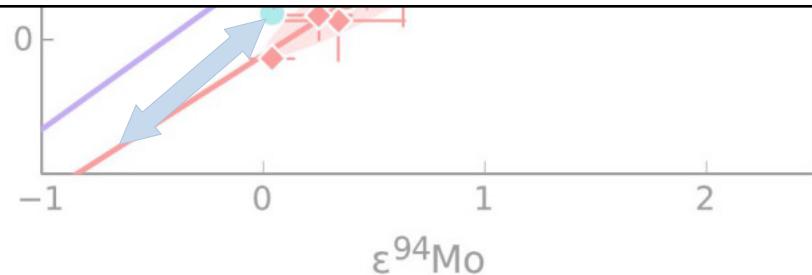
**Fundamental issue with use of nucleosynthetic
isotope anomalies / meteorites**

5M



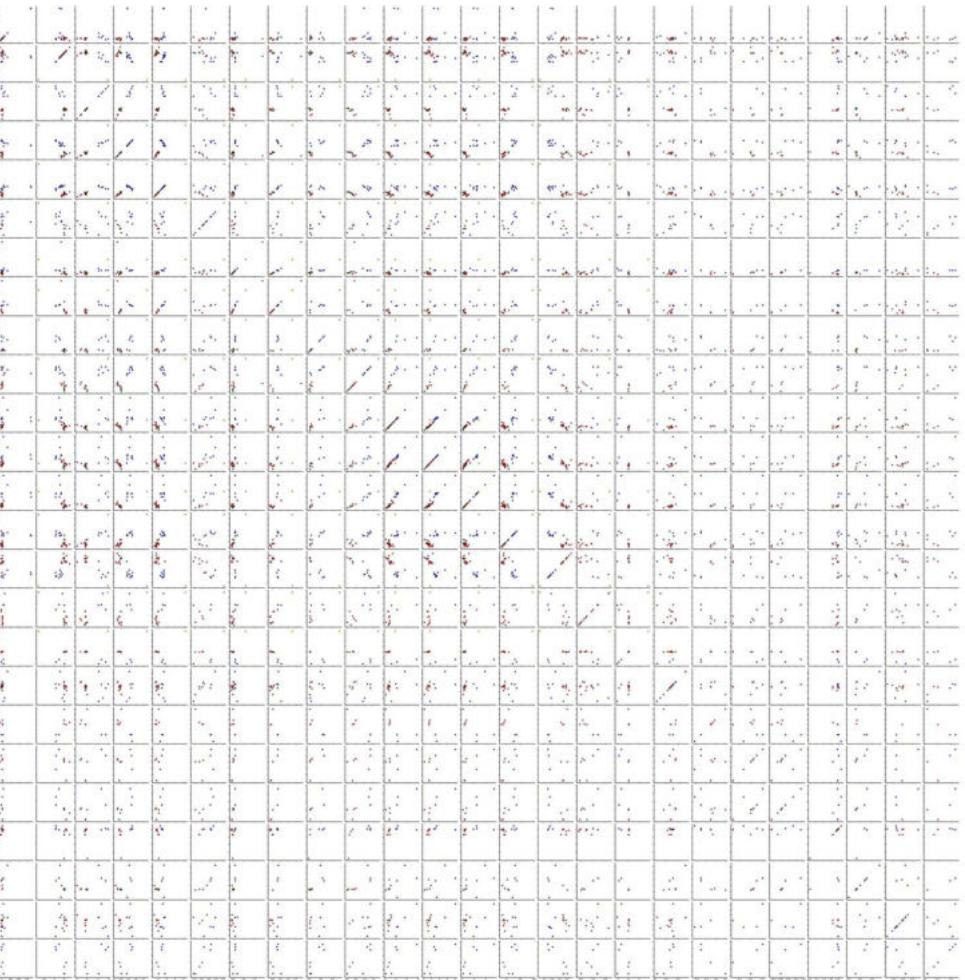
Earth contains material

Aim: sort this out

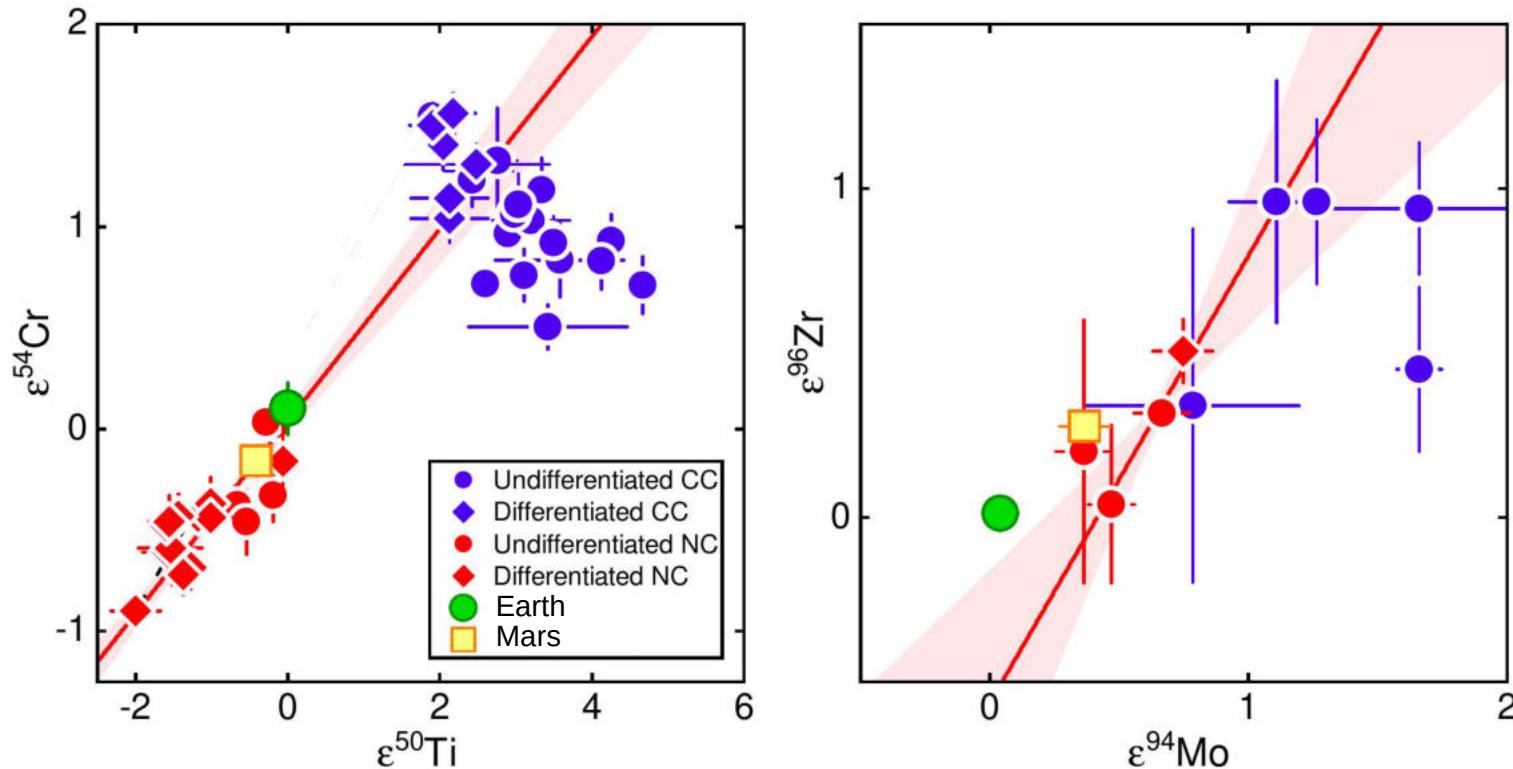


Approach

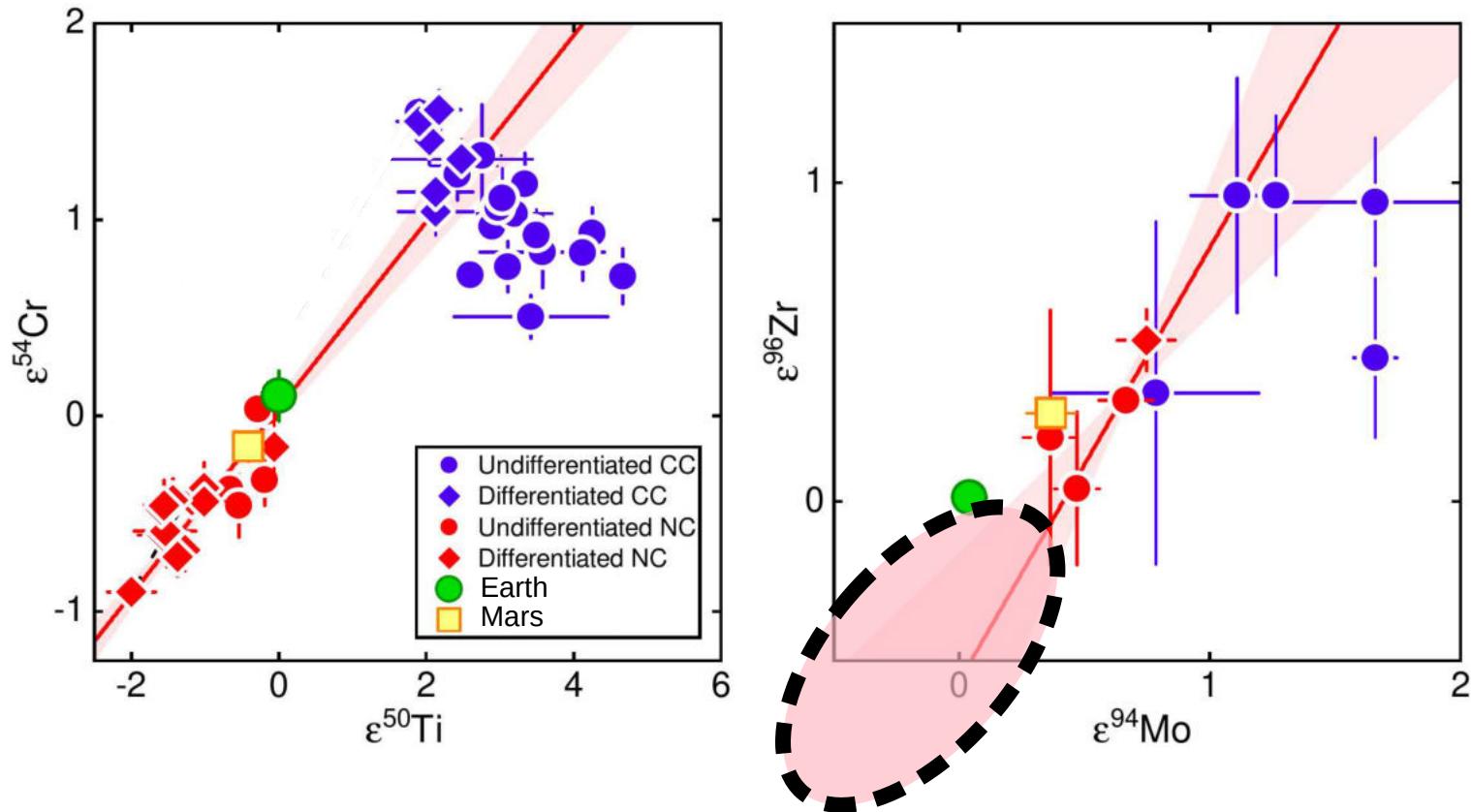
- Look at *all* anomalies in multi-elemental space
 - Constrain composition and origin of unsampled component
 - New data for Mars



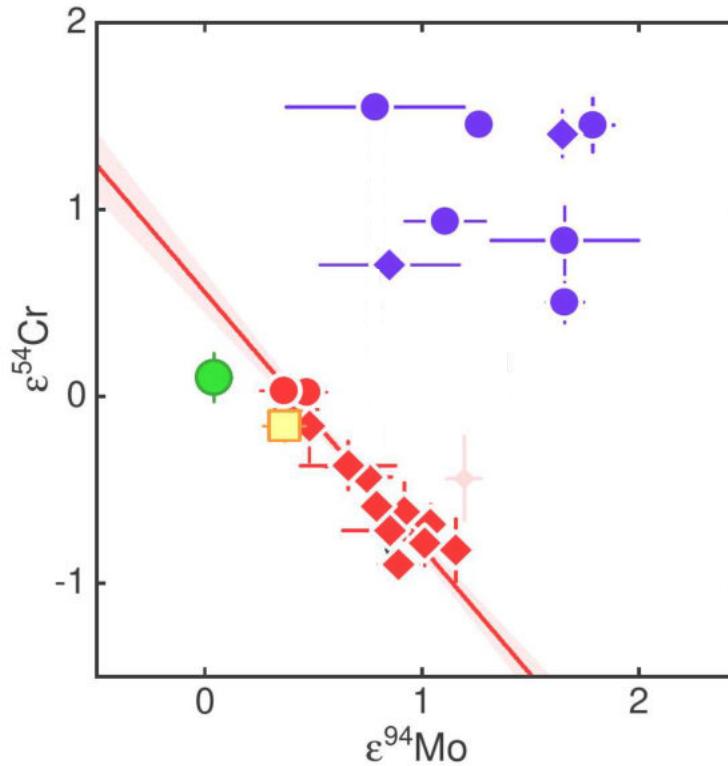
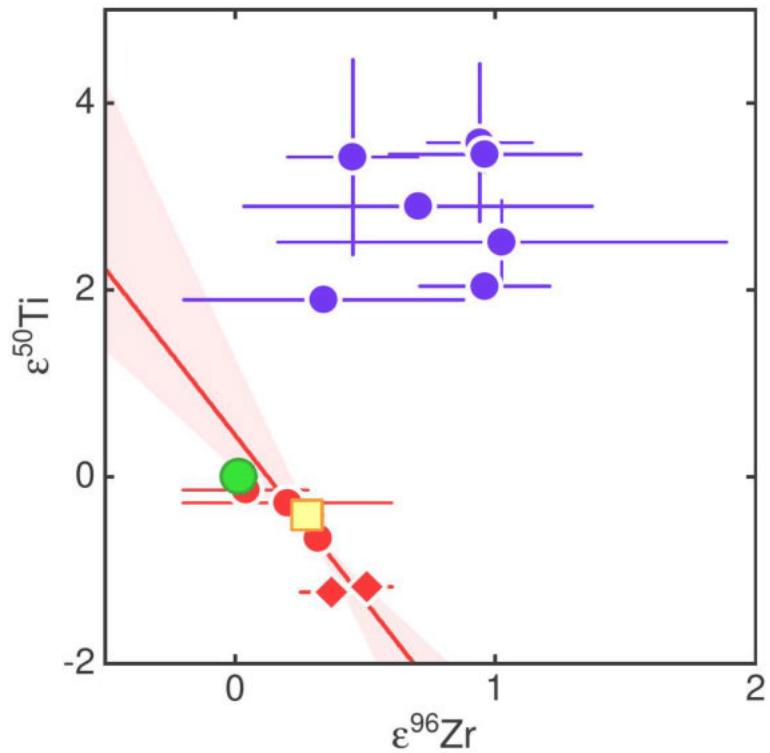
Results



