

Yamdb/Goma - “easy” access to thermophysical properties of liquid metals and molten salts

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Who am I and what am I talking about?

- Institute of Fluid Mechanics /
Department of Magnetohydrodynamics /
Liquid Metal Batteries (LMBs)
- chemical engineer by training
(Merseburg/Halle-Wittenberg)
- **recreational programmer**
- experimenting on/simulating of LMBs
requires knowledge of material properties
- many combinations of liquid metals and
molten salts possible → substantial
amount of data needed
- extracting data from literature takes time
and effort
- treat them as a valuable resource

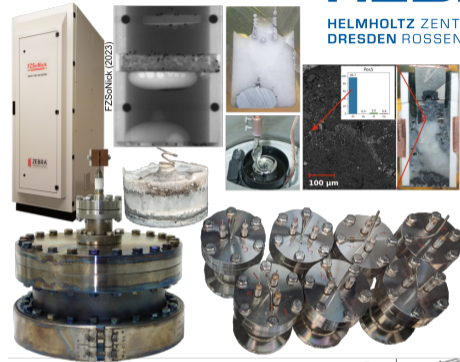


Table 2
Analytical expressions of ionic conductivity of electrolytes measured in this work

Electrolyte	Composition (mol%)	Ionic conductivity (S cm^{-1})
LiCl-LiI	36.4-65.4	$13.0462 \exp(-907.3/T \text{ (K)})$
LiI-KI	63.3-36.7	$10.0001 \exp(-1387.9/T \text{ (K)})$
LiCl-LiI-KI	8.5-59-32 + 5	$11.0055 \exp(-1329.4/T \text{ (K)})$
LiF-LiCl-LiI	11.7-29.1-59.2	$8.895 \exp(-872.6/T \text{ (K)})$
LiBr-LiCl-LiI	19-24.3-56.7	$12.6746 \exp(-925.0/T \text{ (K)})$
LiCl-KCl	58.8-42.1	$18.7876 \exp(-1800.6/T \text{ (K)})$
LiF-LiCl-LiBr	22-31-47	$17.8664 \exp(-1284.24/T \text{ (K)})$
LiF-LiBr-KBr	0.67-53.5-45.83	$20.5817 \exp(-1944.76/T \text{ (K)})$
LiCl-LiBr-KBr	12-36.6-51.4	$14.1221 \exp(-1884.66/T \text{ (K)})$

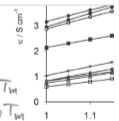


Fig. 6. Evolution of the ionic conductivity (κ) of all electrolytes i
LiCl-LiI, (○) LiCl-LiBr-LiI, (□) LiCl-LiI-KI, (△) LiCl-LiI-KI, (×) LiCl-LiI-KI.

mass%! Tab 1: 58.8-44.2

What is my Framework?

■ goals:

- small project, minimal dependencies
- easy data capture
- portability
- suitability for REPL (IPython)
- dynamic class generation (allowing for tab completion)
- easy to extend (materials, equations, properties)

■ means:

- YAML for coefficients
- BibTeX for references
- SI units only
- prefer implementation of equations over (error-prone) conversion
- unit and integration tests with Pytest
- programming extensions documented in detail

Na:

Tm: &Tm_Na 370.87

Tb: 1156.15

M: 22.98977E-03

density:

IidaGuthrie1988: &density_Na_IidaGuthrie1988

rho_m: 927

lambda: .235

equation: Steinberg1974

Ohse1985: # recommended equation

a:

- 0.89660679

- 0.51613430

default: true

equation: Shpilrain1985

dynamic_viscosity:

Hirai1992:

A: 0.0846

B: 6.412

default: true

equation: Hirai1992

comment: "p. 401/63 Tab. 1"

expansion_coefficient:

IidaGuthrie1988:

<<: *density_Na_IidaGuthrie1988

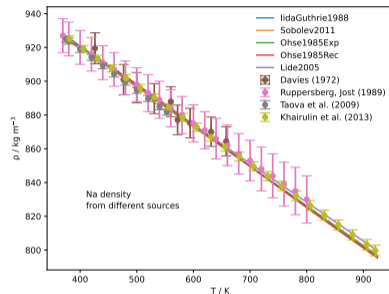
default: true

What are my Challenges and Alternatives?

- interface design, licenses
- Python (Yamdb):
 - + pragmatic choice: lingua franca of scientific computing
 - ± dependencies
 - ± NumPy docstrings are sometimes excessive
 - the Sphinx loves to pose riddles
 - rapid deprecation
 - history of package building



- Go (Goma):
 - authors: proven track record
 - simplicity as goal
 - cross-compilation, fast
 - single binaries (no dynamic libraries)
 - > go mod vendor → dependencies in source tree
 - //go:embed → ~6 MB file contains everything
 - Elisp wrapper → dynamic completion in Emacs
 - promise: Go 2 is never going to happen



Thermophysical properties of Li

property	Temp K	source	value	unit
density	800	Ohse (1985)	483.21	kg/m ³
dynamic viscosity	800	Hirai (1992)	0.00031787	Pa · s
dynamic viscosity	800	Ohse (1985)	0.00031594	Pa · s
expansion coefficient	800	Takamichi lida and Guthrie (1988)	0.0002069	1/K
heat capacity	800	Ohse (1985)	4199.8	J/(kg · K)
resistivity	800	Zinkle (1998)	3.9302e-07	Ω · m
sound velocity	800	Blairs (2007)	4326.9	m/s
surface tension	800	Keene (1993)	0.34787	N/m
thermal conductivity	800	Takachimi lida and Guthrie (2015)	56.301	W/(m · K)

References

Blairs, S. 2007. "Review of Data for Velocity of Sound in Pure Liquid Metals and Metalloids." *International Materials Reviews* 52 (6): 321–44.