

Ambient Pressure XPS and IRRAS Investigations of the CO₂ Hydrogenation Reaction

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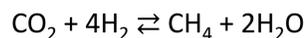
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ABSTRACT

This report is a part of experiment I have done during HESEB Research Stay Program at HIPPIE beamline in MAXIV Laboratory, Lund, Sweden. Here we investigate hydrogenation of carbon dioxide on Cu_xO/Cu(211) surface in different hydrogen to carbon dioxide ratios.

INTRODUCTION

Carbon dioxide hydrogenation to methane, or methanation of carbon dioxide, also known as the Sabatier reaction, is the hydrogenation of carbon dioxide with hydrogen to form methane. The methanation of CO₂ is the conversion of carbon dioxide and hydrogen to methane and water through the following catalytic reaction[1]



CO₂ methanation is gaining interest due to its ability to utilize greenhouse gas in the atmosphere to produce clean and green fuel, to fulfill energy demand for the growing human population[1]. Ambient pressure X-ray photoelectron spectroscopy (APXPS) is a powerful technique to study catalytic reactions, which can provide information about the chemical composition of the interface between solids and gases during the reactions.

EXPERIMENT

HIPPIE beamline at MAX IV is a soft x-ray beamline for Ambient pressure X-ray photoelectron spectroscopy (APXPS) equipped with a Bruker Vertex 70v FTIR spectrometer. Therefore, we can carry out in situ XPS and Infrared reflection absorption spectroscopy (IRRAS) measurements in the presence of a controlled gaseous atmosphere at pressures up to 30 mbar [1 mbar = 100 Pa] as well as under ultra-high-vacuum conditions. The photon energy range is 250 to 2200 eV in planar polarization and with photon fluxes >10¹² photons s⁻¹ (500 mA ring current)[2].

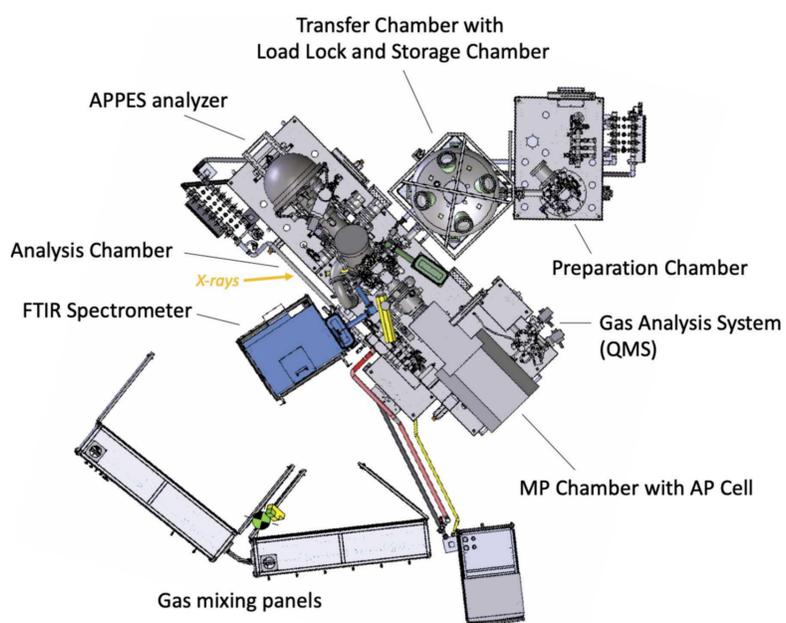


Fig. 1. APXPS endstation at HIPPIE beamline.

The Cu(211) crystal substrate cleaned by using cycles of Ar⁺ ion sputtering (1 keV) followed by annealing in vacuum at 500 °C. The crystallinity of the surface was confirmed by low energy electron diffraction (LEED) which showed a sharp LEED pattern (fig. 1 a). Copper oxide surface was prepared by annealing Cu(211) crystal at 380 °C in the oxygen pressure of 5 × 10⁻⁷ mbar. Then surface annealed at 500 °C in vacuum. Then time resolved APXPS and IRRAS measurements have been done on Cu_xO/Cu(211) surface exposed to CO₂, CO₂/H₂ (1:1), (2:1) and (1:2) at different temperatures.

RESULTS AND DISCUSSION

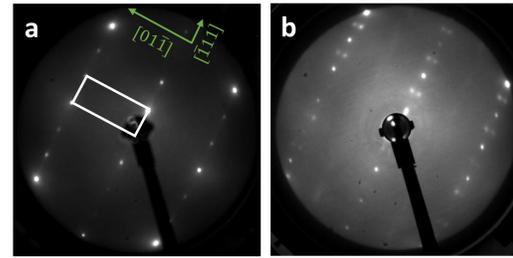


Fig. 2. LEED pattern of (a) clean surface of Cu(211), (b) oxidized Cu(211) surface (Cu_xO/Cu(211)).