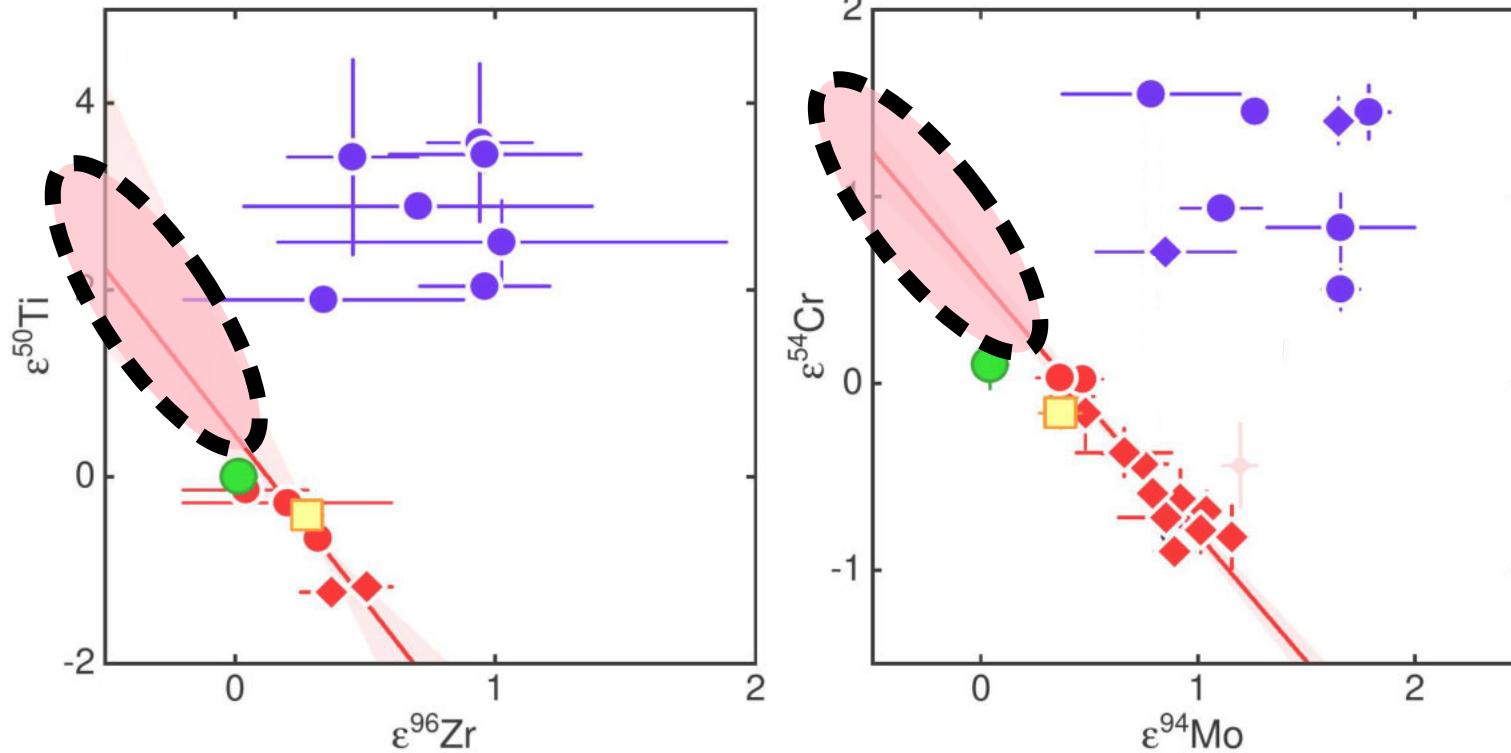
Results



Results

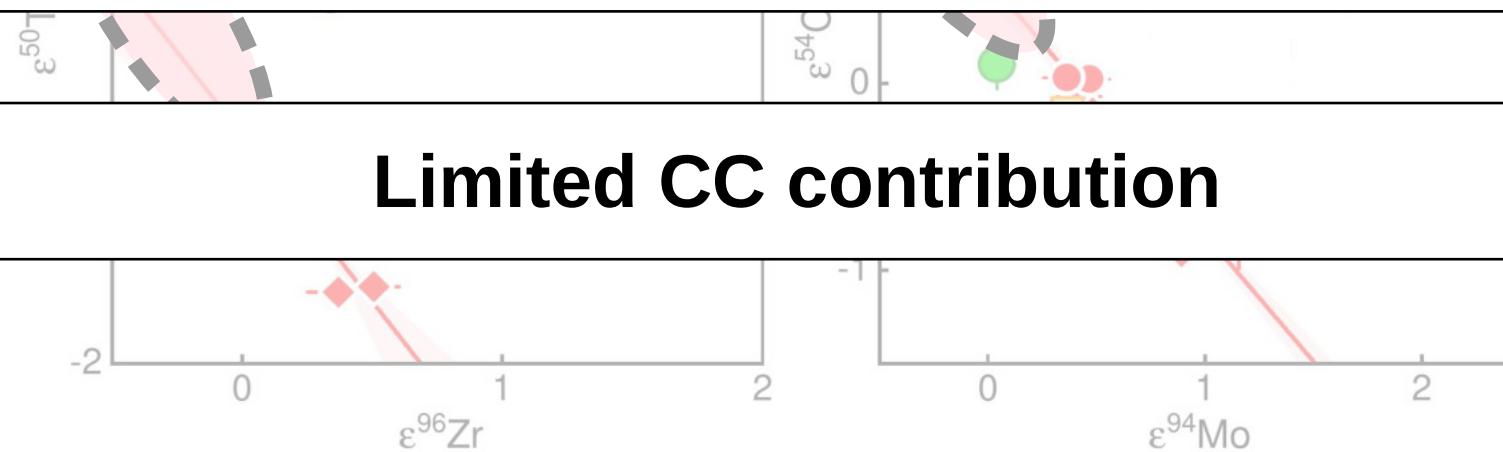


Results



Results

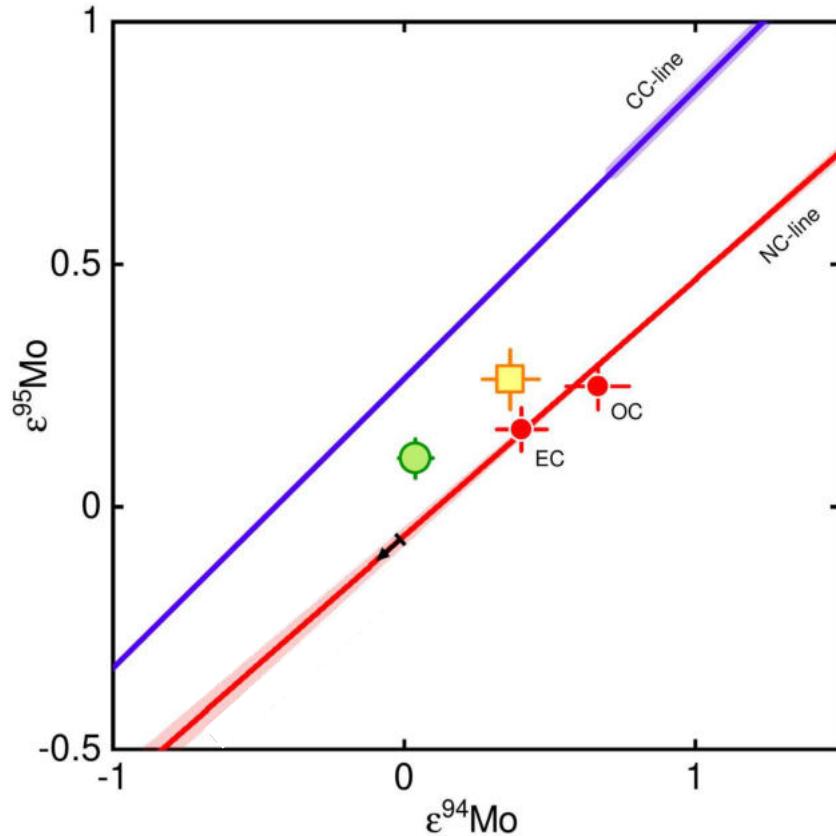
Compositions of Earth and Mars governed by two-component mixing among inner solar system materials



Limited CC contribution

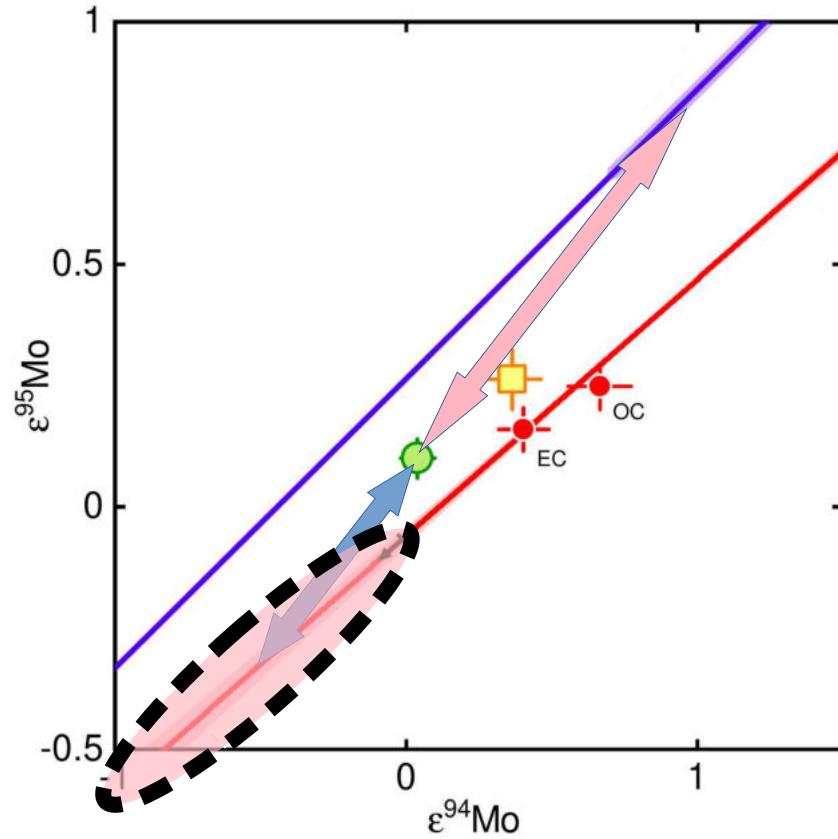
Results

Earth & Mars must contain some CC material



Results

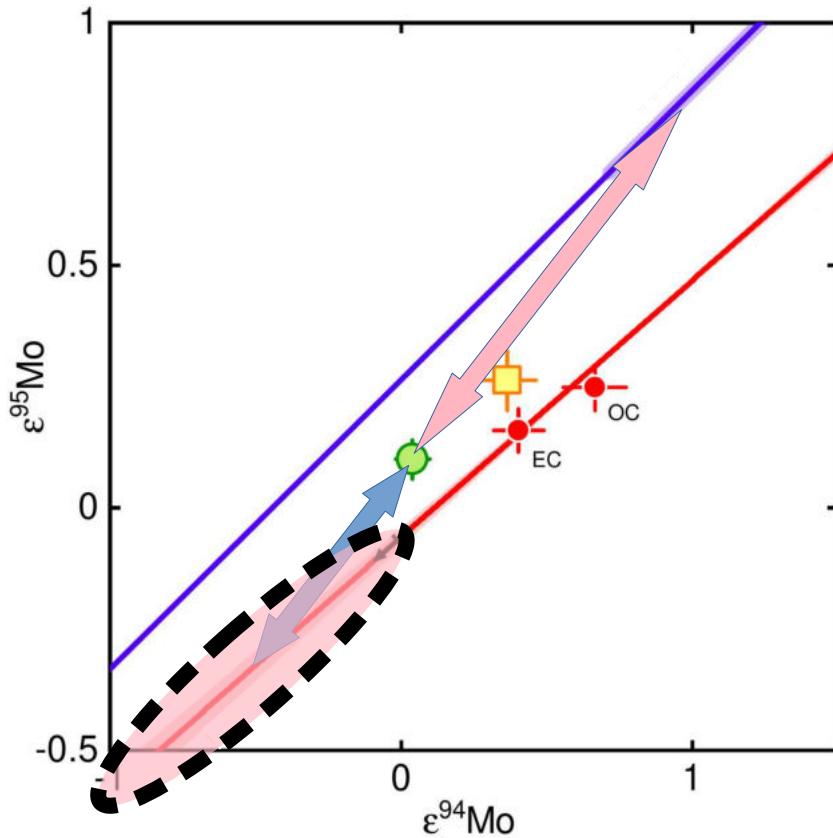
Earth & Mars must contain some CC material



Results

Earth & Mars must contain some CC material

Model:
Mo in Earth, Mars
mixture:
 $\text{NC}_{\text{Late}} + \text{NC}_{\text{Early}} + \text{CC}$

