



## Synthesis of freestanding porous alumina and copper nanowires for high power laser applications

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## Content





Short introduction

Porous alumina and metallic nanowires: an overview





Experimental

Results

### **ELI-NP Targets Laboratory**







- ✓ Provide targets in-house manufacturing and characterization
- Perform R&D activities of new target designs with improved performances
- Total surface area 268 m<sup>2</sup>
- ISO 7 and ISO 6 cleanliness
- 4 designated rooms:

nanofabrication, characterization, chemistry room, microassembly

#### ELI-NP Targets Laboratory - capabilities Deposition techniques Structuring techniques



#### **Cleaning/conditioning methods**

- Plasma  $(O_2, Ar, SF_6)$ 

- Thermal treatments

- Ion beam (Ar)





**Benchtop plasma treatment system** 

#### **Micro-machining / mechanical processing**





#### **Electrochemical set-up**



### **On-going research: micro/nanostructured targets, free-standing targets**

#### Metallic micro/nano dots





Metallic gratings



10 µm

microspheres



triangle-like metallic patterns



6

8

#### **Laser Targets**



#### Types of targets:

solid (thin/thick/ultra-thin films, multi-layer, foams, nanospheres, snow clusters, NWs, gratings, nanoparticle, micro-cone...)



## 1.1. Short introduction Nanowires (NWs)

NWs are particularly interesting as targets for laser experiments  $\rightarrow$  allowing the investigation of physical light-matter interaction phenomena (*particle acceleration*, *dynamic compression*, *radiation source...*)

NWs are structures with high aspect ratio with unique **magnetic**, **optic and electric properties**.



Particle-in-cell simulations and laser-driven ion acceleration experiments on NW arrays (brush-like geometry) showed **improved laser absorption** (on-going research).

Using these structures as targets, the improvement of the energy of the accelerated ion at intensities (laser irradiance) of  $10^{21}$  - $10^{23}$  W/cm<sup>2</sup> can be investigated.

# 1.2.Methods for NW synthesis

#### Vapor-Liquid-Solid

Chemical Vapor Deposition

Hydrothermal

#### Molecular Beam Epitaxy

#### Electrodeposition $\rightarrow$ Advantages:

- wide range of tunable parameters (material, diameter, height)
- low-cost equipment and materials

## Reactive Ion Etching

Metal Assisted Chemical Etching

Lithography

0WD

Top-L

## 1.3. Anodization process

- Al anodization is an electrochemical process (made in an electrochemical cell) that transform Al in porous alumina.
- This cell is equipped with a double glass wall and two electrodes connected to a DC power supply in constant voltage (potentiostatic) or constant current (galvanostatic) mode: anode → Al plate, cathode → inert.
- Between the anodized Al and non-anodized Al is the *barrier layer*, a continuous layer of alumina, at the bottom of the pores.
- Other characteristics of the porous alumina are anodized layer, pore diameter, interpore distance, wall thickness.
- Freestanding alumina can be obtained by detaching the anodized layer from the aluminum substrate by electrochemical means.

Power supply



## 2.1. Electrochemical set-up

Electrochemical set-up, used for aluminum anodization included a double wall glass electrochemical cell connected to a chiller, a DC power supply, 2 bench-top multimeters for current and voltage monitoring, connected to a PC interface, and a stirring and heating plate. Also, for pulsed electrodeposition, an oscilloscope, an amplifier, and a programmable power supply were also used.

## **2.2. Characterization techniques**

Atomic force microscopy (AFM)

Optical profilometer Scanning electron microscopy (SEM)



Energy-dispersive X-ray spectroscopy (EDS) & Electron Backscatter Diffraction (EBSD)





Optical microscope

## 3. Results

- The aluminum plate is prepared by mechanical polishing, degreasing, thermal treatment before the two-step anodization method.
- After the second anodization two routes were tested:
  - A) NWs on aluminum substrate synthesized by AC pulsed electrodeposition
  - B) freestanding porous alumina template by electrochemical detachment.



## 3.1. Aluminum foils pre-processing

## Mechanical polishing

00.um

after

before



reduced roughness; increased surface contaminants

# Thermal treatment avg.\* size=5.18 μm



Electropolishing



100 µm



**3.2. Synthesis of porous alumina and copper NWs on Al substrate One-step anodization and pulsed AC electrodeposition** 

## **Conditions:**

#### **One-step anodization process**

Anodization voltage set at 40V for 10 minutes, after which was raised by 0.5V /5 seconds until it reached 120V and held for 70 minutes.

#### **Barrier layer thinning**



#### **Alumina dissolution**

Partly liberation of NWs: 2min, 1M NaOH at 40C Fully dissolved after 3.5min same solution.

- **Pulsed AC electrodeposition** frequency= 200Hz
- voltage  $\pm 15V$
- Electrodeposition solution: 0.5M CuSO<sub>4</sub> and 0.57M boric acid (H<sub>3</sub>BO<sub>3</sub>) (pH 3-3.5).
- Deposition duration= 30 min
- Edges and back of Al plate was thoroughly covered with nail polish.

## 3.2. Synthesis of porous alumina and copper NWs on Al substrate One-step anodization and pulsed AC electrodeposition Results Anodization Pulsed AC electrodeposition and dissolution



# **3.3. Synthesis of freestanding porous alumina template**

- Freestanding alumina (AAO) was obtained using the detachment of porous alumina from the Al plate.
   Stair-like reverse biases applied after 2<sup>nd</sup> anodization as an environment-friendly method (Hong et al 2015).
- 0.3M oxalic acid chosen as electrolyte for anodization process, as it allows synthesis of **smoother and more uniformly** arranged pores and walls.



## **3.3. Synthesis of freestanding porous alumina template Optimized thick AAO template**

## **Conditions:**

#### **Anodization process**

0.3 M oxalic acid  $H_2C_2O_4$ electrolyte, 40V. Cathode: cylindrical electrode Ti covered with Pt.

#### Alumina detachment

The electrodes are changed:

- aluminum plate becomes the cathode (-)
- Ti/Pt counter electrode becomes the anode(+)
- Stair-like reverse biases are applied starting at 18V to 26V.



	First anodization		Second anodization		AAO
Sequence	Time, h	Temperature, C	Time, h	Temperature, C	thickness, μm
1	6	10	11.5	10	60-63
2	6	15	2	15	7-8
3	6	15	3	15	20-23



## 3.3. Synthesis of freestanding porous alumina template Optimized thick AAO template Results













### **Open pores on both sides**



Chemical etching in: -NaOH 2M -H<sub>3</sub>PO<sub>4</sub> 0.5M

# Open bottom pores



#### Top pores



500 nm

WD: 6.59

View field: 2.18 µm

SEM MAG: 127 kx

Det: In-Beam SE

Date(m/d/y): 05/26/21



## Freestanding copper nanowires On 150nm gold layer











## **Laser Experiments and PIC Simulations**



(A) Front side of the free-standing AAO mounted on a C-shaped Al-frame. (B) Experimental setup with target wheel and online imaging system used for laser focal spot optimization and target alignment. (C) Shadowgraph image of a 20 µm-thick AAO target.



#### AAO targets

\* In **collaboration** with D.Doria, P.Ghenuche, M.Cernaianu

Free-standing AAOs mounted on a frame, prepared for the experimental setup.

#### See J.F.Ong presentation Today at 15:00



#### **Nanowires Characteristics**

Diameter: 40-500 nm

Height: 500 nm- tens of micrometers

#### Interpore distance: 40-100 nm

Material: **Cu, Ni, Co, Au** (Pd, Ag, Zn, Nd, Tb, combinations)

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