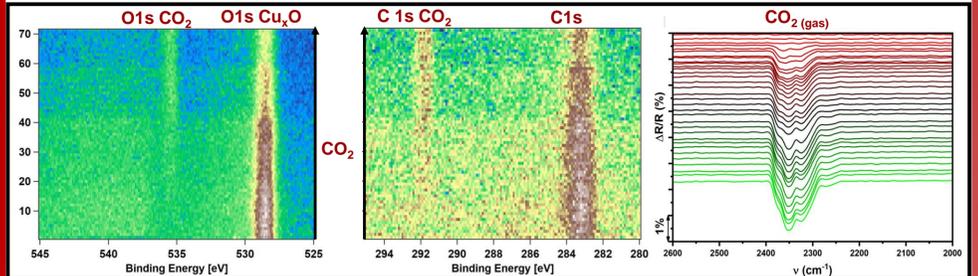


Fig. 3. CO₂ (CO₂ = 5 ml/min, P= 1 mbar)

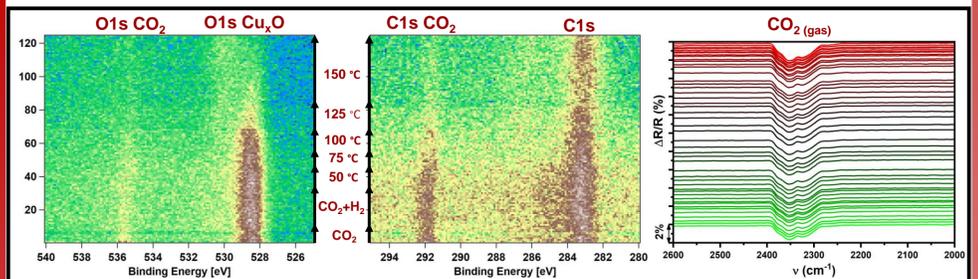


Fig. 4. CO₂/H₂ = 1:1 (CO₂ = 5 ml/min, H₂ = 5 ml/min, P= 1.6 mbar)

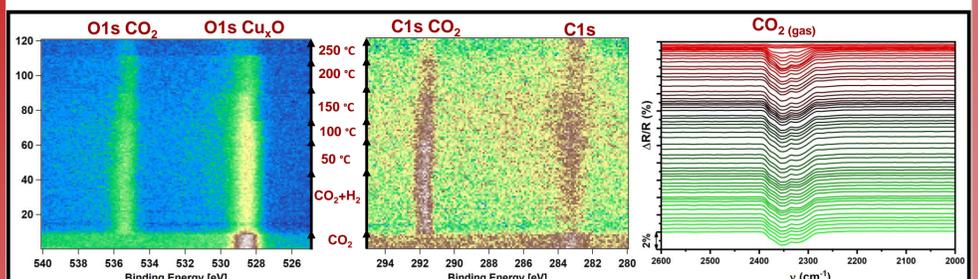


Fig. 5. CO₂/H₂ = 2:1 (CO₂ = 5 ml/min, H₂ = 2.5 ml/min, P= 0.7 mbar)

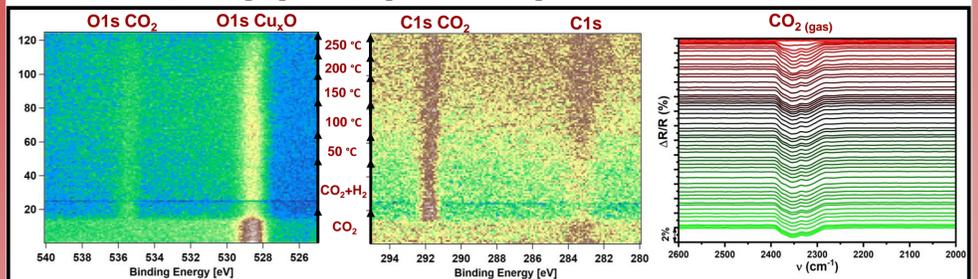


Fig. 6. CO₂/H₂ = 1:2 (CO₂ = 5 ml/min, H₂ = 10 ml/min, P= 2 mbar)

LEED pattern shows reconstruction of the Cu(211) surface after annealing in oxygen (Fig. 1 b). APXPS and IRRAS results shows there is not any CO₂ chemisorption on Cu_xO/Cu(211) surface in CO₂ atmosphere after increasing pressure to P(CO₂) = 1 mbar at room temperature (Fig. 2). Observed peak in the IRRAS is related to CO₂ gas phase. After exposing the sample to H₂ and CO₂ atmosphere in different ratios from CO₂ rich to H₂ rich region, we did not observe any carbonate/bicarbonate formation or hydrocarbon intermediates of the carbon dioxide hydrogenation.

Increase in the temperature of the sample results in the reduction of the copper oxide in the presence of hydrogen which starts at T>100 °C .

Since Cu_xO/Cu(211) surface can not activate CO₂, it is not an active surface for carbon dioxide hydrogenation.

ACKNOWLEDGMENT

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