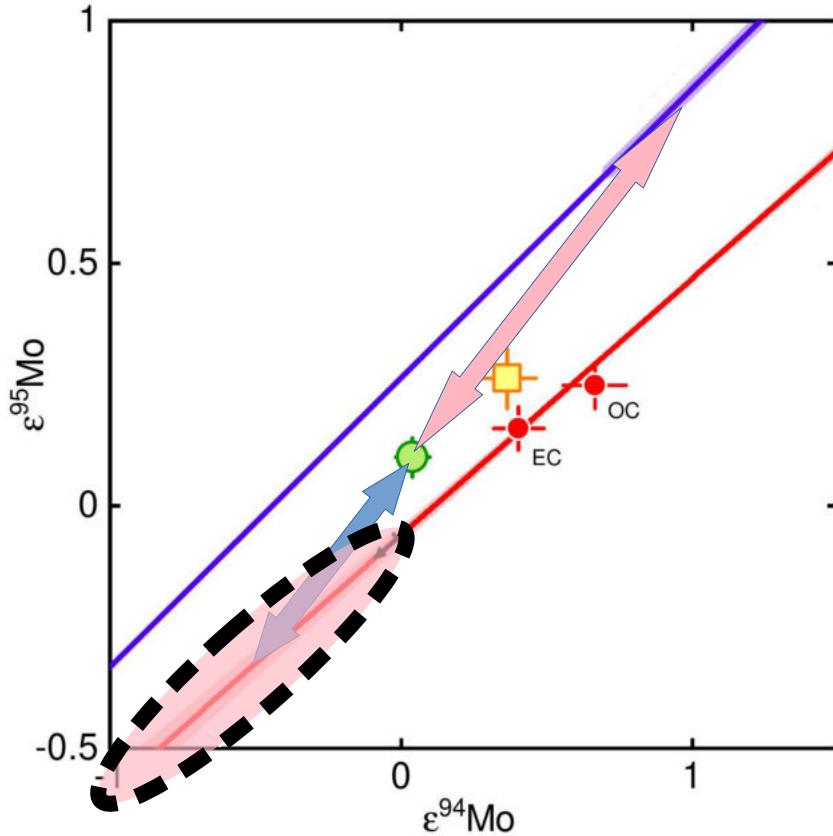


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$$\begin{aligned} \varepsilon^{94}\text{Mo}_{\text{NCLate}, f} & \quad \varepsilon^{94}\text{Mo}_{\text{NCEarly}, f} \\ \text{MoNCLate}, f & \quad \text{BulkCC}, f \\ \text{BulkNCLate} & \end{aligned}$$

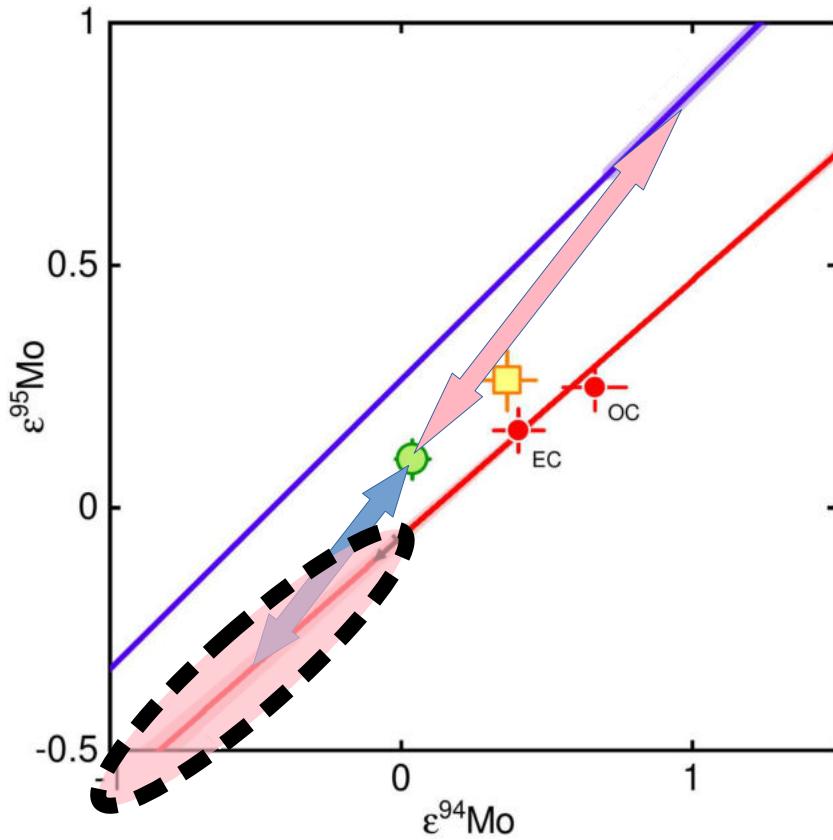


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Earth & Mars must contain some CC material

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$\varepsilon^{94}\text{Mo}_{\text{NCLate}}$, $\varepsilon^{94}\text{Mo}_{\text{NCEarly}}$
 f_{MoNCLate} , f_{BulkCC} , $f_{\text{BulkNCLate}}$

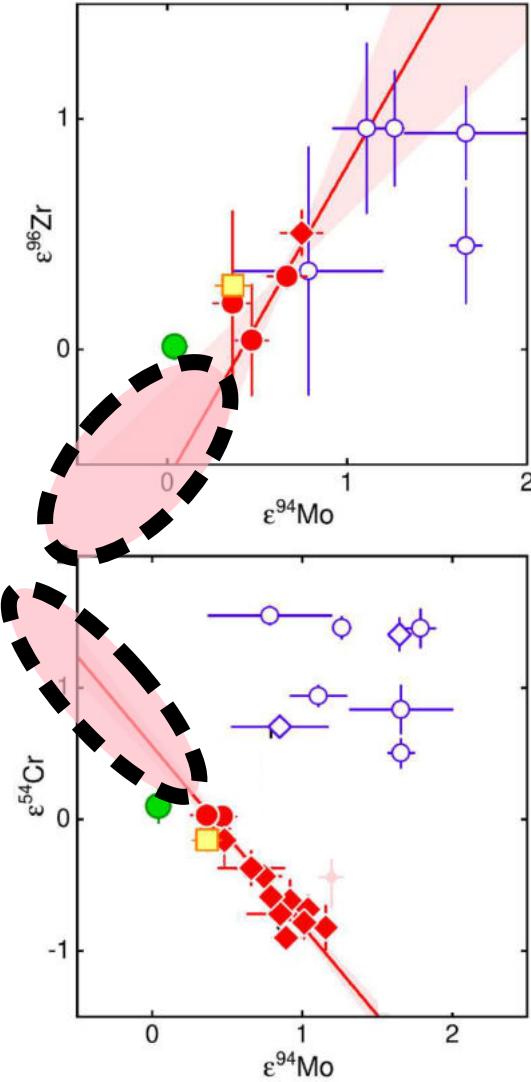


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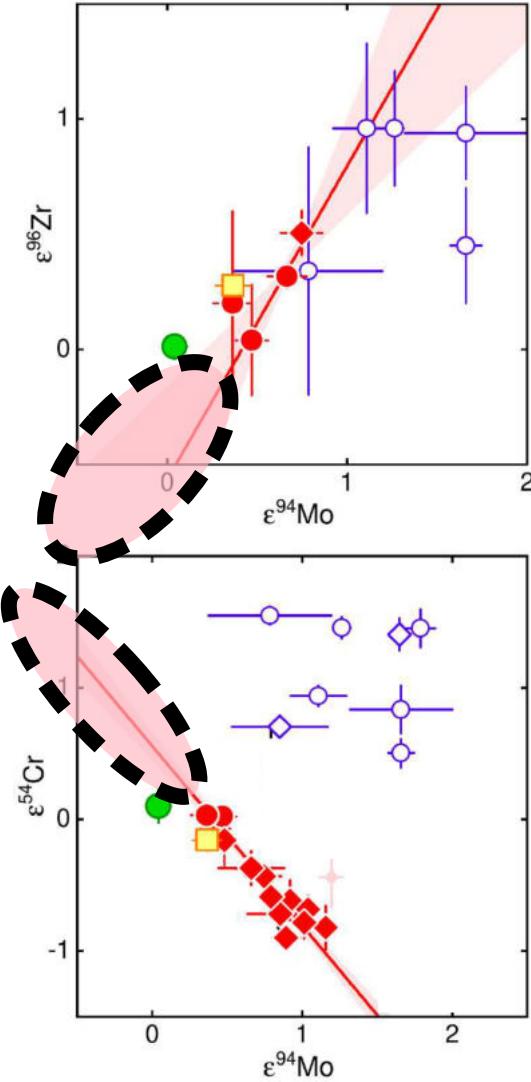


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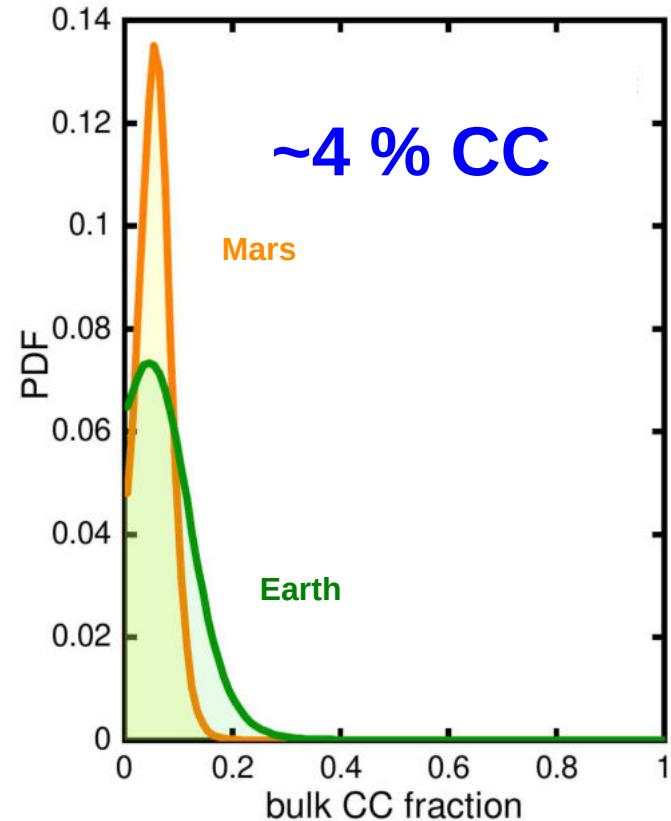
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Earth & Mars must contain some CC material

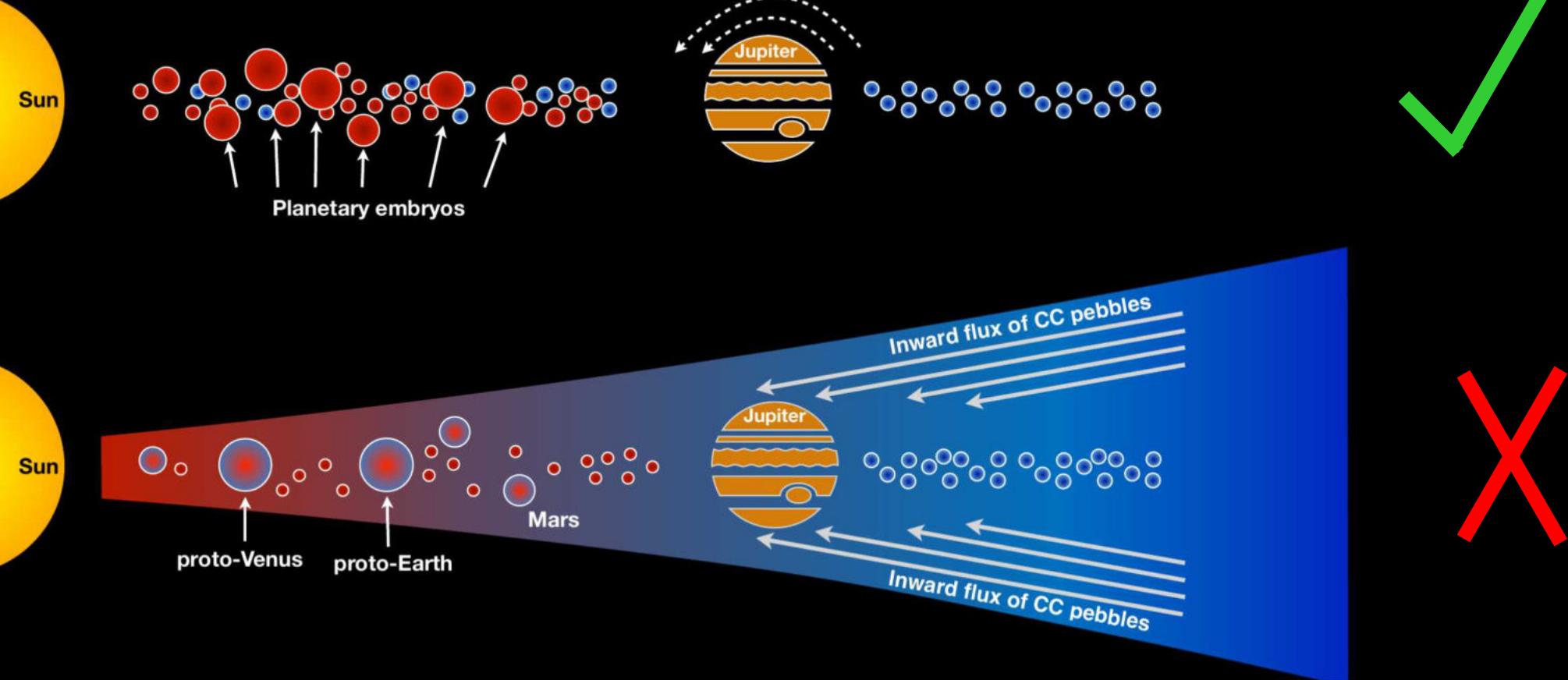
Model:
Mo in Earth, Mars
mixture:

NC_{Late} + NC_{Early} + CC

$\varepsilon^{94}\text{Mo}_{\text{NCLate}}$, $\varepsilon^{94}\text{Mo}_{\text{NCEarly}}$
 f_{MoNCLate} , f_{BulkCC} , $f_{\text{BulkNCLate}}$



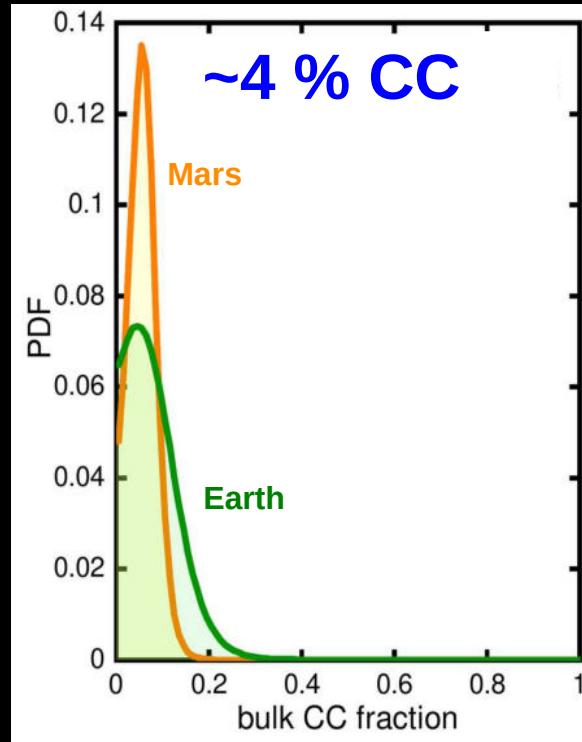
Mode of terrestrial planet formation



Heritage of Earth's volatiles



Few % CC in Earth, Mars



Heritage of Earth's volatiles



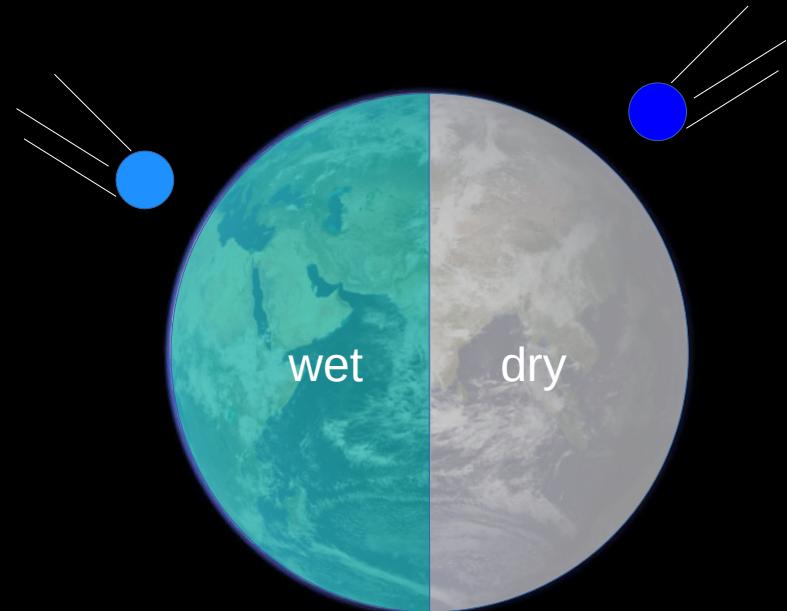
Few % CC in Earth, Mars
Valid for refractory elements

Heritage of Earth's volatiles



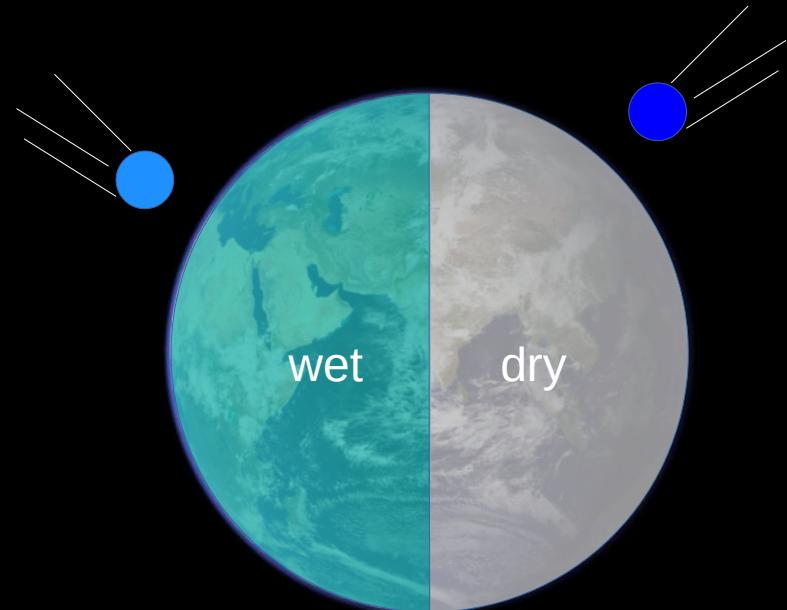
Few % CC in Earth, Mars
Valid for refractory elements
Suffice to explain all of Earth's & Mars' volatiles by CC addition

Heritage of Earth's volatiles



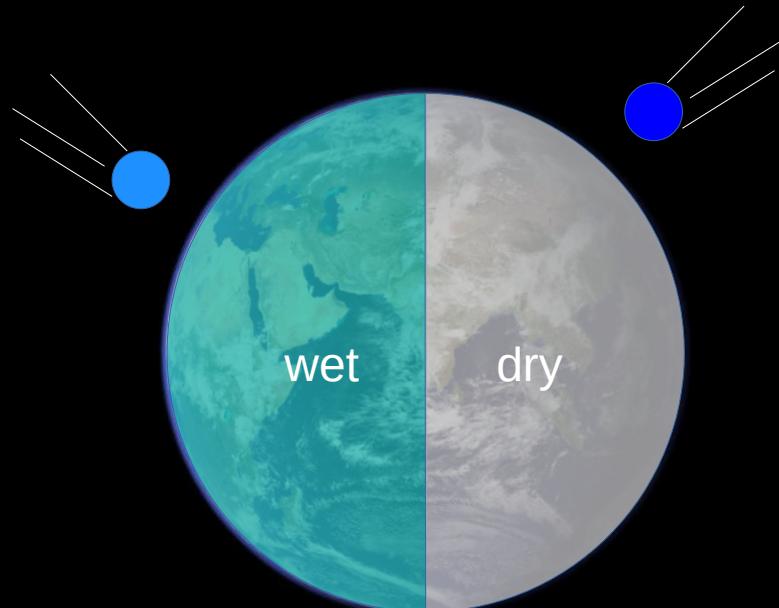
Few % CC in Earth, Mars
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Can't differentiate between **wet** & dry accretion ($f_{CC\ volatiles}$)

Heritage of Earth's volatiles

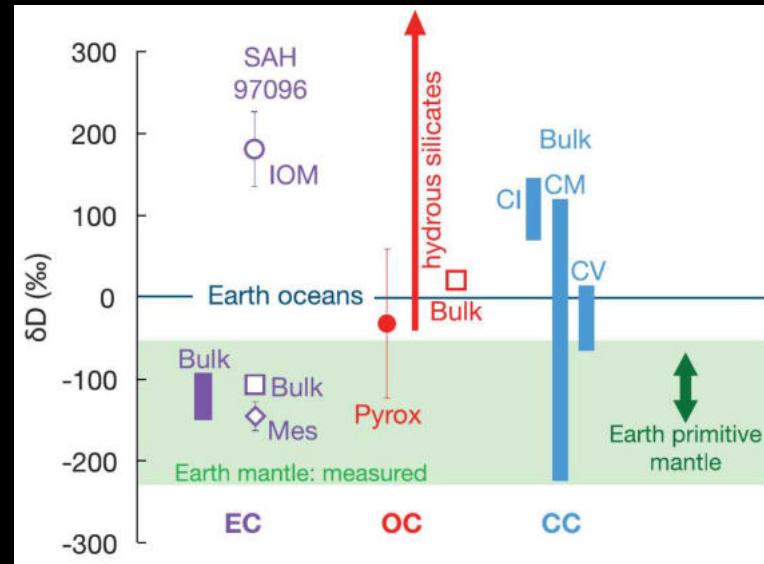


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→ Data from volatile elements

Heritage of Earth's volatiles

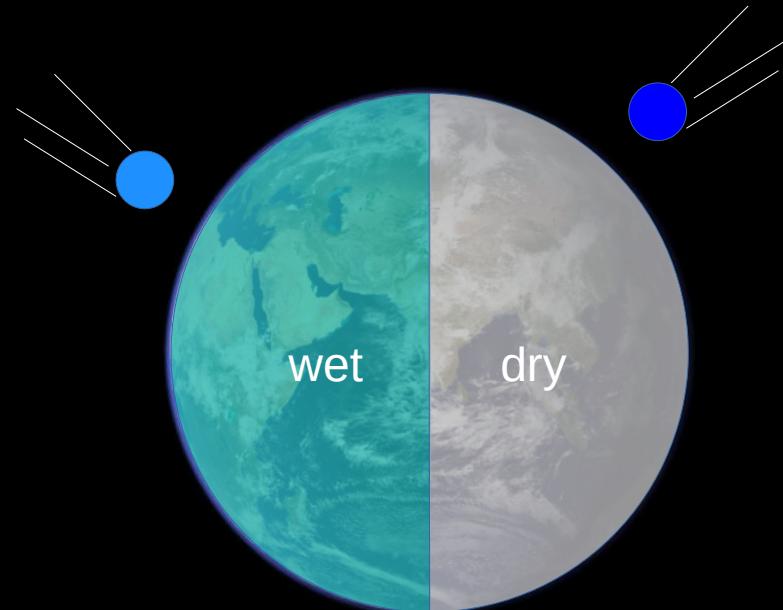


H, C, N only 2 isotopes!

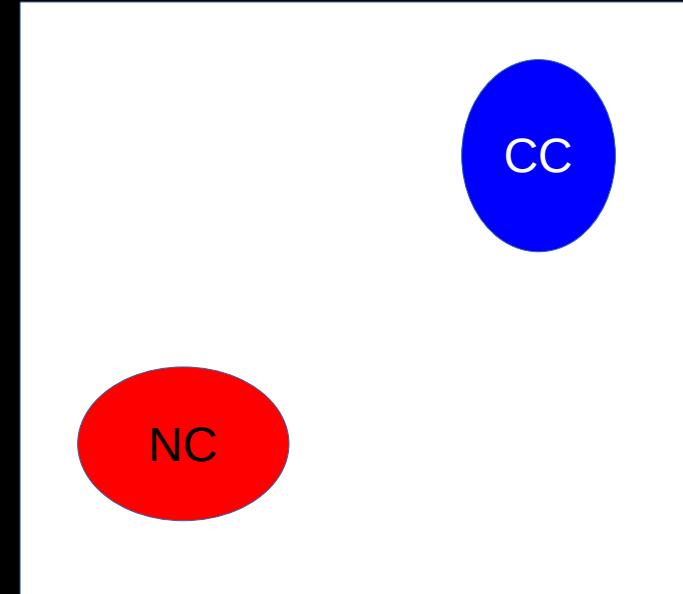


→ Data from volatile elements

Heritage of Earth's volatiles

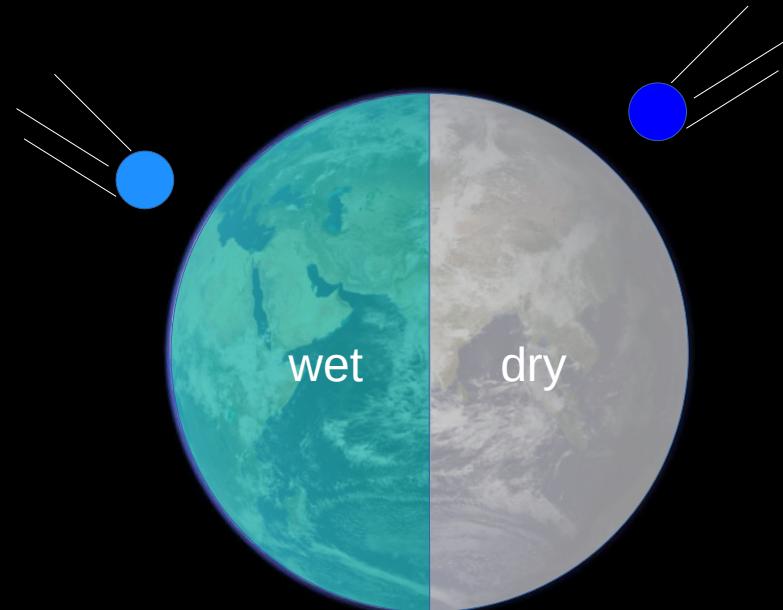


X 4+ isotopes, NC CC distinct

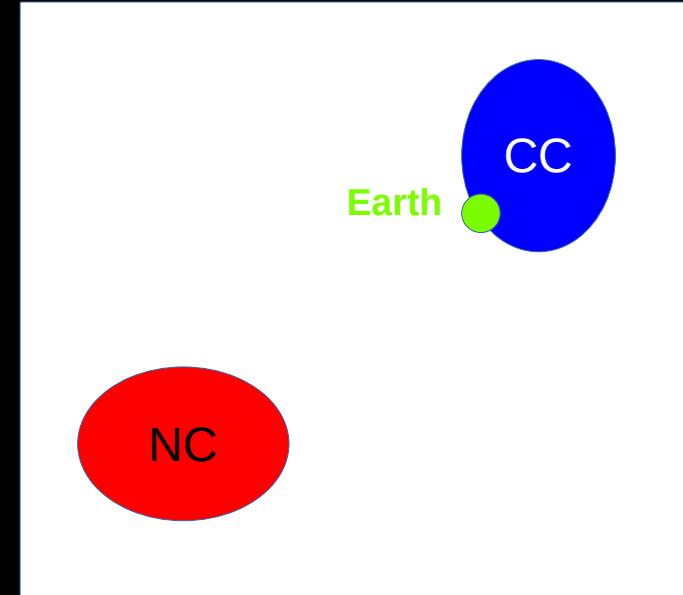


→ Data from volatile elements

Heritage of Earth's volatiles

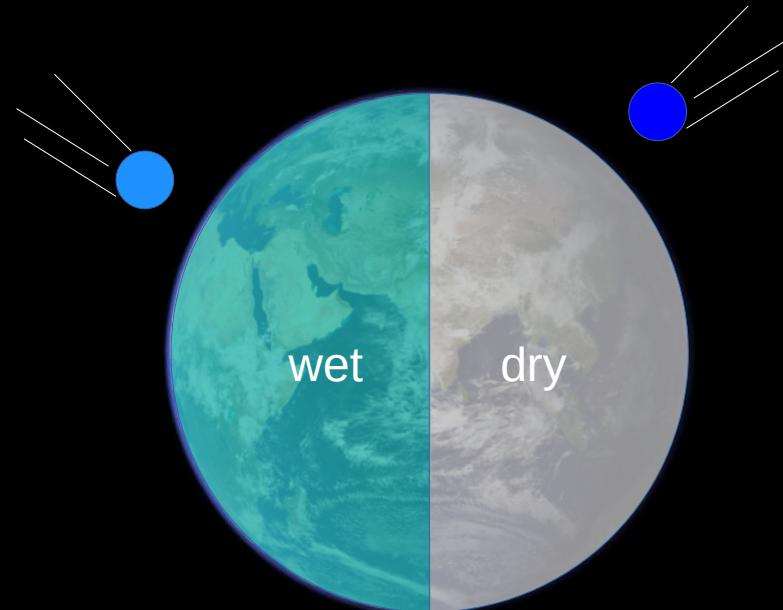


X 4+ isotopes, NC CC distinct

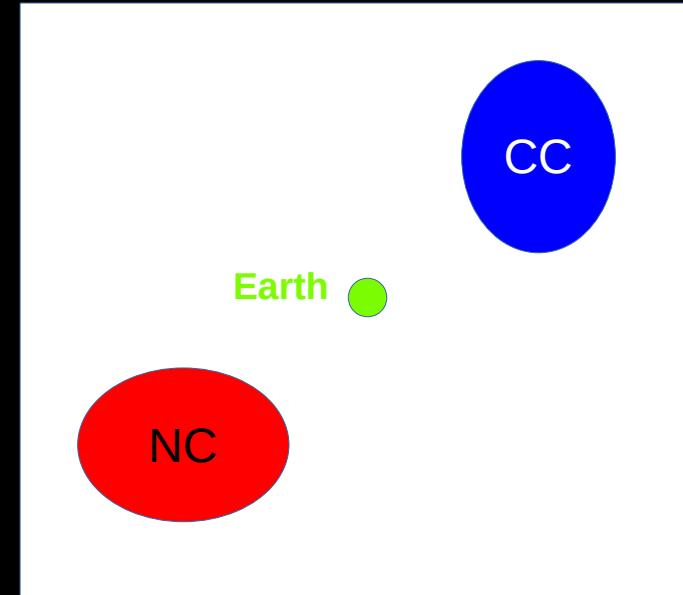


→ Data from volatile elements

Heritage of Earth's volatiles

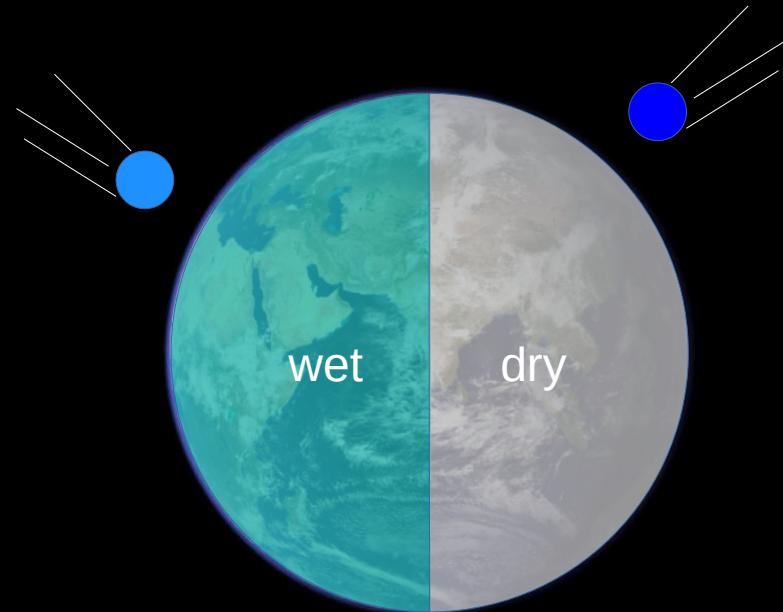


X 4+ isotopes, NC CC distinct

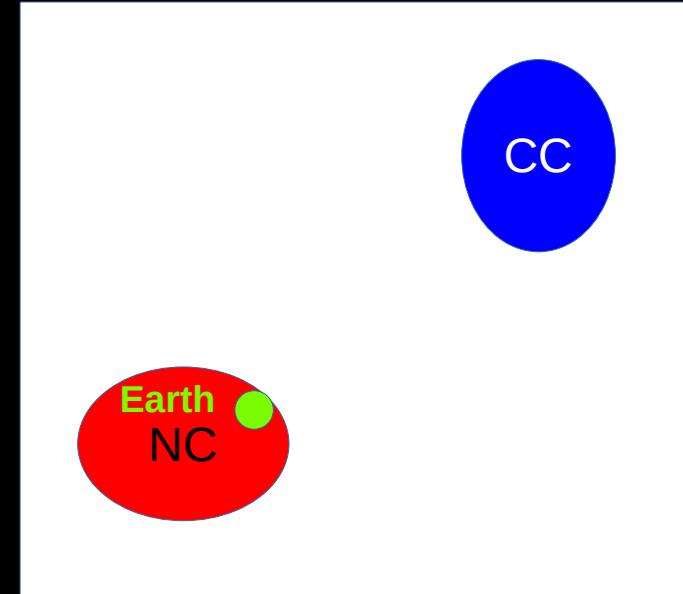


→ Data from volatile elements

Heritage of Earth's volatiles



X 4+ isotopes, NC CC distinct



→ Data from volatile elements

Zinc

$$T_{C,50\%} = 704 \text{ K}$$

lithophile

abundant in meteorites

Fe-peak element

A standard periodic table of elements is shown, with Zinc (Zn) circled in red. Zinc is located in the fourth period, group 12, with an atomic number of 30. It has a blue square and is labeled 'Zinc' with the symbol 'Zn' and the atomic mass '65.39'. The table includes element names, atomic numbers, and atomic masses for all elements from Hydrogen (H) to Ununseptium (Uus).

1	H	Hydrogen 1.0079
3	Li	Lithium 6.941
4	Be	Beryllium 9.01218
11	Na	Sodium 22.989769
12	Mg	Magnesium 24.302
19	K	Potassium 39.0983
20	Ca	Calcium 40.078
21	Sc	Scandium 44.95991
22	Ti	Titanium 47.86
23	V	Vanadium 50.9415
24	Cr	Chromium 51.981
25	Mn	Manganese 54.938
26	Fe	Iron 55.847
27	Co	Cobalt 58.9332
28	Ni	Nickel 58.6994
29	Cu	Copper 63.548
30	Zn	Zinc 65.39
31	Ga	Gallium 69.732
32	In	Inium 71.760
33	Ge	Germanium 72.64
34	As	Arsenic 74.92159
35	Se	Selenium 78.90
36	Br	Bromine 79.904
37	Rb	Rubidium 85.4678
38	Sr	Samarium 87.62
39	Y	Yttrium 88.90585
40	Zr	Zirconium 91.224
41	Nb	Niobium 92.90638
42	Mo	Molybdenum 95.94
43	Tc	Technetium 98.9072
44	Ru	Ruthenium 101.07
45	Rh	Rhodium 102.9055
46	Pd	Palladium 106.42
47	Ag	Silver 107.8682
48	Cd	Cadmium 112.411
49	In	Inium 114.818
50	Sn	Tin 118.71
51	Sb	Antimony 121.760
52	Te	Tellurium 127.6
53	I	Iodine 126.90447
54	Xe	Krypton 131.39
55	Cs	Cesium 132.90543
56	Ba	Boron 137.327
57	La	La 138.90654
72	Hf	Hafnium 178.49
73	Ta	Tantalum 183.8479
74	W	Tungsten 183.85
75	Re	Rhenium 186.207
76	Os	Osmium 190.23
77	Ir	Iridium 192.22
78	Pt	Platinum 195.08
79	Au	Auro 196.9665
80	Hg	Mercury 200.59
81	Tl	Thallium 204.3833
82	Pb	Lead 207.2
83	Bi	Bismuth 208.98637
84	Po	Protactinium 208.9824
85	At	Astatine 209.9871
86	Rn	Radon 222.0176
87	Fr	Francium 223.0161
88	Ra	Radium 226.0204
89-103	Rf	Rutherfordium [98]
104	Db	Dubnium [106]
105	Sg	Seskbergium [106]
106	Bh	Baltsjuk [106]
107	Hs	Hassium [108]
108	Mt	Mendelevium [108]
109	Ds	Darmstadtium [108]
110	Rg	Roentgenium [107]
111	Cn	Copernicium [111]
112	Uut	Ununtrium Unknown
113	Fl	Florium Unknown
114	Uup	Ununpentium Unknown
115	Lv	Livermorium Unknown
116	Uuo	Ununseptium Unknown

Zinc

$T_{C,50\%} = 704 \text{ K}$

lithophile

abundant in meteorites

Fe-peak element

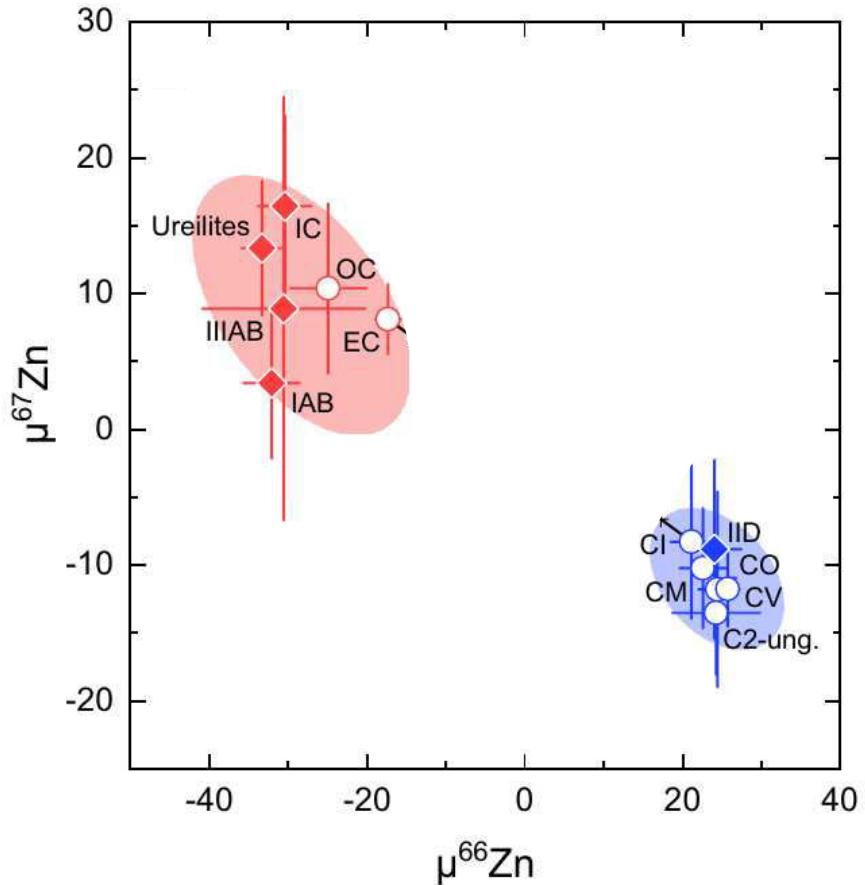
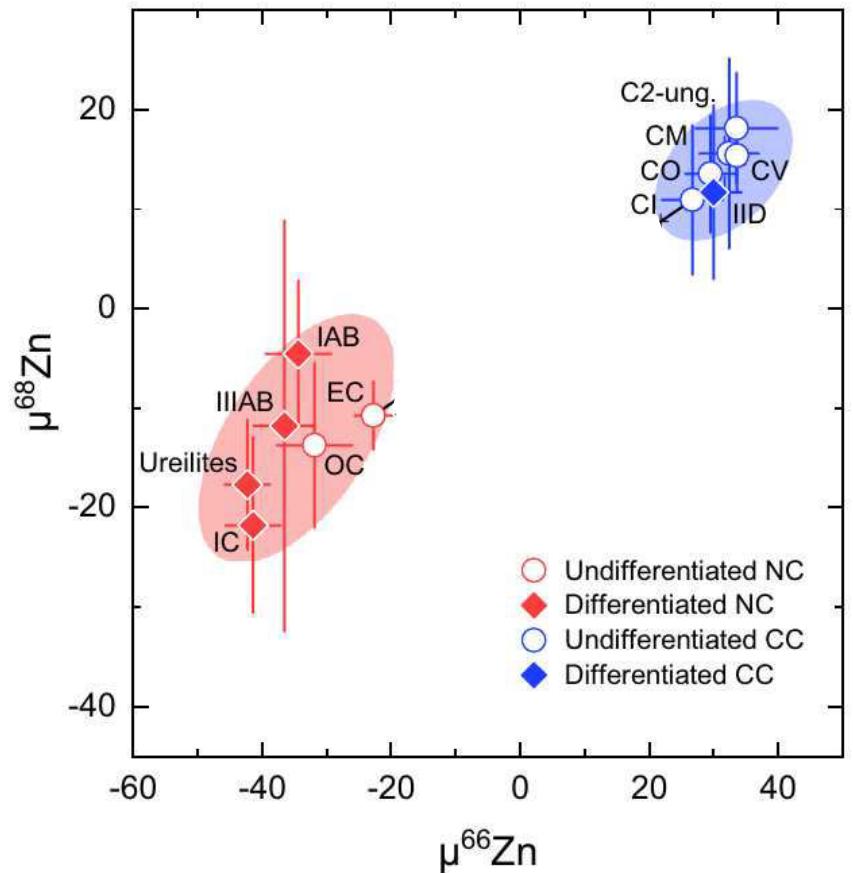
5 isotopes

^{64}Zn
48.6

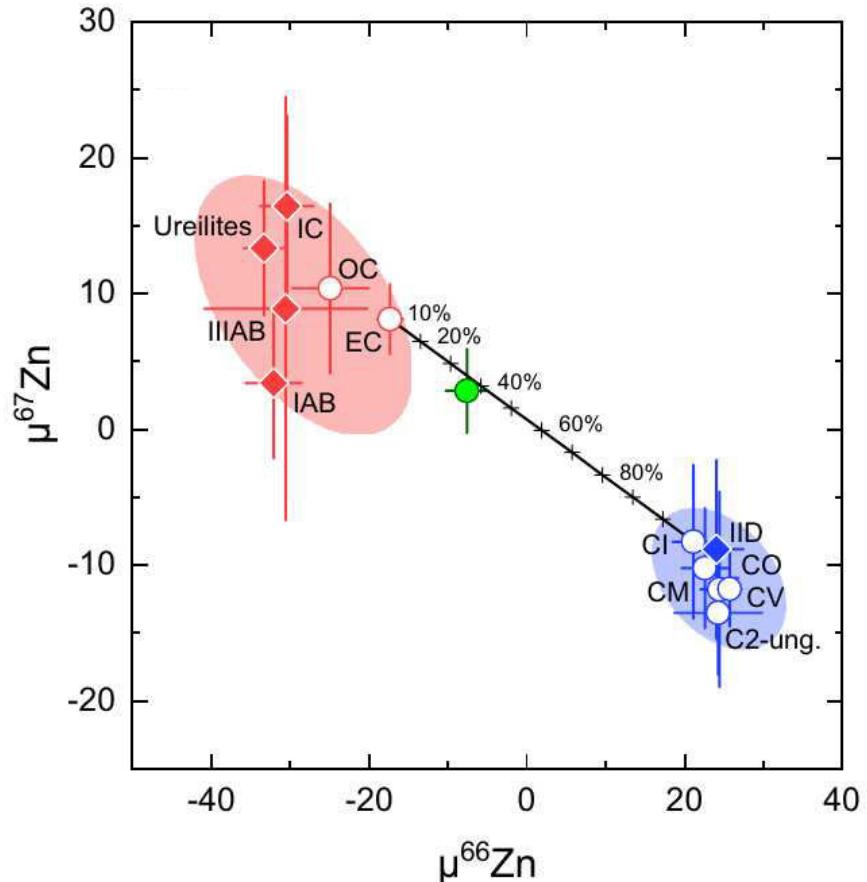
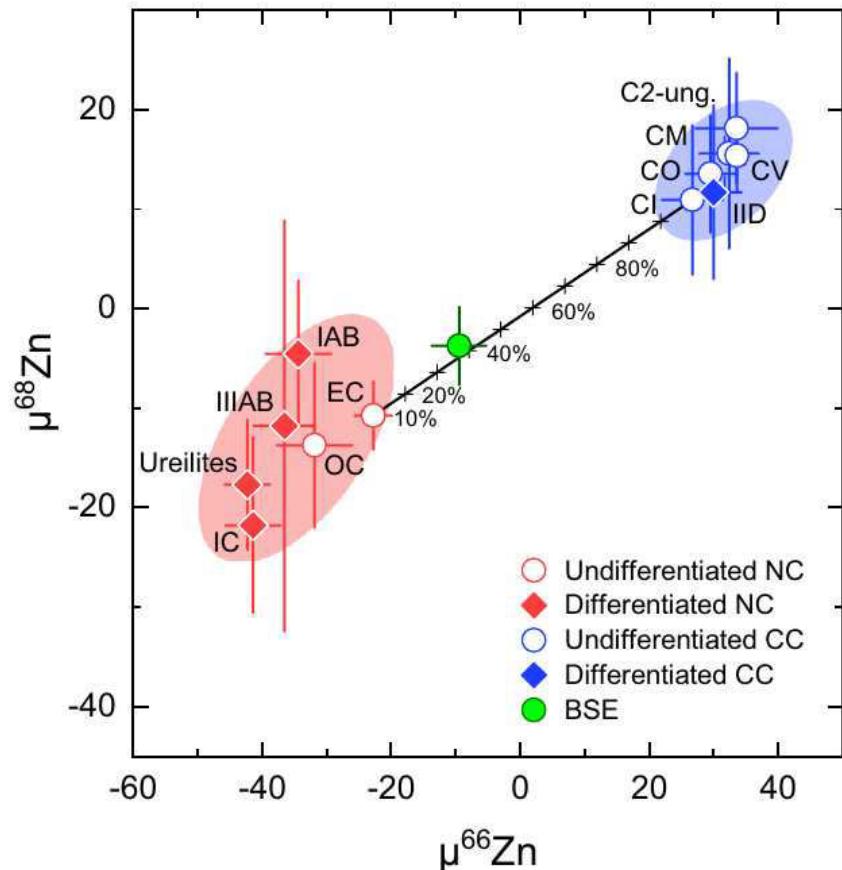
^{66}Zn ^{67}Zn ^{68}Zn
27.9 4.1 4.1

^{70}Zn
0.6

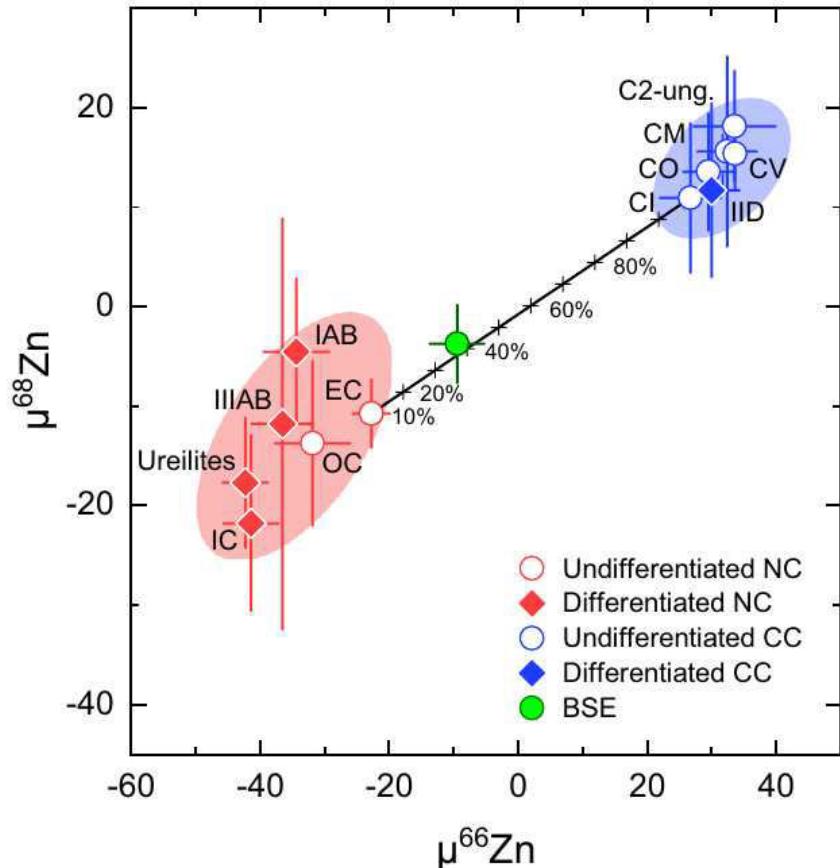
Results



Results



Results

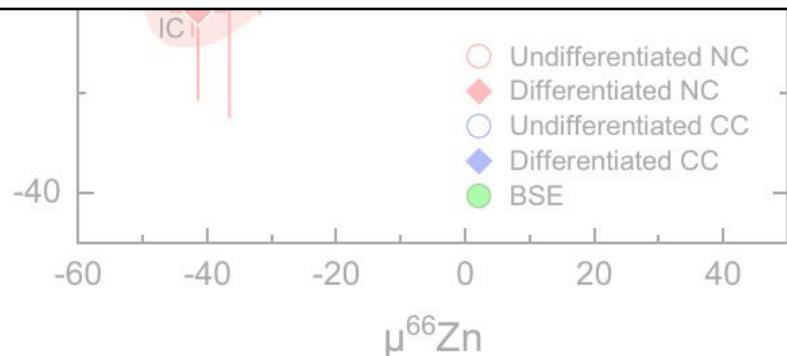


$$f_{CC}(\text{Zn})_{BSE} = \frac{\mu^{66}\text{Zn}_{BSE} - \mu^{66}\text{Zn}_{NC}}{\mu^{66}\text{Zn}_{CC} - \mu^{66}\text{Zn}_{NC}}$$

Results



~70% of Zn in Earth of NC heritage, 30% CC

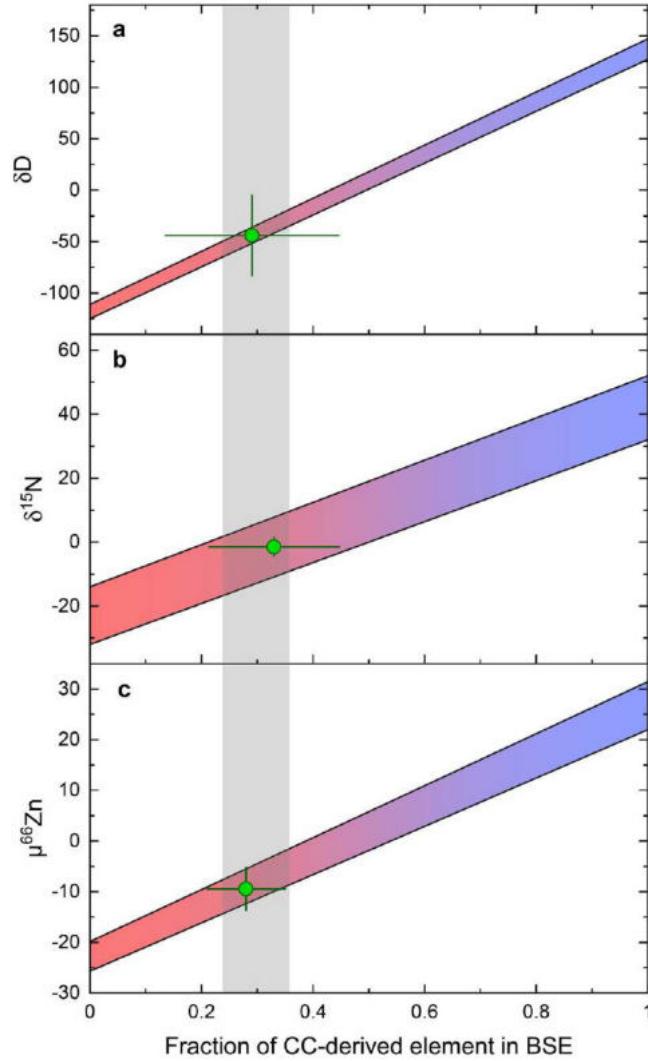


Results

$$f_{CC}(H)_{BSE} = \frac{\delta D_{BSE} - \delta D_{NC}}{\delta D_{CC} - \delta D_{NC}}$$

$$f_{CC}(N)_{BSE} = \frac{\delta^{15}N_{BSE} - \delta^{15}N_{NC}}{\delta^{15}N_{CC} - \delta^{15}N_{NC}}$$

$$f_{CC}(Zn)_{BSE} = \frac{\mu^{66}\text{Zn}_{BSE} - \mu^{66}\text{Zn}_{NC}}{\mu^{66}\text{Zn}_{CC} - \mu^{66}\text{Zn}_{NC}}$$

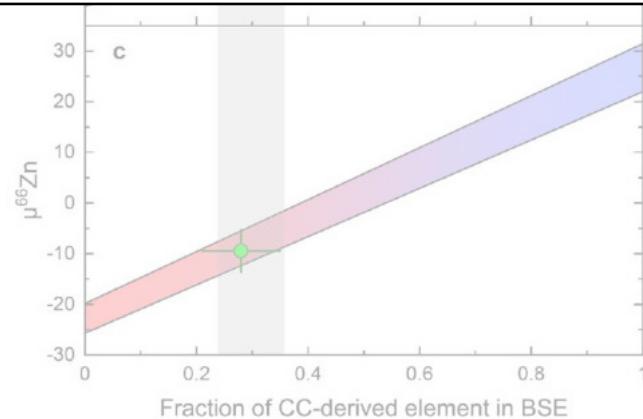
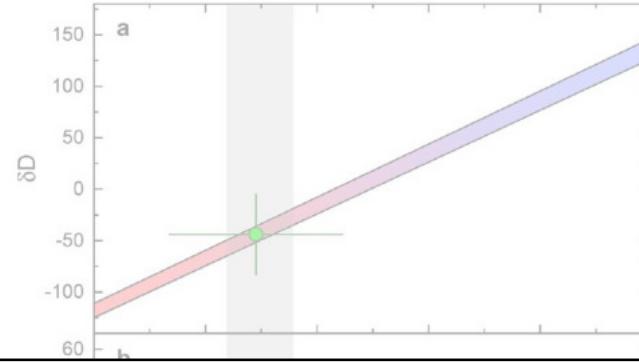


Results

$$f_{CC}(H)_{BSE} = \frac{\delta D_{BSE} - \delta D_{NC}}{\delta D_{CC} - \delta D_{NC}}$$

f_{cc} volatiles_{BSE} ~ 30%

$$f_{CC}(Zn)_{BSE} = \frac{\mu^{66}\text{Zn}_{BSE} - \mu^{66}\text{Zn}_{NC}}{\mu^{66}\text{Zn}_{CC} - \mu^{66}\text{Zn}_{NC}}$$



Fraction of **CC** in bulk Earth?

$$f_{CC}(Earth) = f_{CC}(Zn)_{BSE} \cdot \frac{[Zn]_{BSE}}{[Zn]_{CC}}$$

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$$f_{CC}(Earth) = f_{CC}(Zn)_{BSE} \cdot \frac{[Zn]_{BSE}}{[Zn]_{CC}}$$

CI chondrites: $5 \pm 2\%$

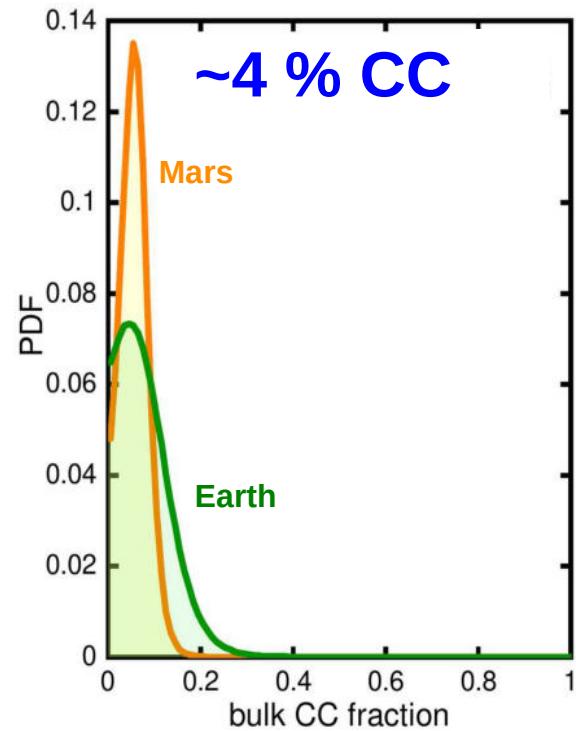
CV chondrites: $12 \pm 4\%$

Fraction of CC in bulk Earth?

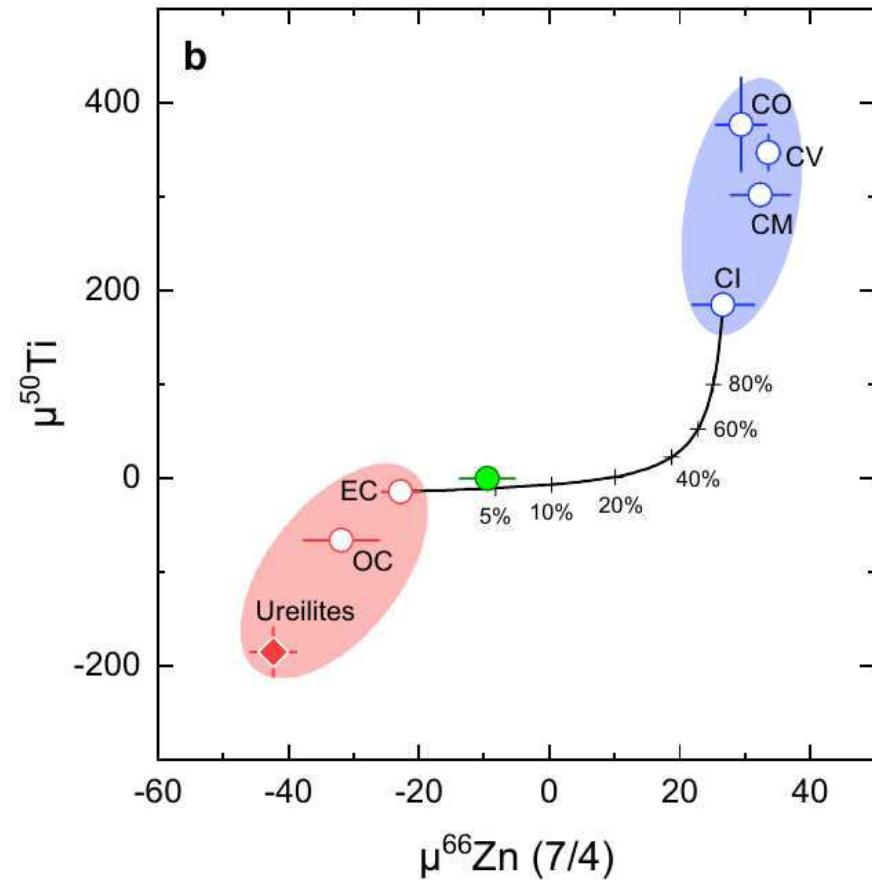
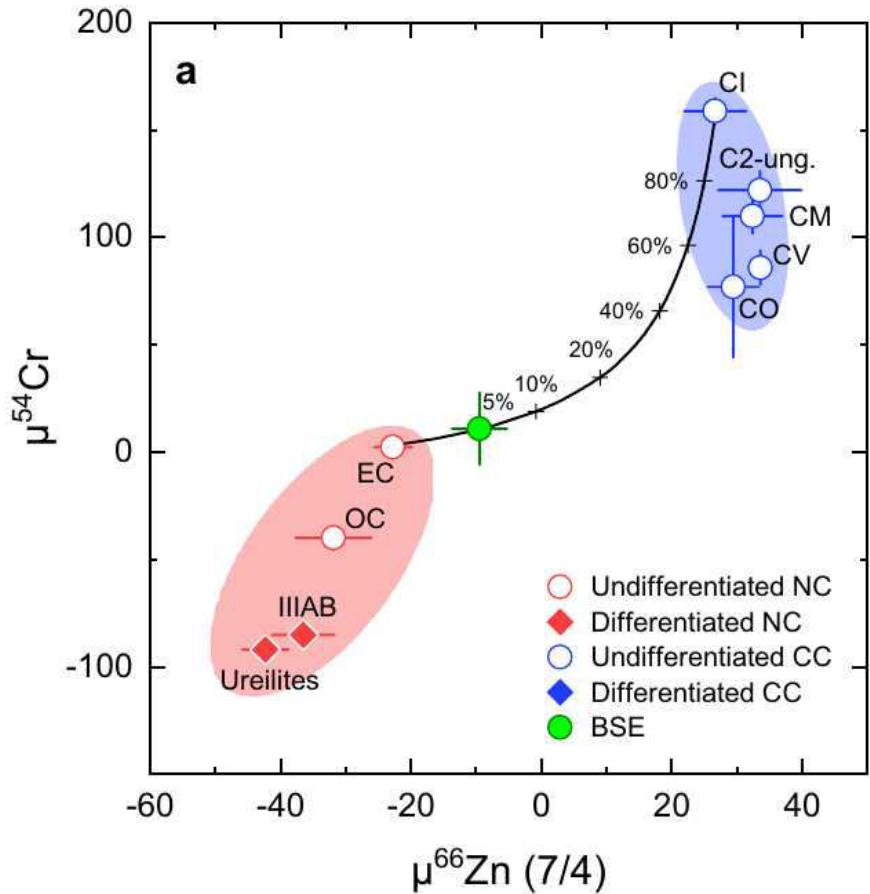
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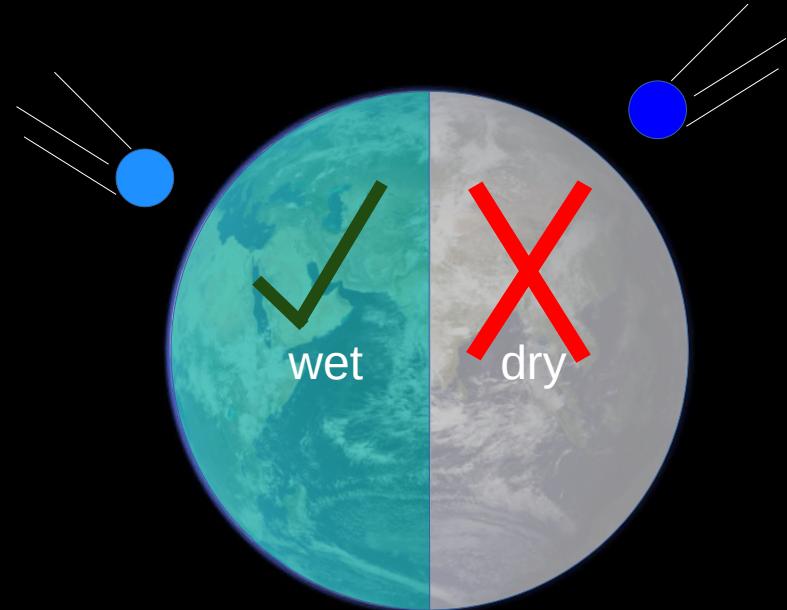
CV chondrites: $12 \pm 4\%$



Fraction of CC in bulk Earth



Heritage of volatiles



Identification of **NC CC** isotopic dichotomy for volatile element Zn

Heritage of BSE volatiles **70:30**

Volatile present in inner solar system
→ Wet accretion!

Isotopic anomalies in meteorites

What is Solar System made of?

How did disk form/disk dynamics?

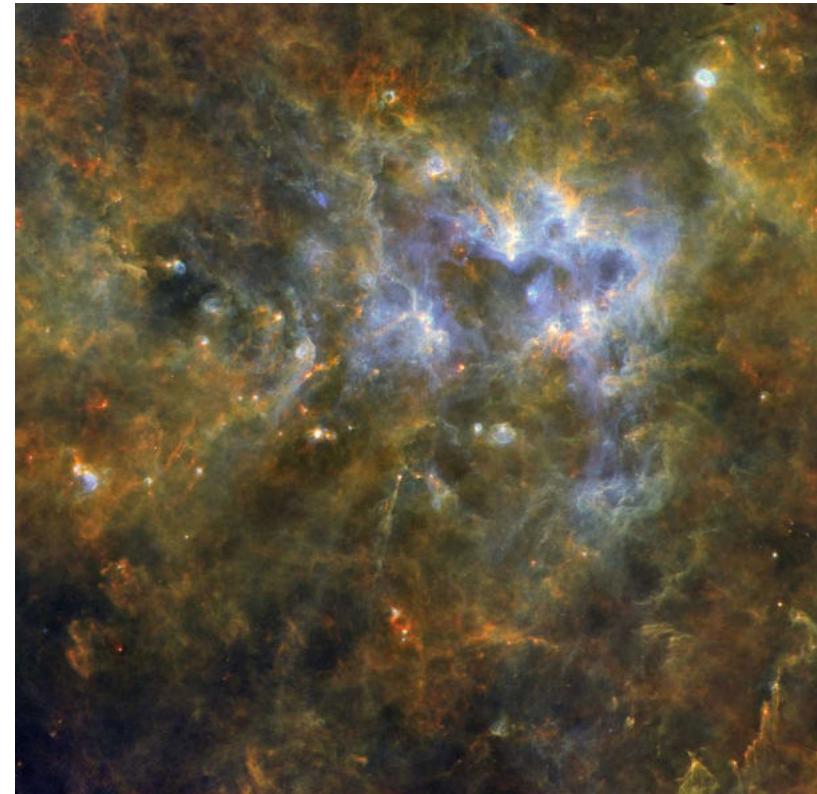
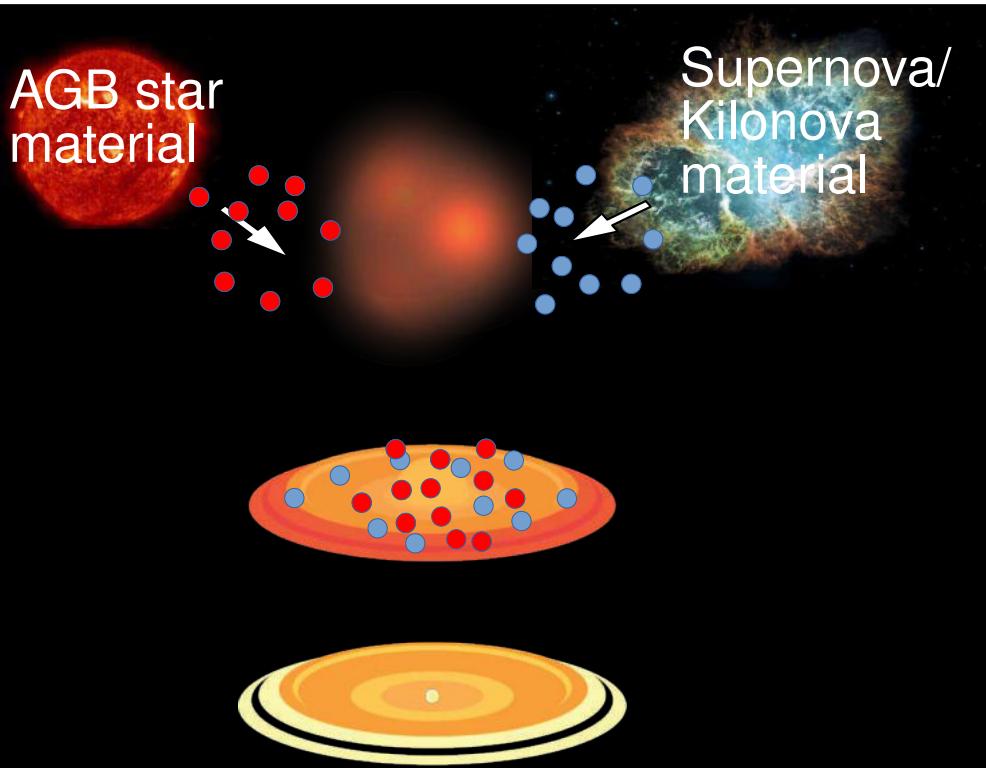
What is Earth made of?

How did Earth form?

Heritage of Earth volatiles?

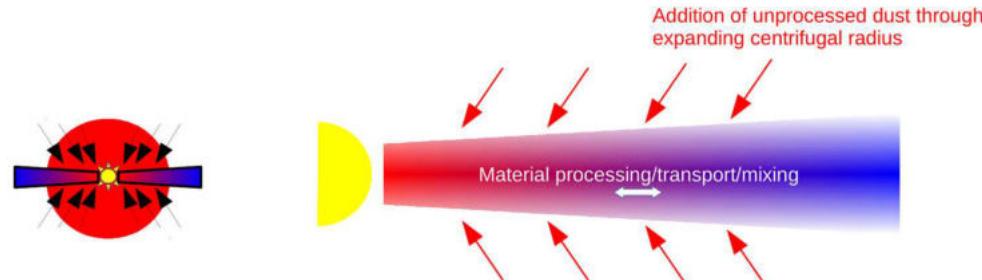


Isotopic anomalies in meteorites



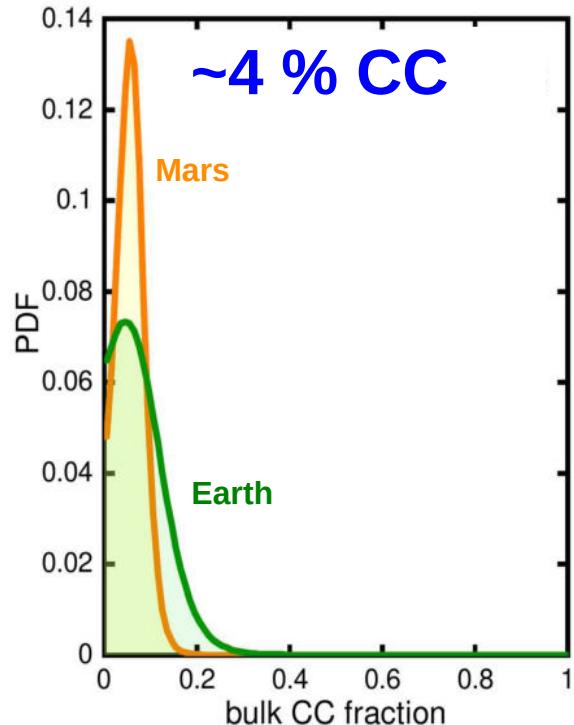
What is the solar system made of?

Isotopic anomalies in meteorites



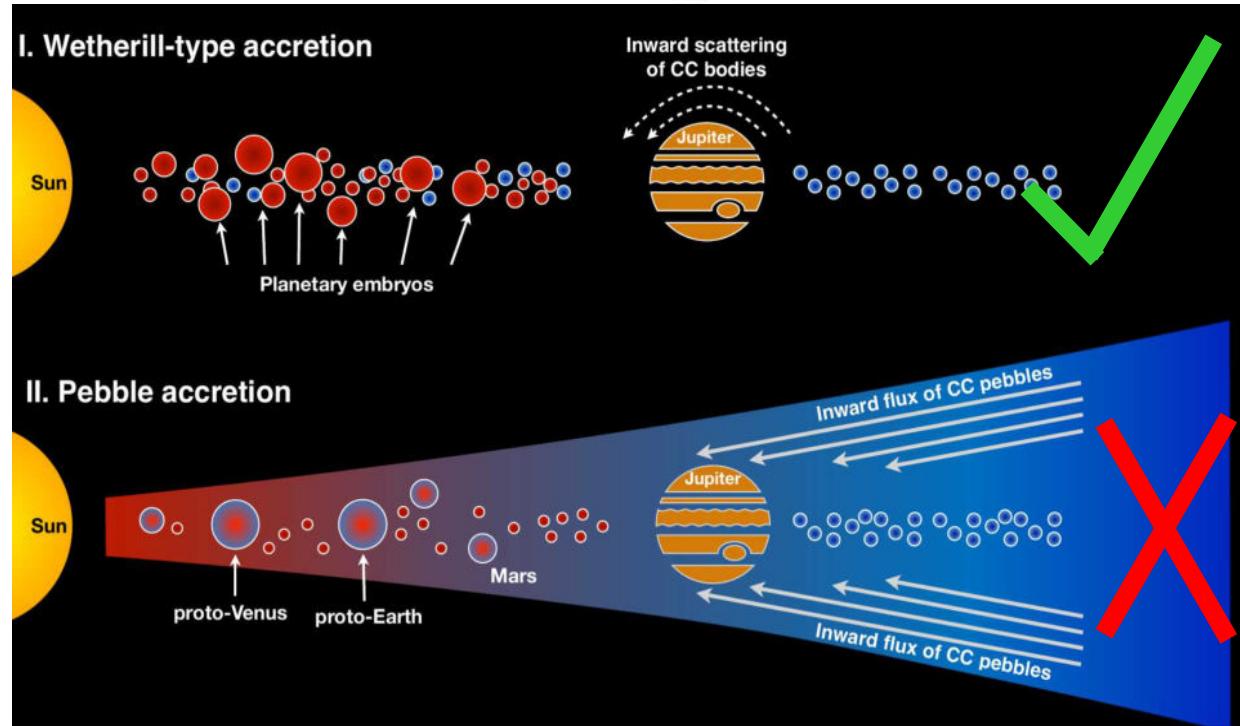
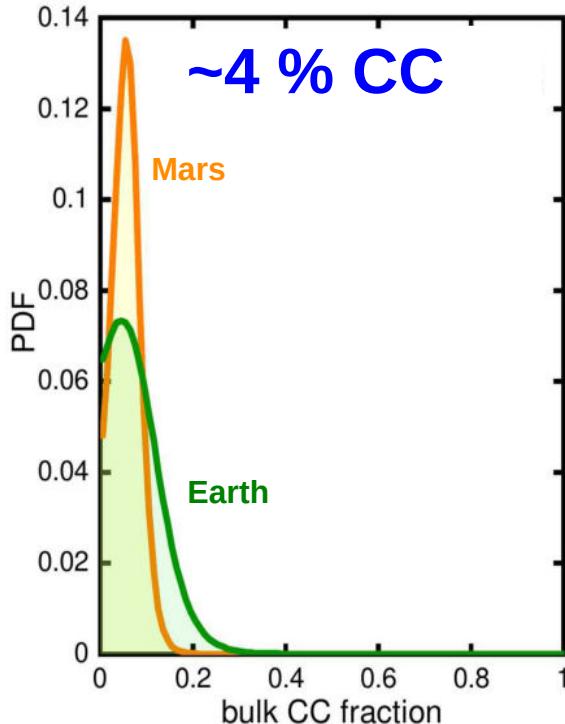
How did the disk form/disk dynamics?

Isotopic anomalies in meteorites



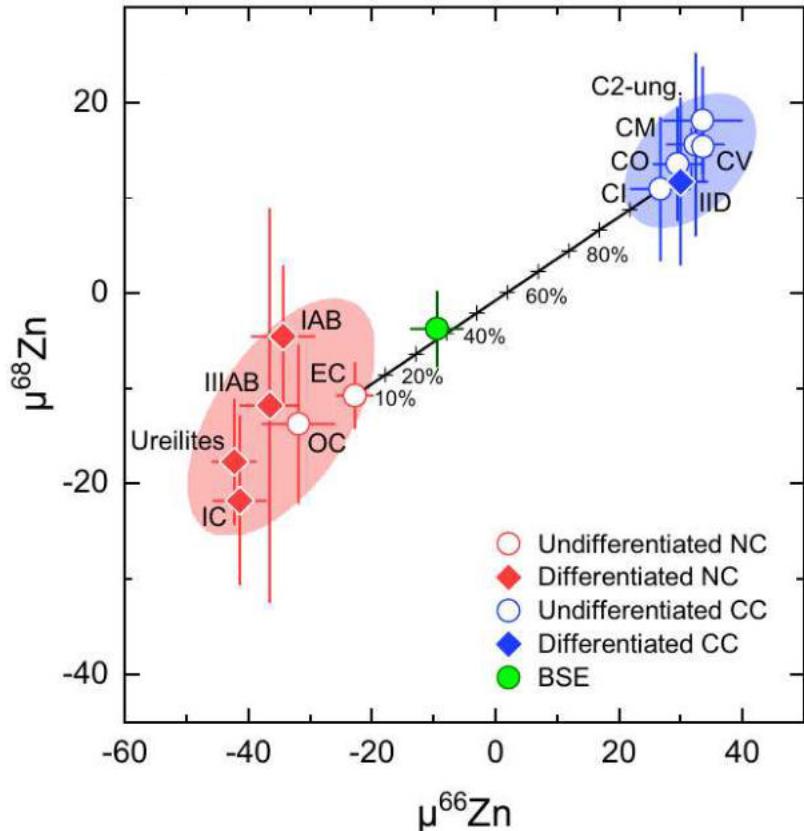
What is the Earth made of?

Isotopic anomalies in meteorites



What is the Earth made of? /How did it form?

Isotopic anomalies in meteorites



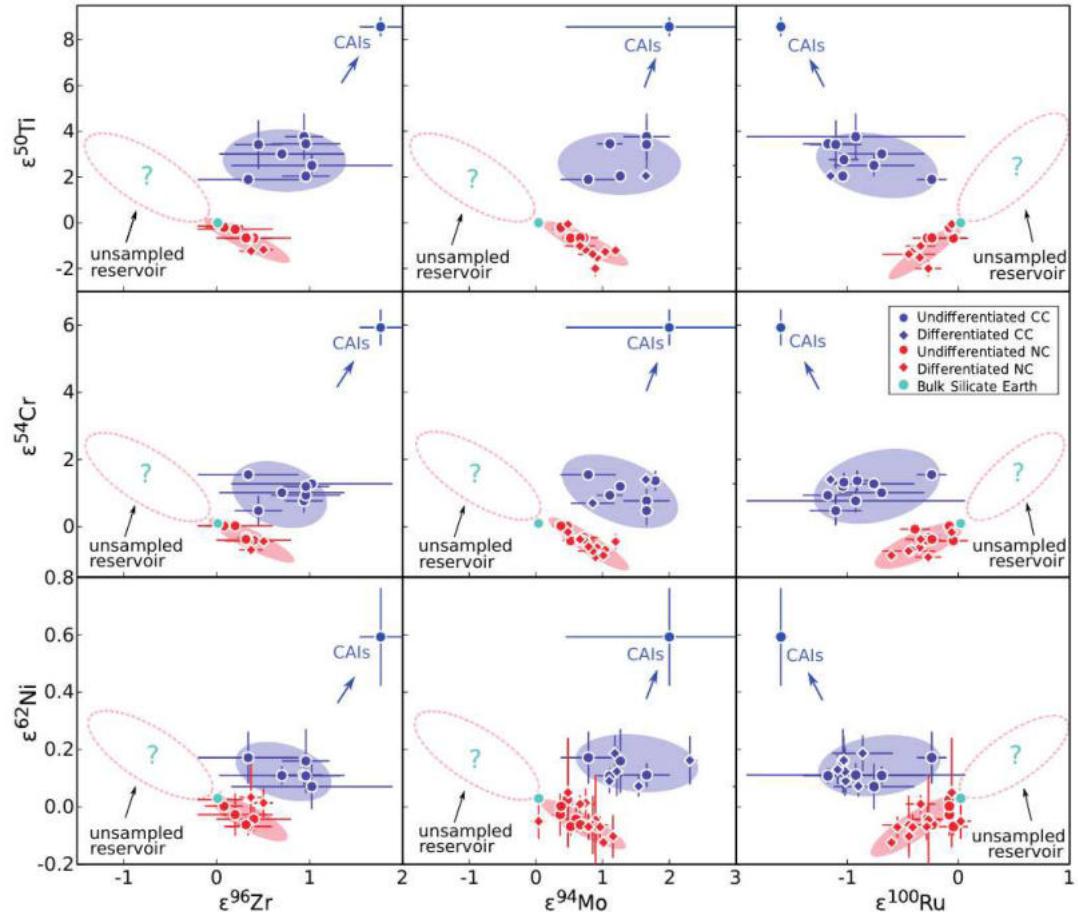
Heritage of Earth's volatiles?

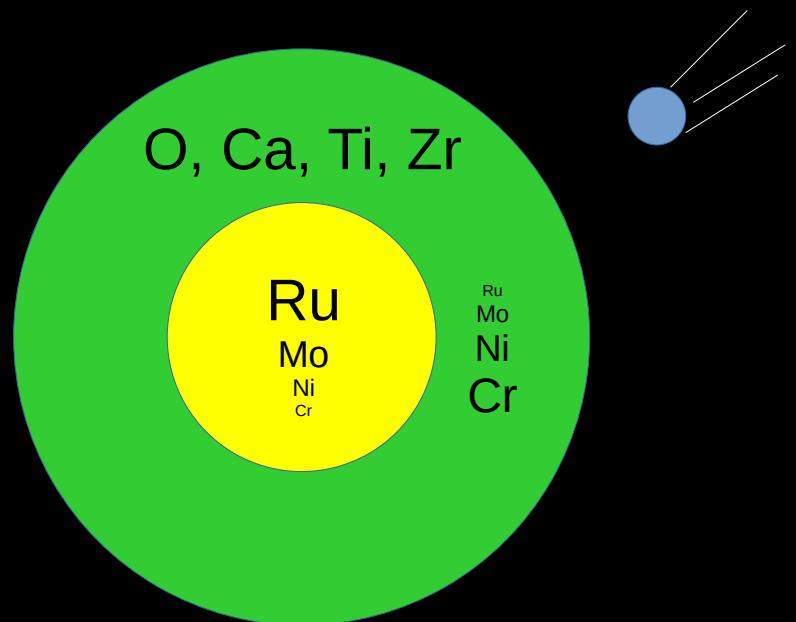
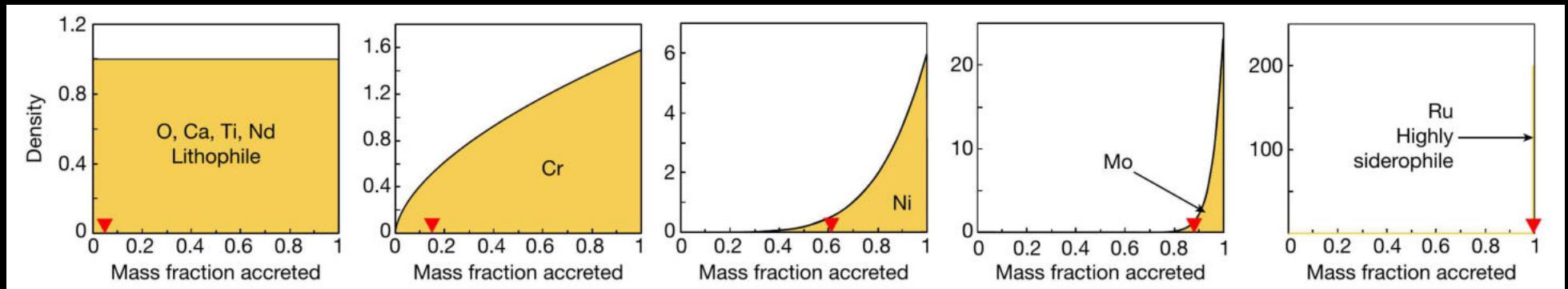
Open questions

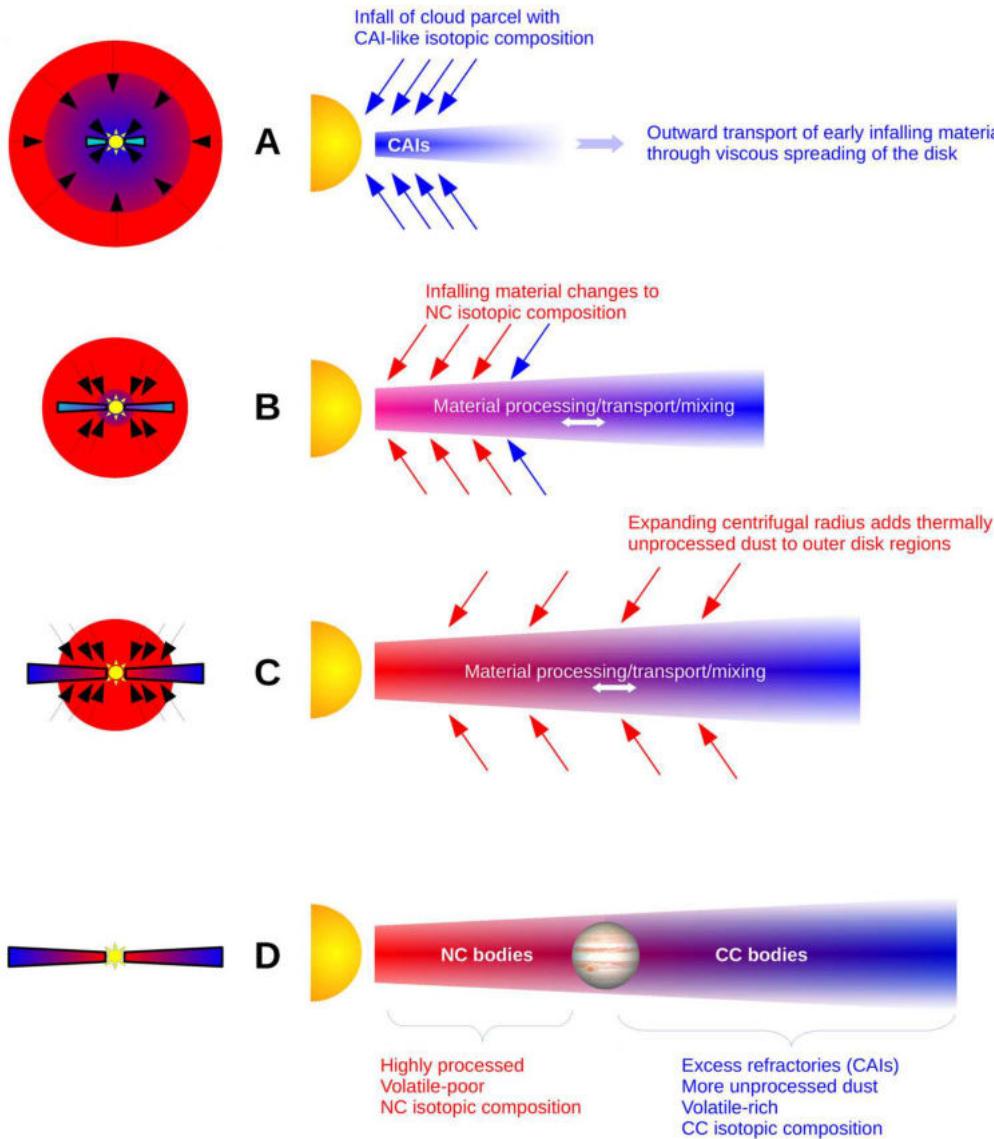
Origin NC trend?

Nature of unsampled reservoir?

Inheritance vs. disk processing?







Star and planet formation **is** diverse!

